

Recent advances in functionalized polymer membranes for biofouling control and mitigation in forward osmosis

Mostafa Dadashi Firouzjaei^{a,1}, S. Fatemeh Seyedpour^{b,1}, Sadegh Aghapour Aktij^{c,1},
Mattia Giagnorio^d, Nasim Bazrafshan^b, Arash Mollahosseini^e, Farhikhteh Samadi^f,
Shahin Ahmadalipour^g, Fatemeh Dadashi Firouzjaei^h, Milad Rabbani Esfahani^{a,**},
Alberto Tiraferri^{d,***}, Mark Elliottⁱ, Marco Sangermano^j, Amira Abdelrasoul^{e,****},
Jeffrey R. McCutcheon^{k,****}, Mohtada Sadrzadeh^c, Amirsalar R. Esfahani¹,
Ahmad Rahimpour^{b,d,i,*}

^a Department of Chemical and Biological Engineering, University of Alabama, Tuscaloosa, AL, 35487, USA

^b Department of Chemical Engineering, Babol Noshirvani University of Technology, Shariati Avenue, Babol Mazandaran, 4714871167, Iran

^c Department of Mechanical Engineering, 10-367 Donadeo Innovation Center for Engineering, Advanced Water Research Lab (AWRL), University of Alberta, Edmonton, AB, T6G 1H9, Canada

^d Department of Environment, Land, and Infrastructure Engineering (DIATI), Politecnico di Torino, Corso Duca Degli Abruzzi 24, 10129, Turin, Italy

^e Department of Chemical and Biological Engineering, University of Saskatchewan, Saskatoon, SK, Canada

^f Department Civil and Environmental Engineering, University of Texas at Arlington, Arlington, TX, USA

^g School of Chemistry, University of Tehran, Tehran, Iran

^h AR Membrane Lab, Shariati Av., Babol, Mazandaran, Iran

ⁱ Department Civil and Environmental Engineering, University of Alabama, Tuscaloosa, AL, 35487, USA

^j Department of Applied Science and Technology, Politecnico di Torino, C.so Duca Degli Abruzzi 24, Turin, 10129, Italy

^k Chemical & Biomolecular Engineering, University of Connecticut, 191 Auditorium Road, Unit 3222, Storrs, CT, 06269-3222, USA

¹ Department of Civil, Environmental and Construction Engineering, University of Central Florida, Orlando, USA

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ABSTRACT

Forward osmosis (FO) is an osmotically driven process widely studied for water desalination, wastewater treatment, and water reuse, as well as dilution and concentration of aqueous streams. However, its application is still hampered by the lack of ideal draw solutes, high-performance membranes, and fouling/biofouling. Biofouling is particularly challenging when FO is applied for seawater desalination and wastewater treatment. Over the last decade, many attempts have been made to exploit advances in materials science to obtain membranes with anti-biofouling properties to prevent or to reduce the detrimental effects of this phenomenon. In this review, we address the various approaches of membrane surface functionalization for biofouling control and mitigation. Recent developments in surface modification of thin-film composite and asymmetric membranes using surface coating, surface functionalization, and incorporation of tailored materials for biofouling control in FO are critically discussed. The future perspectives of anti-biofouling materials and FO membranes are reviewed to shed light on the future research directions for developing the true potential surface modification approach for the FO process.

* Corresponding authors., Department of Chemical Engineering, Babol Noshirvani University of Technology, Shariati Avenue, Babol Mazandaran, 4714871167, Iran.

** Corresponding authors.

*** Corresponding authors.

**** Corresponding author.

***** Corresponding author.

E-mail addresses: mesfahani@eng.ua.edu (M.R. Esfahani), alberto.tiraferri@polito.it (A. Tiraferri), amira.abdelrasoul@usask.ca (A. Abdelrasoul), jeff@engr.uconn.edu (J.R. McCutcheon), ahmad.rahimpour@polito.it, ahmadrahimpour@nit.ac.ir (A. Rahimpour).

¹ The authors had equal contribution as first author.

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1. Introduction: FO opportunities, challenges, and process description

Forward osmosis (FO), introduced by Kessler and Moody in 1976, is one of the promising energy-efficient and sustainable membrane technologies for aqueous separation processes [1]. Although FO was introduced over four decades ago, research interest in FO has increased recently, with over 1700 total publications since 2006 and the numbers increasing year-on-year [2]. FO membranes are used in diverse applications, such as desalination and wastewater reuse [3], osmotic membrane bioreactors [4], power generation [5], and food processing [6]. Despite the developments achieved by membrane-based technologies, fouling is still the most challenging problem in their industrial implementation. Fouling is defined as a process in which colloidal particles, organic and inorganic matter, and biological materials deposit on the surface of the membrane or in the membrane pores during filtration [352-353]. Membrane fouling results in higher feed pressures, higher operational costs, higher demand for chemical cleaning, and shorter overall membrane life. Biofouling is a specific type of fouling caused by the deposition and growth of living organisms and microorganisms on the surface [7,8].

Filtration of biologically contaminated wastewater by membranes results in microbial cell multiplication, colonization, and attachment to the surface of the membrane. The exopolymeric substances that microorganisms typically produce form a biofilm that impairs membrane performance [9]. The intensity of biofouling differs based on wastewater characteristics, including: (1) nutrient content, (2) available biological species and concentration, (3) temperature, (4) light exposure, (5) turbidity and (6) currents (tides and waves) [10]. Depending on the interactions between microorganism and membrane surface, biofouling can be promoted or prevented. Table 1 classifies the parameters affecting biofouling, including membrane, environment, and microbial characteristics [10,11]. Several studies have been published discussing advances in FO membrane fabrication and properties [12-14]. However, to the best of our knowledge, there is no comprehensive study to review the surface modification strategies of TFC FO membranes for fouling and biofouling control and mitigation.

FO involves two streams flowing tangentially on the two sides of a semipermeable membrane. The driving force for mass transport is provided by the difference in osmotic pressure between the two sides of the membrane (feed solution (FS) and draw solution (DS)). This chemical potential difference results in a net transport of mass from the stream with lower osmotic pressure (feed solution) to the other stream (draw solution) with higher osmotic pressure. In other words, the concentrated stream or draw solution exerts a higher osmotic pressure and draws water, solvents or a targeted substance from the feed solution [16]. Since

the FO process is osmotically-driven, there is no need for an external energy sources, other than a small pressure of around 1–2 bar to overcome the frictional resistance in the flow channels. In particular, the absence of hydraulic pressure results in lower biofouling in FO compared to reverse osmosis (RO). That being said, when it comes to the production of freshwater from a contaminated feed stream, FO cannot be applied as a single unit process since it only dilutes the draw solution and concentrates the feed solution. A second separation stage is needed to regenerate the draw solution [17]. This is the reason why FO is mostly used only as a pretreatment for purification of highly contaminated waters [18,19]. The physicochemical characteristics of the draw solution are also determinants of the FO performance. Several reviews have covered different aspects of draw solution development [20,21]. These studies have assessed the effect of novel polymeric, organic, inorganic, and magnetic draw solution on the FO membrane performance [22,23].

1.1. FO membranes: requirements and design criteria

FO uses polymeric membranes similar to those used in many other membrane-based separation processes. Two main types of membranes have been applied in FO: (1) thin-film composites (TFC) membranes formed by interfacial polymerization on a porous substrate, and (2) phase inversion cellulosic membranes. Both FO membrane types can come in one of three structures: flat sheet, hollow fiber, or as electrospun layers. Additionally, numerous pre- and post-modification techniques can be used to improve the membrane performance for specific applications, including mixed-matrix membrane, surface coating, surface functionalization, and incorporation of nanoparticles [24,25].

1.1.1. Thin film composite membranes (TFC)

TFC membranes used in FO are commonly formed by interfacial polymerization (IP) via the polycondensation reaction between two monomers on the surface of a porous substrate (usually an ultrafiltration membrane). The IP process starts with the immersion of the substrate into the aqueous solution containing an amine-rich monomer. Then, the amine saturated membrane is immersed into an organic solvent containing acylchloride monomers. The reaction between the monomers at the interface of the two immiscible solvents results in the formation of a thin and dense layer on top of the substrate [26,27]. It is also worth mentioning that selecting more hydrophilic sublayers can enhance water flux and reduce internal concentration polarization (ICP) [26,28]. Typical monomers used for IP are listed in Table 2, while typical supports used to fabricate TFC FO membranes are summarized in Table 3.

1.1.2. Cellulosic FO membranes as integrated asymmetric membranes

The invention of the Loeb-Sourirajan membrane fabrication procedure in the 1960s has had a substantial and lasting impact on the application of membranes [29,30]. Since then, cellulosic membrane attracted attention due to the simplicity of their synthesis [31,32]. Cellulose FO membranes are commonly produced through nonsolvent-induced phase separation methods [33]. However, pH sensitivity, temperature resistance, and biofouling mitigation are three critical weaknesses of the cellulosic FO membrane [21].

2. Biofouling in forward osmosis membranes

2.1. Definition and mechanisms

Biofouling is defined as microbial deposition onto a surface with subsequent growth to form a biofilm. Biofouling can occur in any membrane filtration unit where a membrane is submerged in a non-sterile aqueous system, even in oligotrophic environments. The membrane represents an ideal habitat for diverse microorganisms (such as *Pseudomonas*, *Aeromonas*, *Corynebacterium*, and *Fluviicola*) [89], which can attach, grow, and replicate on its surface, eventually forming mixed-species communities [90]. This phenomenon has been studied for

Table 1
Parameters affecting biofouling in membrane processes [10,15].

Affecting Parameters Category			
Microorganism	Membrane Properties	Feed Water	Draw solute
Species	Chemical composition	Temperature	Solute type
Composition of mixed population	Surface charges	pH	Concentration
Population density	Surface tension	Dissolved organic matter	Size/Size distribution
Growth phase	Hydrophobicity	Dissolved inorganics	pH
Nutrient status	Conditioning film	Suspended matter	Viscosity
Hydrophobicity	Roughness	Viscosity	Temperature
Surface charges	Porosity	Shear forces	Velocity
Physiological responses		Boundary layer	Shear Forces
		Flux/Velocity	

the past decades, during which researchers have demonstrated the formation of complex biofilms on the membrane surface due to bacterial colonization [91]. The biofilm acts as an additional layer that increases the overall hydraulic resistance to mass transport, and it compromises the membrane integrity, in turn affecting system productivity [92]. Among the different fouling phenomena, biofouling is considered the most complicated and challenging phenomenon. This complexity is due to the microorganisms' capability of fast reproduction and high resilience to the surrounding environment [93]. Organic and inorganic fouling can generally be managed by reducing the concentration of the undesirable species in solution through feed pretreatments [94], but bacterial growth is usually inevitable whenever nutrients are present in the water [95,96].

Fig. 1 schematically depicts the biofilm formation and growth. Biofouling initiates with bacteria cell transportation through the solid-liquid interface and continues with the attachment of bacteria cells to the membrane surface, often as a result of electrostatic interactions between membrane and bacteria [97]. Along with electrostatic interactions, biofouling is influenced by other important factors, i.e., the membrane physical properties (roughness, hydrophobicity, electro-kinetic charge, and pore size), the physio-chemical characteristics of the feed solution (temperature, pH, ionic strength and composition, nutrients, pollutants, and osmotic pressure), and by microbial properties (size, cell surface hydrophobicity, and charge). It is also influenced by the presence of pre-absorbed macromolecules, often generated by the bacterial cells themselves [98]. One of the major categories of biofouling precursors is transparent exopolymer particles (TEPs). TEPs are organic compounds, mostly comprising polysaccharides, dissolved in water and with typical gel-like structure and sticky properties [99]. TEPs can contribute actively to bacterial attachment providing a nutritious substratum for bacteria colonization and enhancing the adhesion of microorganisms to the membrane

surface.

The formation of floating proto-biofilms in water, composed by TEPs and microorganisms, can strongly enhance biofouling in membrane systems [100]. Control of organic fouling is thus essential for the prevention of biofouling in membrane-based water treatment processes. It should also be noted that synergetic biological, organic, and colloidal fouling usually leads to a more severe flux decline than the algebraic sum of flux declines by the individual foulants. Flagellar motility can also help microorganisms overcome possible membrane repulsive forces [101].

Extracellular polymeric substances (EPSs) are organic macromolecules consisting mostly of proteins and polysaccharides that can surround the bacterial cell, embed the microorganism, and also detach from the cell [102]. EPSs prevent the direct contact of the cell with the adverse surrounding environment and are responsible for biofilm development, thanks to their aptitude of adhesion to the membrane surface [103]. EPSs are characterized by numerous functional groups and large surface area, which facilitate the attachment [103]. Not only do EPSs cause bacterial proliferation and biofilm formation, but they can also affect the membrane performance directly through pore blocking or cake-enhanced concentration polarization. If EPSs generate a stagnant liquid layer on the membrane surface, this unstirred layer lowers both the turbulence flow at the solid-liquid interface and thwarts the back-diffusion of solutes, which consequently decreases the driving force to water permeation and increases solute passage through the membrane [103].

Biofilms present heterogeneous compositions, which may vary consistently from the suspended biomass. Based on different system operating conditions, a different biomass distribution between external and internal biofilm has been reported [104]. Also, after biofilm formation, other foulants can react with the biofilm and increase the intensity of biofouling [105]. Usually, cell development at the surface

Table 2

Recent monomers for the fabrication of thin film composite membranes and recent sublayers for used for FO membranes [21]. (reproduced with permission from Elsevier).

Aqueous Phase Monomer	Organic Monomer	Ref	Aqueous Phase Monomer	Organic Monomer	Ref
m-phenylenediamine (MPD), 2,5-diamino-N-(4-amino-phenyl) benzamide (DABA)	trimesoyl chloride (TMC)	[34]	MMPD, MPD	Cyclohexane-1,3,5-tricarboxyl chloride (HTC), TMC	[35]
Dopamine	TMC	[36]	Bisphenol A (BPA)	TMC	[37]
MPD	2,4,4',6-biphenyl tetraacyl chloride (BTAC), 2,3',4,5',6-biphenyl pentaacyl chloride (BPAC), 2,2',4,4',6,6' biphenyl hexaacyl chloride (BHAC), 3,3',5,5'-biphenyl tetraacyl chloride (BTEC), TMC	[38]	MPD, o-aminobenzoic acid-triethylamine salt (o-ABA-TEA), m-aminobenzoic acid-triethylamine salt, (m-ABA-TEA), 2-(2-hydroxyethyl) pyridine, 4-(2-hydroxyethyl) morpholine	TMC	[39]
Piperazine (PIP)	BETEC	[40]	MPD, 2,2'-benzidinedisulfonic acid (BDSA)	TMC	[41]
PIP	BHAC	[40]	PIP, N-aminoethyl piperazine propane sulfate (AEPPS)	TMC	[42]
Disulfonated bis[4-3 aminophenoxy]phenyl sulfone (S-BAPS)	TMC	[43]	Triethanolamine (TEOA), N-methyl-diethanolamine (MDEOA)	TMC	[44]
Ethylenediamine (EDA), Hyperbranched polyethyleneimine (PEI), Diethylenetriamine (DETA)	TMC	[45]	Polyamide polyvinylamine (PVAm)	IPC	[46]
MPD	TMC, 5-isocyanato-isophthaloyl chloride (ICIC), Chloroformyloxyisophthaloyl chloride (CFIC)	[47]	4-aminobenzoylpiperazine (4-ABP)	TMC	[48]
MPD	CFIC, Terephthaloyl chloride (TPC), Isophthaloyl chloride (IPC)	[49]	MPD	2,4,6-pyridinetri-carboxylic acid chloride (PTC), TMC	[50]
m-phenylenediamine-4-methyl (MMPD), n,n"-dimethyl-m-phenylenediamine (DMMPD), MPD	CFIC, TMC	[51]	MPD, 1,2-diamino,2-hydroxypropane (DAHPP)	TMC	[52]
MPD	1,2,4,5-benzene tetracarboxyl chloride (BTC), TMC	[53]	Bis-2,6-N,N-(2-hydroxyethyl) diaminotoluene	TMC	[54]
PIP, 2,2'-oxybis-ethylamine (2,2'-OEL)	TMC	[55]	melamine	TMC	[56]

Table 3
Recent sublayers used for FO membranes [52]. (reproduced with permission from Elsevier).

Support Material	Geometry	Performance				Operating Condition		Ref	Support Material	Geometry	Performance				Operating Condition		Ref
		FO Mode		Feed Solution		Feed Solution	Draw Solution				FO Mode		Feed Solution		Feed Solution	Draw Solution	
		Water Flux (LMH)	Salt Flux (gMH) or % R	Water Flux (LMH)	Salt Flux (gMH)						Water Flux (LMH)	Salt Flux (gMH)	Water Flux (LMH)	Salt Flux (gMH)			
PSf	Flat sheet	9.5/12.0	2.4/4.9	18.1/20.5	6.3/5.9	10 mM NaCl	0.5 M NaCl	[57]	PSf-PET fabric	Flat sheet (co-casting)	31.1	8.5	60.3	17.6	DI water	1.0 M NaCl	[58]
PSf	Flat sheet	20.1	2	33.1	2.6	DI water	0.5 M NaCl	[59]	PSf/silica (3 wt %)-PET fabric	Flat sheet (co-casting)	31	7.4	60.5	16	DI water	1.0 M NaCl	[60]
PSf/SPEK	Flat sheet	35	7	50	9	DI water	2.0 M NaCl	[61]	PSf/TiO2(0.5 wt)	Flat sheet	17.1	2.9	31.2	6.7	10 mM NaCl	0.5 M NaCl	[62]
PES/PESU-co-sPPSU	Flat sheet	21	2.2	33	2.8	DI water	2.0 M NaCl	[63]	PSf/rGO-modified graphitic carbon nitride (0.5 wt%)	Flat sheet	41.4	9.6	–	–	DI water	2.0 M NaCl	[64]
sPPSU	Flat sheet	48	7.6	54	8.8	DI water	2.0 M NaCl	[65]	PSf/zeolite (0.5 wt %)	Flat sheet	40	28	86	57	DI water	2 M NaCl	[66]
Hydrolyzed CTA	Flat sheet	–	–	2.4–6.7	8.0–47.8	DI water	1.5 M NaCl	[67]	PES/MWCNTs (2.0 wt%)	Flat sheet	–	–	12	94.7	10 mM NaCl	2 M glucose	[68]
PDA-modified PSf	Flat sheet	8.2	1.4	24	1.8	DI water	2.0 M NaCl	[69]	PVDF nanofiber	Flat sheet	11.6/28	3.5/12.9	47.6	6.4/21.6	DI water	1.0 NaCl	[70]
PSf/PES	Flat sheet	27.6	37.5	–	–	DI water	2.0 M NaCl	[71]	Polyketone	Flat sheet	12.6–29.3	2.0–3.8	22.6–41.5	2.8–5.0	DI water	0.6 M NaCl	[72]
PES/SPSfa	Flat sheet	26	8.3	47.5	12.4	DI water	2.0 M NaCl	[73]	PAN	Flat sheet	9.25	5.8/6.4	11.6/13.9	5.8/6.4	DI water	0.5 M NaCl	[74]
Carboxylated PSf	Flat sheet	18	2.2	27	5.5	DI water	1.0 M MgCl2	[75]	Nylon 6,6 microfiltration membrane	Flat sheet	6	0.7	21.5	0.8	DI water	1.5 M NaCl	[76]
PSf-PET fabric	Flat sheet	0.5–25	95.8–99.3	–	–	DI water	1.0 M NaCl	[77]	CAP	Flat sheet	80.1	10	128.8	19.4	DI water	2.0 M NaCl	[78]
PSf-PET fabric	Flat sheet	18.2	97.4%	–	–	DI water	1.5 M NaCl	[79]	CAPa	Flat sheet	–	–	31.8/35.0	1.6/1.9	DI water	2.0 M NaCl	[80]
PEI/PAA-coated hydrolyzed PAN-PET fabric	Flat sheet	24.6	2.4	32.9	3.8	DI water	0.5 M NaCl	[77]	PES	Hollow fiber	5/14	2.1/1.8	12.9/32.2	5.0/3.5	DI water	0.5 M NaCl	[81]
Hydroxyl functionalized PTA-POD	Flat sheet	37.5	5.5	78.4	12.3	DI water	1.0 M NaCl	[82]	PES	Hollow fiber	32.1–34.5	6.2–9.9	57.1–65.1	6.9–12.3	DI water	2.0 M NaCl	[83]
PES	Hollow fibre	16.7–18.7	1.2–2.0	43.6–49.4	2.8–4.0	DI water	0.5 M NaCl	[84]	PES	Hollow fiber (double-skinned)e	14.2–17.3	3.5–4.2	32.7–38.4	3.6–4.0	DI water	0.5 M NaCl	[85]
PES	Hollow fiber	–	–	42.6	4	DI water	0.5 M NaCl	[86]	PAI	Hollow fiber (double-skinned)e	16.9	16	41.3	5.2	DI water	2.0 M NaCl	[87]
PVDF	electrospun nanofiber	30.4	6.4	–	–	DI water	1.0 M NaCl	[70]	PVDF-0.5SiO2	electrospun nanofiber	83	63.91	–	–	DI water	2.0 M NaCl	[88]

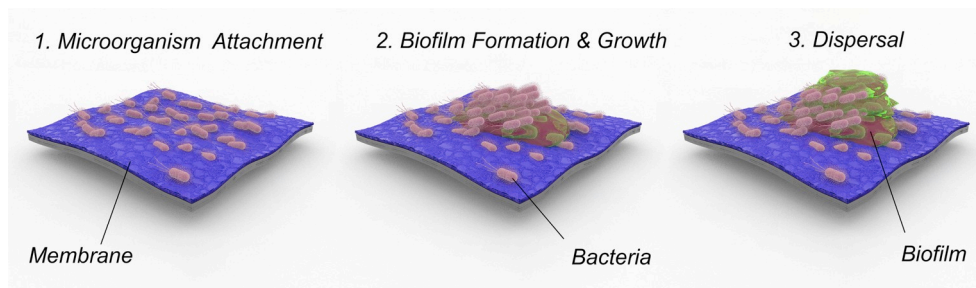


Fig. 1. Biofilm formation steps, (1) attachment to the surface, often preceded by a surface conditioning step and/or by deposition of organic material (2) growth and creation of microcolonies, and (3) expansion of the biofilm to other areas, often followed by detachment of part of the biofilm and dispersal.

contributes more to biofilm accumulation than does the adhesion of other cells from the feed suspension, leading to an exponential development of the biofilm right after bacteria attachment to the surface [106]. Therefore, pretreatment to reduce the concentration of nutrients in the feed solution is a key factor in slowing down the kinetics of biofouling. Specifically, not only more robust feed pretreatments are necessary to remove nutrient and organic macromolecules from solution, but also periodic membrane cleaning with biocides or oxidizing agents is compulsory to extend the membrane lifetime [107,108].

2.2. Parameters influencing biofouling in forward osmosis

It is generally accepted that fouling in osmotically driven membrane processes (ODMPs) is less problematic than in pressure-driven membrane processes (PDMPs). The fouling layer formed on the FO membrane surface is usually less compact than that formed in RO or NF, due to the lack of hydraulic pressure [109]. Therefore, fouling in FO is usually more reversible [110]. However, biofouling still represents one of the major challenges for the advancement of FO and it will be one of the most important topics for future research activity, especially considering that FO is increasingly being employed in processes treating complex feed sources, such as in bioreactor systems or microalgae separation for biodiesel production, where PDMPs are prone to irreversible fouling [111].

Unlike PDMPs, nutrients for biofilm development in FO can derive not only from the feed water but also from the draw solution, as a result of the reverse solute diffusion through the membrane into the feed solution [95]. For example, this can occur with draw solutions containing fertilizer solutes or ammonia-carbon dioxide salts. This mechanism may be particularly important when the draw solutes consist of short-chain organic molecules, such as salts of volatile fatty acids [112]. Even simple inorganic draw solutes can enhance biofilm formation (e.g., EPS secretion from the utilization of magnesium-based draw solutes) [113]. Also, cellulose triacetate (CTA) membranes used in FO membrane bioreactors can be biologically degraded by the microorganisms during the development of their colonies and provide themselves a nutrient source [114].

The configuration of the spacer used in the FO membrane module is another important factor affecting biofouling. Valladares Linares et al. have studied the impact of the spacer thickness on biofilm formation, showing the benefits derived from the utilization of a thicker spacer during FO filtration. These results are rationalized with the higher turbulence flow generated at the solid-liquid interface, which decreased both the attachment and the extent of solute concentration polarization, hence the availability of nutrients [115]. This is also in accordance with organic/inorganic fouling where higher linear flow velocity at the membrane interface generally improves the system performance [116].

Experiments performed with the same membrane materials and under identical processing conditions in FO and RO have shown different biofilm characteristics. The biofilm formed in RO was thinner and more compact [117]. Also, the bacteria concentration in the biofilm

was found to be 1.5 times higher in RO compared to FO. Kwan et al. have reported a 10% flux decline in FO, compared to a decrease of almost 30% observed during the analogous RO process as a result of a more compacted biofilm [118]. Therefore, the increase of the hydraulic resistance and the enhancement of the concentration polarization are more significant in the case of the RO system. Characterizations have shown a sparser, thicker, and more distended finger-like/mushroom structure of the biofilm developed on FO membranes [117].

One of the most promising applications for FO is its integration into membrane bioreactor systems (MBRs) to form FO-MBRs. Nevertheless, biofouling is also recognized as a challenge adversely affecting the performance of FO-MBR systems. Specifically, there are reports of large water flux decline due to biofouling in FO-MBR even with the implementation of osmotic backwashing [119], suggesting the necessity of chemical cleaning in the presence of microorganisms and bacteria in the feed water. Several factors influence the extent of biofouling in FO-MBR. The level of aeration and the presence of soluble microbial products are arguably the most important aspects [120]. Studies have proven that increasing the overall oxygen concentration in the feed water results in less biofouling. In bioreactor systems, bacteria are also commonly producing organic compounds denoted as soluble microbial products (SMPs), which can impact the quality of the feed and permeate water [121]. Previous studies have shown higher fouling potential of SMPs compared to common organic macromolecules, such as colloidal NOM or BSA [122]. SMPs may thus act as a potential substratum, enhancing bacteria attachment and their proliferation on the membrane surface [123].

2.3. Biofouling monitoring and characterization techniques

The purpose of biofouling monitoring and characterization in membrane processes is to identify the formation and growth of biofilms, as well as to monitor cell accretion and suggest the proper strategies for biofouling mitigation [10]. Moreover, characterization can help engineers to better predict membrane performance in different processes. The methods used to monitor and characterize membrane biofouling include a range of techniques, from simple primary detection via eyesight to advanced techniques, such as microscopy techniques (light microscopy, epifluorescence microscopy (EFM), electron microscopy, confocal laser scanning microscopy (CLSM), and atomic force microscopy (AFM)) and spectroscopy techniques (Raman spectroscopy, infrared spectroscopy, bioluminescence, fluorometry techniques, nuclear magnetic resonance (NMR) spectroscopy, photoacoustic spectroscopy (PAS), Fourier transform infrared spectroscopy (FTIR), and three-dimensional excitation-emission matrix fluorescence spectroscopy (3DEEM)). However, the application of spectroscopic methods is still challenging as there are few examples of the use of spectroscopy for biofilm characterization in the literature [124]. Therefore, these methods must be combined for a better understanding of a biofilm layer formation and growth [125]. Four common techniques used to monitor the membrane biofouling are shown in Fig. 2. Among these methods,

confocal laser scanning microscopy (CLSM) is one of the most elegant methods to evaluate a biofilm because of its ability to provide high resolution optical images, profiles of porosity-depth, to provide a measurement of the percentage of dead bacteria, and because it also allows the recognition of different constituents of biofilms, such as polysaccharides, proteins, and dead or living cells. In the CLSM technique, the membrane is stained by fluorescent dyes. Then, a laser penetrates into the biofilm sample to excite these molecules. The fluorescence emitted from the excited molecules transforms into photons, which are then converted to electron signals, in turn transformed into three-dimensional images via the help of an image processor [126–128].

The bacterial communities on fouled membranes have also been evaluated by heterotrophic total plate count, colony-forming units (CFU) methods, and adenosine triphosphate (ATP) bioluminescence assay tests [131]. ATP is a biochemical method that indicates the biological activity of live cells and biomass. These methods are not able to provide the amount of dead bacteria cells. Also, in many biofouling investigations in MBR systems, more often molecular methods, such as clone library, polymerase chain reaction-denaturing gradient gel electrophoresis (PCR-DGGE), and fluorescence in situ hybridization (FISH), have been used to detect and describe bacterial communities [132].

2.4. Biofouling modeling in forward osmosis

Numerical modeling integrated with experimental data is a helpful tool for predicting the biofouling behavior of the FO membranes. The integration of these methods also show promise for identifying feasible ways to biofouling mitigation with advanced materials. Bucs et al. evaluated the influence of biofouling on the FO membrane performance in a cross-flow system. They applied two-dimensional numerical computer simulations combined with experimental data to investigate the

effects of (i) biofouling on external concentration polarization (ECP) under various FO conditions, and (ii) biofilm attributes (porosity, hydraulic permeability, and thickness), biofilm membrane surface coverage, and biofilm size (draw channel, feed channel or both) on the FO membrane performance. The results demonstrated that: (i) the biofouling surface remarkably affected the ECP and could be the controlling parameter for water permeation; (ii) the biofilm formed in the draw side decreased the water flux due to higher internal concentration polarization (ICP); and (iii) the biofilm hydraulic resistance and membrane surface coverage also considerably affected the water flux. The authors also reported all biofilm properties, such as hydraulic permeability, average thickness, porosity, and spatial heterogeneity are important factors influencing the system productivity. Due to the dynamic behavior of biofilm, measurements should determine how the biofilm properties are evolving overtime. The geometry of the FO model, the effect of biofilm and related factors on the water flux are illustrated in Fig. 3. Overall, the numerical models can help researchers to better understand the effect of biofilm/biofouling in FO membrane processes and to apply feasible strategies in biofouling control and mitigation. Additionally, future experimental studies should focus on the use of new modification approaches for the development of state-of-the-art anti-biofouling membranes and novel cleaning strategies. Also, it is strongly suggested to integrate numerical modeling methods with experimental data to reach a comprehensive evaluation of product quality and energy parameters [133,134].

3. Membrane biofouling control strategies in forward osmosis

3.1. Pretreatment technologies and cleaning strategies

Delaying or reducing the intensity of biofouling is typically easier

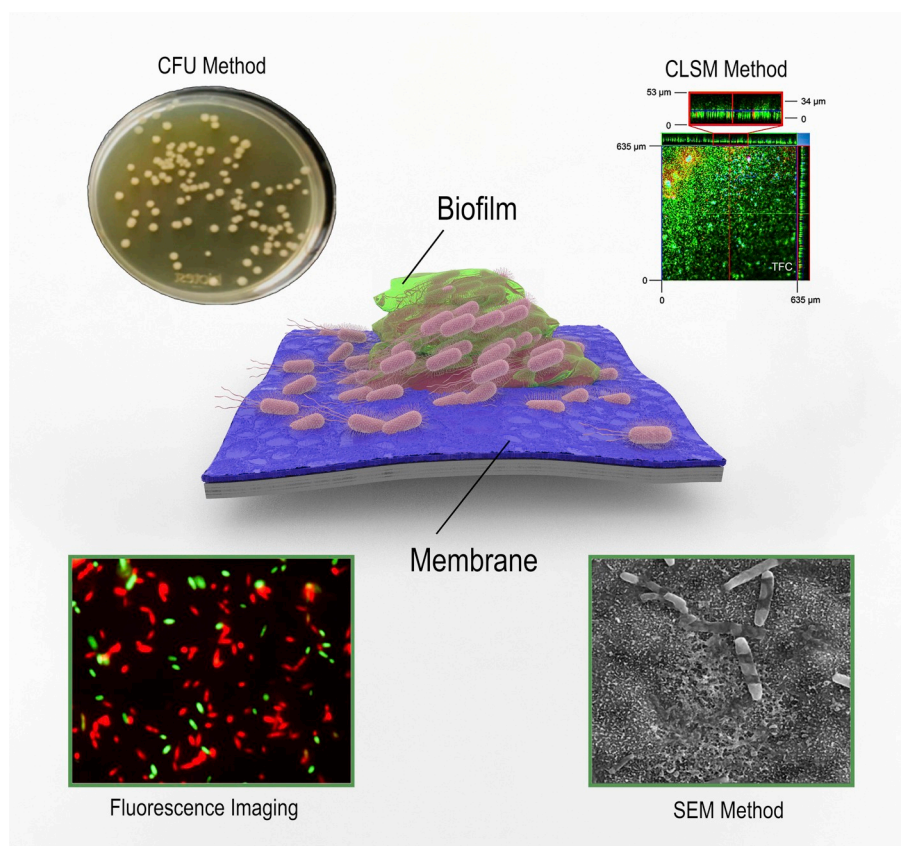


Fig. 2. Biofouling characterization methods; SEM image of *E. coli* on top side of TFC FO membrane, CFU test of TFC FO membrane, fluorescence imaging [129], reproduced with permission from Elsevier, and CLSM method [130], reproduced with permission from ACS Publications.

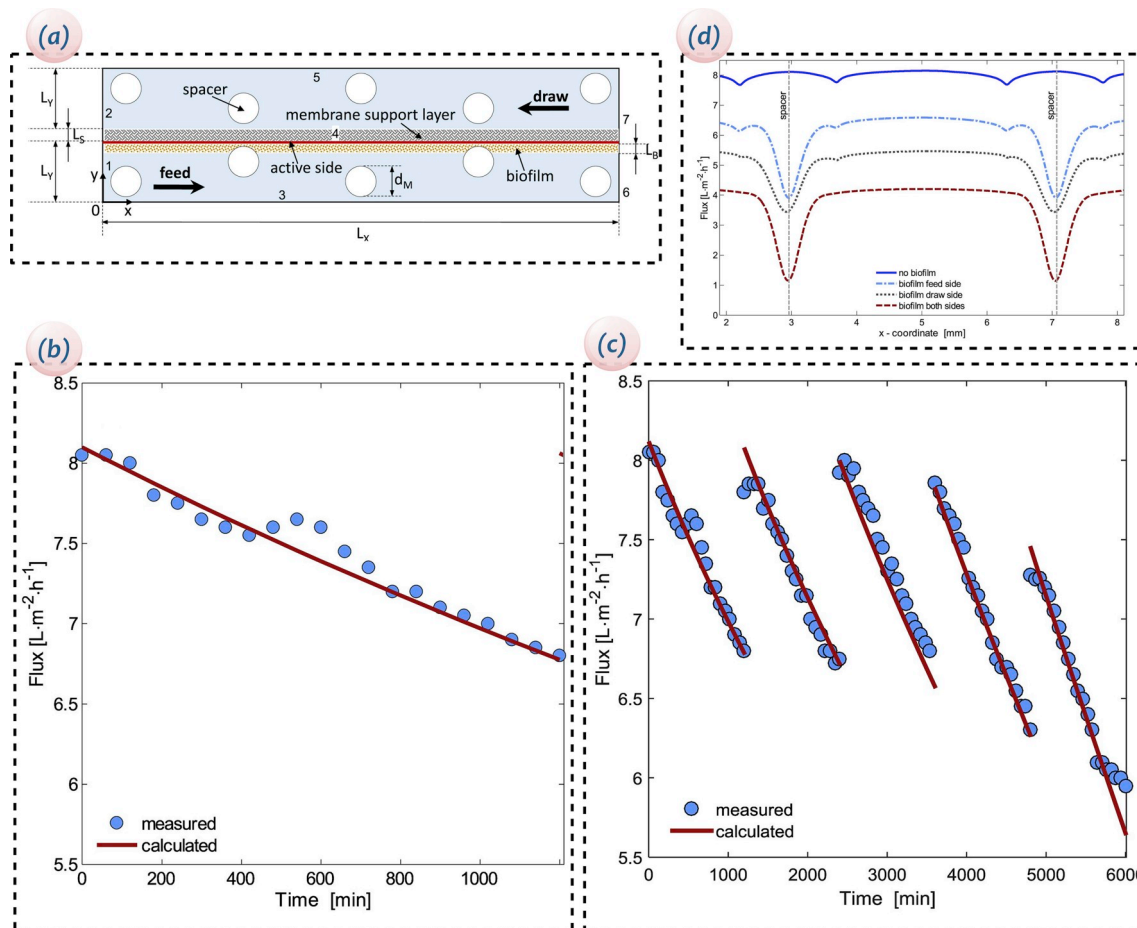


Fig. 3. (a) Illustration of the forward osmosis (FO) model geometry. The model space is divided to three domains: feed and draw channels (blue color, with bold arrows indicating the flow directions), membrane support layer (woven grey) and the membrane active layer (red line) is facing the feed channel side. The biofilm (yellow dotted zone) constitutes a fourth domain. (b) Combined impact of dilution and biofilm formation on flux in forward osmosis system operated in repeated batch mode, measured (blue circles) and modeled (red lines). (c) Calculated flux decline at different biofilm thickness on the membrane feed side, with spacers in the flow channels (Effect of different biofilm porosities (ϵ_B)). (d) Calculated water flux along the membrane: without biofilms (continuous blue line), with patchy biofilm (dotted blue line and dotted black) and with continuous uniform biofilm (dashed red line), on the membrane active side. The average water flux obtained was: 7.9 ($L \cdot m^{-2} \cdot h^{-1}$) (no biofilm), 6.4 ($L \cdot m^{-2} \cdot h^{-1}$) (patchy biofilm) and 6 ($L \cdot m^{-2} \cdot h^{-1}$) (continuous biofilm) [135], reproduced with permission from Elsevier. (For interpretation of the references to color in this figure legend, the reader is referred to the Web version of this article.)

than either attempting to prevent it completely or than removing an established biofilm. As mentioned in previous sections, the rate of microbial growth is dependent on nutrients and other environmental conditions. Pretreatment of the feed solution before filtration is one of the easiest means to postpone or decrease the intensity of biofouling, but it comes with capital and operational costs. Pretreatments can be applied chemically, e.g., by oxidation or by decreasing the concentration of the nutrients by precipitation (i.e., phosphate [131]), or physically, e.g., via filtration.

Regarding membrane cleaning strategy, while physical cleaning can be effective for the removal of organic/inorganic foulants, it is largely ineffective for biofouling reduction, regardless of the nature of the driving force [117]. Chemical cleaning of the FO membrane normally yields better results in terms of flux recovery than in RO systems, and it may represent a potential treatment for FO membrane maintenance and flux restoration in future large-scale applications [117]. The major drawback of chemical cleaning is its high cost and the required disposal of undesirable by-products to the waste stream that must be treated before discharge to the environment. Additionally, many cleaning chemicals can damage membranes. However, the resistance of extracellular polymeric substance (EPS) to physical cleaning dictates that intense chemical treatment may be necessary to clean membranes encased in EPS. Wang et al. used acidic cleaners coupled with alkaline

ones, and they found that a mixture of 0.1% NaOH or 0.1% sodium dodecyl sulfate (SDS) followed by 2% citric acid or 0.5% HCl can be utilized to remove foulants from the membrane surface [136]. Another method is the addition of acids, alkalis, and surfactants as well as oxidizing or chelating agents, such as chlorine, directly to the feed water before membrane treatment. Fujioka et al. applied the target concentration of 3–8 mg/L of chloramines for wastewater pretreatment before the FO process by CTA membranes. After 96 h of continuous membrane experiment, flux reductions of 6% and 35% were observed in the chloraminated and non-chloraminated FO system, respectively [137]. Despite the potential efficacy of chlorination for biofouling control, the potential formation of harmful and carcinogenic disinfection by-products must be considered [138]. Also, the presence of oxidants often diminishes the membrane lifetime, and therefore these substances must be removed from the feed before initiating the filtration process [137]. Wang et al. reported that H_2O_2 oxidizes the microbial community and converts them to more desirable organic substances, but high concentration and prolonged contact time of the surface with H_2O_2 can finally lead to membrane degradation. Therefore, it is of prime importance to control the concentration of oxidants and their contact time in membrane cleaning strategies. The results of SEM observations indicated that the optimum concentration of H_2O_2 and contact time for membrane cleaning should be 0.5% and 6 h [139].

Considering the difficulties generated by the application of biocides, other pretreatment strategies have been applied. Quorum sensing inhibition (QSI) can control biofilm growth. This method creates disturbance in a bacterial communication system. In 2017, Choi [140] used three natural organic molecules, namely, cinnamaldehyde (CIN), vanillin (VAN), and zingerone (ZIN), as QSI to control bacterial growth, resulting in decreased biomass accumulation on the membrane surface up to 68%, 41%, and 15% in the presence of CIN, VAN, and ZIN, respectively. Another biological control method is the use of bacteriophages to hinder or destroy biofilm development on membrane surfaces by infecting the host bacteria. However, bacteriophages can also inhibit the growth of bacteria necessary for wastewater treatment when the treatment train includes biological processes, and this phenomenon limits their application for pretreatment of wastewater.

Beside chemical pretreatments, physical approaches such as aerobic pretreatment are applicable. Zhang et al. investigated biofouling development in four FO stages under low (6.54 mg/l-(stage A) and 4.59 mg/l-(stage B)) and high aeration (8.95 mg/l-(stage C) and 8.53 mg/l-(stage D)) conditions. The results of this assessment demonstrated that biofilm formation in low aeration (average of stage A and B) were 31 and 12 times higher than those in high aeration stages (average of stage C and D), respectively.

Ozonation of feed water is also an important disinfecting method. However, ozone may produce assimilable organic carbon (AOCs), especially in the case of wastewater pretreatment, and cannot be applied to feed waters containing bromate, due to the formation of carcinogenic by-products [15]. Ultraviolet (UV) irradiation is reported as a common physical pretreatment for the inactivation of microorganisms. Despite being able to reduce AOCs, the use of UV irradiation is limited in UV-absorbing or light scattering solutions, and it is not effective for microorganisms that have been deposited on the membranes [15]. Overall, despite these difficulties, FO is usually reliable for the treatment of highly polluted waters and has lower fouling propensity compared to the RO process. Its cleaning procedures are also easier and require less time and effort, which makes some pretreatment strategies unnecessary in most cases [141].

3.2. Membrane surface functionalization for biofouling mitigation

On account of polyamide (PA)-TFC membrane higher water permeability and higher salt rejection compared to CTA asymmetric membrane, their fouling behavior has been the topic of the vast majority of recent studies. Apart from the hydrodynamic operation conditions, the FO fouling tendency is closely related to the membrane surface characteristics that dictate the interaction with foulants. In general, a fouling-resistant membrane should be hydrophilic, smooth, and negatively charged to prevent attachment of foulants to its surface. TFC PA membranes are intrinsically prone to fouling (especially biofouling) owing to their inherent chemistry in terms of relative hydrophobicity, high surface roughness, and the existence of carboxyl groups on the membrane surface [142]. Consequently, there is an essential need for the alleviation of fouling by innovative strategies to increase membrane lifespan. A variety of approaches have been implemented to functionalize the TFC PA membranes, including surface grafting of monomers or polymers, chemical coupling, and in situ growth. In these approaches, an agent chemically reacts with the functional groups of the surface, preferentially via covalent bonding. The membrane surface is initially activated with physical or chemical reactions and subsequently subjected to functionalization. It should be noted that the main feature of chemically functionalized surfaces is their long-term stability compared to those modified only by physical means [143]. In the case of application of functional nanoparticles, surface functionalization decreases the possibility of nanoparticle aggregation compared to protocol attempting to incorporate nanomaterials within the membrane structure [144,145].

As mentioned above, the physiochemical properties of the

membrane surface are the most important factor affecting cellular attachment. Surface hydrophobicity/hydrophilicity, charge and roughness are key parameters in biofouling control and mitigation, as they dictate the interaction between the membrane and the bacterial cells [146]. It is generally accepted that membranes with smoother and more hydrophilic surfaces provide better antifouling performance [147]. It has been reported that a hydrophilic surface imparts high surface tension and is prone to form hydrogen bonds with the surrounding water molecules to create a thin boundary of water between the membrane and the bulk solution. This layer can prevent the undesirable adsorption or adhesion of hydrophobic foulants on the surface of the membrane. Although the membranes with a rougher surface generally present lower hydrophilicity [148] the effect of surface functional groups on hydrophilicity is greater than that of surface roughness. Generally, inconsistent results have been reported in the literature on the relationship between biofouling and the roughness of the membrane surface [149, 150]. Higher surface roughness increases the total surface area accessible to bacterial cell accumulation and consequently biofilm formation. On the other hand, the presence of brushes or biocidal agents can improve the anti-biofouling tendency, even when these materials increase the surface roughness [151]. Although the approaches listed above result in a significant increase in hydrophilicity and decreased foulant-membrane interfacial adhesion forces, there is an emerging paradigm shift to develop fouling resistant TFC PA FO membranes. This shift is from passive (fouling resistance and fouling release) to active (off-surface antibacterial and on-surface antibacterial) antifouling strategies, which includes binding antibacterial agents to the TFC PA active layer [120].

The surface charge of the membrane is an important parameter in membrane fouling reduction when treating a feed with charged foulants [152]. In this context, one of the frequently used methods is to induce the same sign of charge to the membrane surface as that of the foulants. This approach results in electrostatic repulsion, thus decreasing deposition of the foulant [153]. Since bacterial cells are made up of phospholipids and lipopolysaccharides, they normally carry negative electrical charges in a pH range of 4–9 [154]. A negatively-charged surface is thus less prone to bacterial deposition [155]. Accordingly, it seems that the functionalization of the TFC PA layer with negatively charged agents can provide stronger electrostatic repulsion toward bacteria cells and hence lower biofouling tendency [156]. However, strong electrostatic attraction may also inhibit biofilm formation; many bacteria deposited on positively charged surfaces show little to no growth. This mechanism has been attributed to the strong, attractive force, hindering elongation and division necessary for bacteria growth [157].

Here, we review recent trends on the surface functionalization strategies with the aim of biofouling mitigation. Various aspects of different functionalization strategies are discussed in terms of simplicity/versatility, reproducibility, transport performance after modification, cost-effectiveness, and environmental side-effects.

3.2.1. Approaches for the surface functionalization of the TFC FO membrane

There are numerous studies reporting the presence of native negatively charged moieties on the surface of the TFC PA membranes. On account of incomplete crosslinking of the MPD and TMC monomers, the two main reagents used to synthesize dense and highly cross-linked polyamide films, amine and especially carboxyl functional groups are present on the surface of the freshly prepared TFC PA membrane. When the TFC PA membrane is soaked in an aqueous solution, the residual acyl chloride groups from the unreacted TMC monomers convert to carboxyls [158]. These negatively charged moieties at the surface may serve as reactive sites for subsequent binding of functional monomers, polymers, or nanoparticles [159]. For this reason, functionalization approaches to improve antifouling properties can potentially be facilitated on the TFC PA surface [160]. TFC PA.

3.2.1.1. Surface grafting. Surface grafting is a promising method and involves the covalent binding of grafted monomers or chains onto the membrane surface. The free radical and ionic active sites are generated via initiators on the TFC PA chain, and afterward, the radical ion reactions propagate to complete the polymerization with the target monomer. Generally, grafting reactions can be divided into two categories including grafting-to and grafting-from. The former consists of covalent attachment of reactive functional polymeric chains, while the latter comprises the initiation of grafting from the reactive species present on the surface. The grafting-from process can be chemically initiated via chemical, UV-irradiation, photochemical, and plasma-induced treatment, followed by the propagation of different reactive groups [161]. The grafting procedure depends on the chemical structure of the grafting agent, the chemistry of the surface, and the desired properties after surface treatment. Among all procedures, photochemical/redox-initiating grafting is less expensive, facile, and very convenient for industrial applications [162].

One example of grafting is discussed by Li et al. [163] where cheap and efficient photosensitizer porphyrin molecules (Por) were grafted to design photodynamic antibacterial TFC PA FO membranes. The porphyrin molecules were grafted on the TFC PA surface via the reaction between the amine group of porphyrins and the barboxyl groups on the TFC membrane. Due to the strong covalent bond between the porphyrin molecules and the TFC surface, the Por-TFC membrane showed good chemical and photo stability. The key concept in this antibiofouling strategy is that, when exposed to sunlight radiation, the porphyrin molecules can produce a large amount of ROS (singlet oxygen or other reactive oxygen species), which are highly toxic to bacteria [164,165]. In this regard, further progress in the development of the photo-bactericidal membranes offers good opportunities to address biofouling implications and pave the way towards practical FO applications.

3.2.1.2. Surface functionalization with the aid of EDC/NHS coupling reaction. N-ethyl-N-(3-(dimethylamino)propyl) carbodiimide (EDC) and Nhydroxysuccinimide (NHS) coupling reactions are widely used in membrane functionalization where amide bonds form between carboxyl groups and amine groups by the mediation of these two molecules. In these reactions, first, the carboxylic acids are activated by using carbodiimide with the aid of NHS followed by the amine addition [166]. The EDC/NHS coupling reaction has several advantages, including high conversion efficiency, mild reaction conditions, accessibility in an aqueous phase, and exceptional biocompatibility with minimal effect on the bioactivity of a target structure [167]. Applying similar strategies, a number of antibiofouling FO membranes were fabricated. Typical examples of such reactions were reported for functionalization with graphene oxide (GO) and carbon nanotubes (CNTs) [168], bound to the TFC membrane surface [169]. The GO-functionalized TFC surface exhibited considerable antibacterial activity, most likely attributed to the physical disruption of the bacterial cell membrane, membrane stress, oxidative stress, and/or release of reactive oxygen species (ROS).

3.2.1.3. Other chemical coupling approaches. Chemical coupling is a commonly used approach to tune the membrane surface with a variety of target-designed functional groups. Also in this case, free carboxylic acids and primary amine groups present on the PA layer serve as reactive sites for surface functionalization via chemical reaction or coupling. Substitution, addition, adhesion, and oxidation are the main methods applied for chemical coupling. Carboxylation [170] hydroxylation [162] and amination [171] are frequently used approaches for altering the TFC PA layer chemistry [172]. However, the main challenge of chemical functionalization is that the modification agent or the reducing agent may deteriorate the selective layer integrity and reduce the membrane rejection [173]. A reducing agent, e.g., sodium borohydride, is a compound or element that donates an electron to another chemical

species in a redox reaction. The main drawback of reducing agents is their high reactivity, which possibly pose environmental and biological risk [174]. Moreover, they may lead to surface segregation and hence deteriorate the membrane integrity by repeated usage. Therefore, the utilization of green, non-toxic, and eco-friendly chemicals for stabilizing the functional groups or nanoparticles on the surface is a major challenge associated with reducing agents. Following these criteria, green reducing agents, such as ascorbic acid [175,176] and citric acid [177] have been reported to eliminate some side-effects and minimize the generated waste.

3.2.1.4. In situ growth of metal nanoparticles. In situ growth is a simple procedure in which metal nanoparticles (NPs) anchor on the TFC PA membrane surface. This approach is frequently applied for the functionalization with Ag [178] and Cu [179] NPs. The interaction between the metal precursor and the PA layer is of prime importance as it regulates the particle size distribution, stabilizes the interaction strength, and prevents NP release. Generally, two procedures, namely, classical generation-before and generation-after, have been reported for in situ formation of metal NPs on the membrane surface. The generation-before method involves the binding of pre-prepared metal NPs to the surface [174]. Although this method is beneficial in controlling the size and morphology of the metal NPs, the improper dispersion of NPs on the surface restricts its application. The fabrication procedure in case of generation-after is instead based on the in-situ formation of metal particle from their precursor salt solutions in the presence of a chemical reducing agent or thermal driving force. Typically, this procedure involves two steps: (i) interaction of metal ions with the pretreated surface and then (ii) reduction of metallic ions to metal NPs with the aid of a reducing agent or an external stimulus, such as thermal inducement or UV irradiation. Before applying the chemical reducing agent, the residual amount of metal precursor should be removed from the surface to ensure the all the NPs are covalently bound to the reactive sites of the surface. Since NPs are thus present on the surface, where fouling develops, direct exposure activates two biocidal mechanisms, one based on direct contact and the second on the release of metal ions. Moreover, this approach offers a higher yield of NPs dispersion and prevents aggregation compared to attempts of NPs incorporation into the TFC membrane.

3.2.1.5. Self-assembly via coordination chemistry. Self-assembly via coordination bonds is an interesting subset of in situ growth in which, e.g., metal organic frameworks (MOFs), are grown at the surface following the chemical reaction of metals and ligand species. Self-assembly offers a powerful platform for the design of the well-defined architecture of functional materials at the molecular level with various shapes and structures. The physicochemical properties of the surface play an essential role in regulating the quality of crystal formation during the heterogeneous growth of MOFs on the substrate [180]. For this reason, the substrates are usually subjected to physical or chemical pretreatment to improve the reactivity of nucleation sites [181]. Since the synthesis of MOFs involves the coordination between metal ions and organic ligands, surfaces modified with amine and carboxyl functional groups are beneficial for MOF crystal growth, as these moieties can serve as covalent ligands in the MOF structure [182,183]. The carboxyl functional groups present on the TFC surface offer the reactive sites for subsequent nucleation and growth of MOFs. The MOF seeds are first anchored on the reactive sites of the surface, followed by the growth of the MOF framework to form a thin film or individual crystals. Coordination and H-bonding interactions are the main mechanisms responsible for MOF structure attachment. The metal and ligand act, respectively, as the core and the linker to form coordinate covalent bonds of the MOF crystals. Most of the surface functionalized with MOFs are applied in gas separation, few reports involve water separation, and very limited research relates to TFC PA membranes for FO application [184,185].

The emergence of these nanostructures is regarded as one of the most

promising approaches for antibiofouling TFC PA membranes. MOFs are porous nanostructures in which the active cores are the same or similar to those present in metal oxide NPs [186]. The metal ion release at the boundary layer of the membrane surface acts as a source of biocides [187] and creates an inhibition zone [188]. The key advantage of MOFs compared to inorganic microporous compounds is that their structure is highly tunable by applying diverse metal centers or altering the organic ligand [186]. Antibacterial MOFs have thus the advantage of broad-spectrum antibacterial activity against both Gram-negative and Gram-positive bacteria, high efficiency, and long lasting stability [189, 190]. Nevertheless, limited approaches can be applied for MOFs functionalization since the PA layer is intrinsically sensitive to high temperature and harsh solvents [191]. As a result, developing a well-designed approach with minimal risks to membrane integrity, performed at ambient temperature, with relatively rapid precipitation, and employing green solvents, is necessary for the future of MOFs-functionalized membranes.

Surface functionalization strategies provide tailored control and target design of surface characteristics, generating valuable frameworks with improved or novel properties, such as superior antimicrobial activity. Regardless of the increasing interest in applying biocidal agents for the development of antibiofouling membranes for water purification, a practical strategy remains challenging. Moreover, the traditional contact-killing antibacterial strategy is restricted to surface accessibility, and the antibiofouling effect decreases once the membrane surface is completely covered with a fouling layer. However, compared to the bulk incorporation of antibacterial agent or nanomaterials, surface functionalization has minimal influence on the intrinsic performance of the TFC PA membranes. Additionally, the possibility of NPs regeneration after their release is practically very low when they are embedded within the polymer matrix. Considering its simplicity and prolonged stability, the surface functionalization of TFC PA membranes with antibiofouling materials is a dominant trend today and has important practical implications for water treatment. While several studies discuss the strong antibacterial activity of biocidal nanostructures, many of the proposed methods do not address all the limitations related to industrial-scale fabrication and operation of the membranes. Applying biocidal nanostructures on the TFC PA surface necessitates mild, facile, rapid, low-cost, and reproducible procedures without the need to disassemble the membrane module or sacrifice the membrane intrinsic transport properties.

3.2.2. Different types of materials applied in surface functionalization

Generally, antibiofouling strategies can be divided into three main categories, including (i) microorganism antiadhesion, (ii) bactericidal activity, and (iii) integration of antiadhesion with bactericidal approaches [184]. Strategies to decrease adhesion forces between bacteria and membrane surface comprise the attainment of superhydrophilic or superhydrophobic surfaces. Superhydrophilic surfaces are often obtained with zwitterionic materials by promoting the formation of a hydration layer on the surface leading to a mitigation of the bacterial adhesion due to non-specific interaction [173]. Approaches that have been developed to functionalize membrane surfaces with zwitterions include surface grafting [192,193] and polydopamine (PDA)-assisted deposition [194]. Several studies have reported zwitterion functionalized TFC PA surfaces for antifouling FO applications without extensive characterization of the antibiofouling behavior [195–197]. Liu et al. [198] prepared a highly antifouling TFC PA membrane by grafting a controlled architecture zwitterionic polymer brush via atom transfer radical polymerization (ATRP). By bioinspired catechol chemistry, the initiator molecules of the polymerization reaction were embedded on the membrane surface and resulted in the grafting of a dense zwitterionic polymer brush layer. The functionalized membrane surface exhibited decreased roughness, improved hydrophilicity, and larger negative charge. Moreover, the zwitterionic-modified membrane presented remarkably lower (one order of magnitude) foulant-membrane

interaction forces compared to that of the pristine TFC membrane. This behavior led to considerably reduced adsorption of both proteins and bacteria, providing a highly fouling resistant surface. The water permeability of the modified membrane decreased somewhat while the salt permeability increased after the formation of the zwitterionic polymer brush layer on the surface. However, precise control of the thickness of the modifying layer via ATRP allowed further optimization of the membrane transport properties without sacrificing antifouling properties (Fig. 4).

Other than zwitterions, several materials, such as PEG-based and natural biopolymers, e.g., polydopamine and chitosan, have been applied to prepare such hydrophilic membrane surfaces with antibiofouling properties [199,200]. Lu et al. [201] chemically modified the surface of the TFC membrane with Jeffamine, an amine-terminated poly (ethylene glycol), to enhance organic fouling resistance against synthetic wastewater containing alginate. Castrillón et al. [202] functionalized the TFC FO membrane surface with PEG to reduce organic (alginate) fouling, where a second interfacial reaction was adopted between ethylenediamine and free carboxyl groups to facilitate PEG grafting.

Liu et al. in Ref. [203] proposed a novel strategy to functionalize the TFC surface with superhydrophilic 3D hyperbranched polyglycerol for simultaneous improvement of both transport performance and antifouling properties (BSA protein adsorption). Additionally, Shen et al. [204] grafted polyethyleneimine (PEI) to the TFC FO surface to reduce alginate fouling. In a similar attempt, Bao et al. [205] prepared poly-amidoamine (PAMAM) dendrimers grafted to the TFC surface to simultaneously increase ammonia selectivity and antifouling properties of the FO membrane for resource recovery from domestic wastewater.

Polydopamine (PDA) is a bio-inspired polymer that can robustly attach on the TFC PA membrane surface with strong interactions [206, 207]. The PDA layer can form on any surface via in situ spontaneous oxidative self-polymerization of dopamine under mild conditions, via a range of interactions including covalent bonding, coordination, hydrogen bonding and π - π stacking [208,209]. The presence of the residual catechol, quinone, and amine functional groups on PDA offer reactive sites for further treatment with functional molecules [210].

Surface modification with dendrimers, highly branched molecules with a symmetric repetitive structure, provides a high density of hydrophilic functional groups on the membrane surface and diminishes protein and bacteria adsorption [211]. Moreover, a 3D star-like nanostructure of dendrimers exerts high steric hindrance to foulants and forms an antifouling layer without significantly sacrificing the water permeability [204], as intermolecular cavities facilitate the transport of water molecules. Bao et al. [212] grafted PAMAM dendrimers (generations 0, 1, 2 and 3) with abundant easily-protonated amine functional groups on the TFC PA FO membrane surface. The grafting approach was accomplished by forming chemical amide bonds between the unreacted carboxyl groups on the PA surface and amine groups of PAMAM dendrimer. This covalent bond rendered the TFC PA ζ -potential highly negative and imparted superior hydrophilicity, to simultaneously improve the ammonia rejection and antifouling performance of the FO membrane for treating domestic wastewater (Fig. 5). Although providing effective ammonia selectivity is challenging because of the similar polarities and hydraulic radii of the ammonium cation and water molecules, grafting PAMAM dendrimers enabled the FO membrane to achieve high ammonia rejection of 98.2% and significantly reduced reverse salt flux due to the electrostatic repulsion between the protonated amine layer and the compound. The antifouling properties of the grafted membrane were enhanced owing to the hydrophilic nature of amine groups and the electrostatic repulsion to positively charged ions.

Membrane surface modification via PDA grafting can effectively alter the surface wettability [213] and decrease the adhesion energy for organic foulants' deposition [213,214]. The protonated amine groups possibly exert antibacterial activity, as they cause cell lysis in contact with the bacteria cell membrane [215]. Owing to the simplicity of this

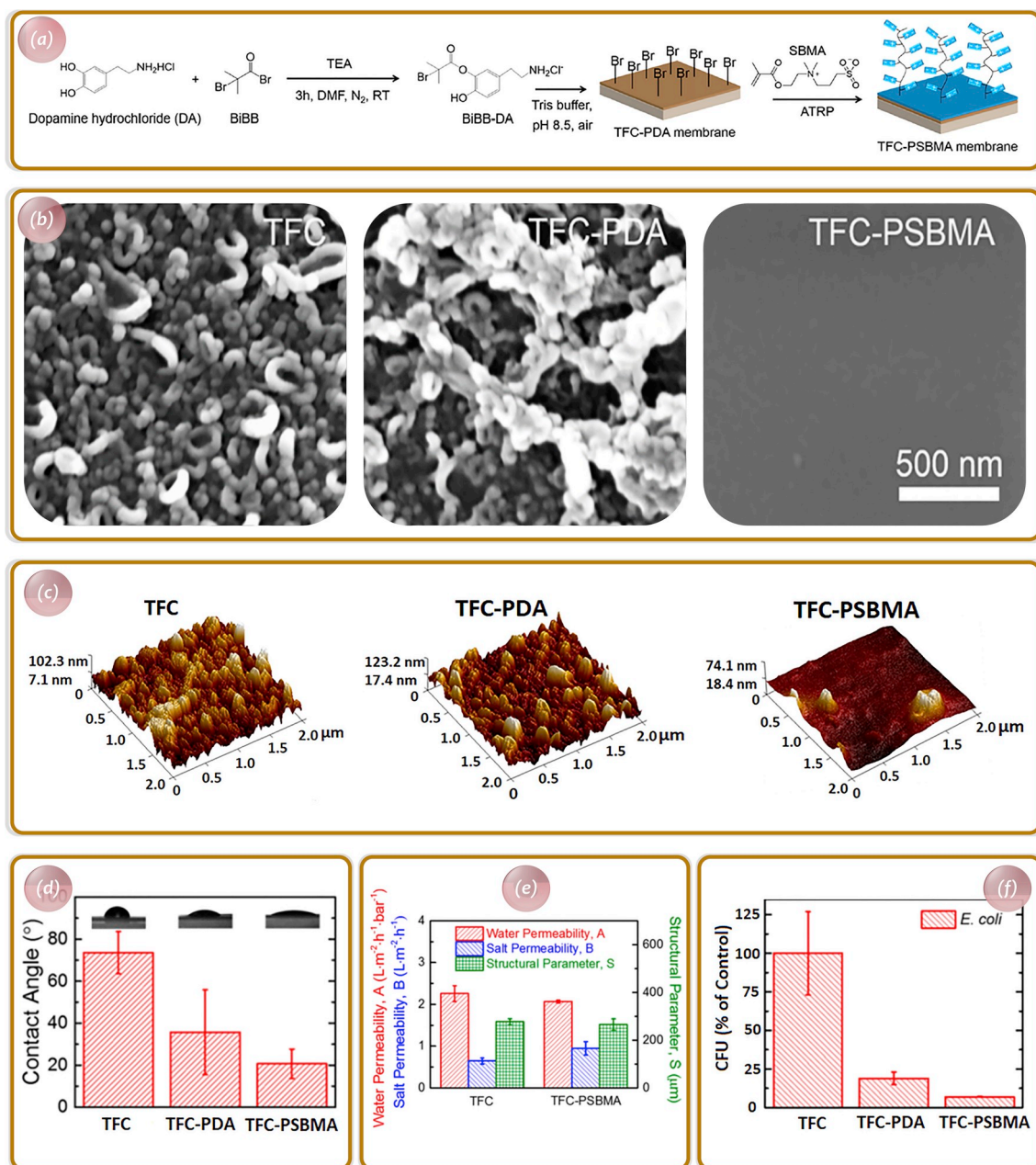


Fig. 4. a) Schematic illustration of TFC membrane grafting by zwitterionic polymer via ATRP in which BiBB stands for α -bromoisobutyryl bromide, TEA for triethylamine, SBMA for [2-(methacryloyloxy)-ethyl]-dimethyl-(3-sulfopropyl)ammonium hydroxide (also named sulfobetaine methacrylate), and PSBMA for poly (sulfobetaine methacrylate), b, c) SEM and 3D AFM images of the pristine TFC, TFC-PDA, and TFC-PSBMA membranes, d) Water contact angles, e) transport and structural properties of pristine TFC and TFC-PSBMA membranes, and f) *E. coli* adhesion of the of pristine and modified membranes after 3 h contact time [198]. reproduced with permission ACS publication.

type of modification to improve the fouling resistance, PDA was applied to several TFC PA FO membranes to suppress biofouling [216]. Li et al. [217] evaluated the biofouling behavior of cellulose triacetate (CTA) surface-modified with PEG-grafted PDA (PDA-g-PEG) in a submerged forward osmosis membrane bioreactor (FO-MBR) over 61-days. At first, the PDA was applied to coat the CTA surface, and then mPEG-NH₂ was separately grafted on the PDA-coated CTA to improve the antifouling properties and transport performance (A of 0.45 LMH/bar and B of 1.91×10^{-8}). The PDA-g-PEG-modified membrane displayed enhanced surface hydrophilicity, lower flux decline, and hence better antifouling ability than that of the pristine CTA membrane in FO-MBR.

Bactericidal activity is dominantly obtained with antibacterial nanomaterials, such as silver, copper, and carbon-based nanomaterials

[218]. This strategy is classified into two categories, namely, release killing and contact killing. Whereas both result in toxicity to microorganisms, sometimes these two principles are combined to accomplish synergistic effects. The long-term bactericidal activity can be attained by exploiting the controlled release of biocidal agents [219] and the properties of rechargeable surfaces [218]. The release killing strategy provides antibiofouling effect via leaching of the antibacterial agents from the membrane surface to the environment of interest, whereby they attack planktonic cells deposited on the membrane surface [220], for example by generating reactive oxygen species (ROS) formed through catalytic reactions [221,222]. This mechanism is commonly proposed to account for the off-surface active antifouling properties of silver-based nanomaterials [223].

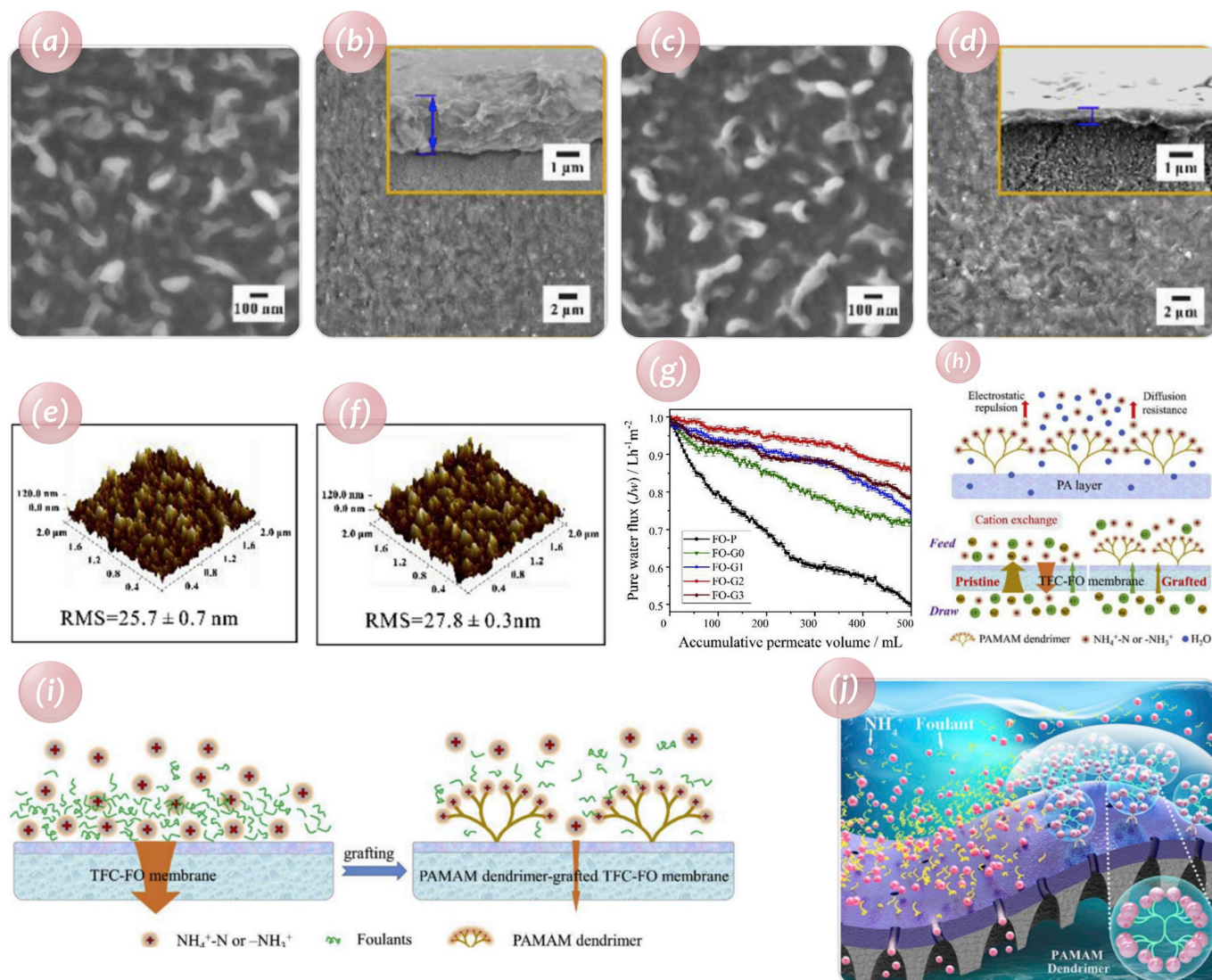


Fig. 5. (a, and b) SEM images of the pristine membrane, (c, and d) SEM images of PAMAM dendrimer-grafted TFC-FO membrane, (e) AFM images of pristine membrane, (f) AFM images of the PAMAM (generation 2) dendrimer-grafted TFC-FO membranes, (g) normalized water flux reduction trend, (h) schematic illustration of the TFC-FO membrane rejection of NH_4^+-N and the mixture of NaCl with NH_4Cl as the feed solution, (i) schematic diagram of the effect dendrimer grafting on the membrane fouling, (j) Schematic PAMAM dendrimer's grafting on the TFC-FO membrane [212]. reproduced with permission from Elsevier.

Silver-based biocides are the most extensively used nanomaterials and are based upon the release of free silver ions (Ag^+) [224,225]. The toxicity of Ag NPs against bacterial strains can be related to the quantity of Ag^+ ions released [226] and their reactivity can be increased by manipulation of their physicochemical properties, such as particle size [227] and morphology [228]. It is generally believed that the antibacterial mechanism of Ag NPs is multifold: (i) direct adhesion of Ag NPs to bacterial cells, with physical damage to the cell membrane, (ii) release of Ag^+ and then infiltration into the bacteria, and (iii) indirect generation of reactive oxygen species by Ag NPs and Ag^+ , causing defects to the bacteria cell structure and eventually preventing biofilm formation on the surface (Fig. 6) [229,230]. It is preferable to locate the silver NPs near the membrane surface, where they have maximum interactions and direct exposure to the bacteria [231]. This way, the two antibacterial mechanisms, release killing and contact killing, occur simultaneously. Significant progress has been made in the development of in situ functionalized Ag NPs on the TFC PA surfaces, and some of the recent research is reported here. The antibacterial performance of Ag NPs functionalized TFC PA membranes essentially depends on the NPs size, distribution, and stability [232].

Liu et al. [231] developed a simple and facile approach to covalently bind Ag NPs on the TFC PA surface via layer-by-layer interfacial polymerization. Their findings revealed that this approach provided less influence on surface roughness and overall charge, yielded slight improvements in water flux and salt rejection, lowered the Ag release rate, and provided stability during filtration, excellent antibacterial properties and hence, high resistance to biofouling. Moreover, Liu et al. [178] functionalized Ag NPs on hydrolyzed polyacrylonitrile (PAN) membrane surfaces and attained high antimicrobial activity against *E. coli* for 14 days under laboratory conditions, imparting biofilm growth resistance (Fig. 7). Although there are many advantages to functionalizing membrane surfaces by silver NPs, researchers still doubt about their safety in commercial use.

Copper-based NPs (Cu NPs) are regarded as a valuable biocide for the preparation of antibacterial membranes, with lower costs than silver [234]. In addition to contact killing, Cu NPs promote the formation of hydroxyl radicals that destroy essential proteins and DNA inside bacteria cells [235]. Similar to silver ions, Cu NPs show rapid release if their deposition on the surface is only physical. This release mechanism reduces its antibacterial property and poses contamination problems for

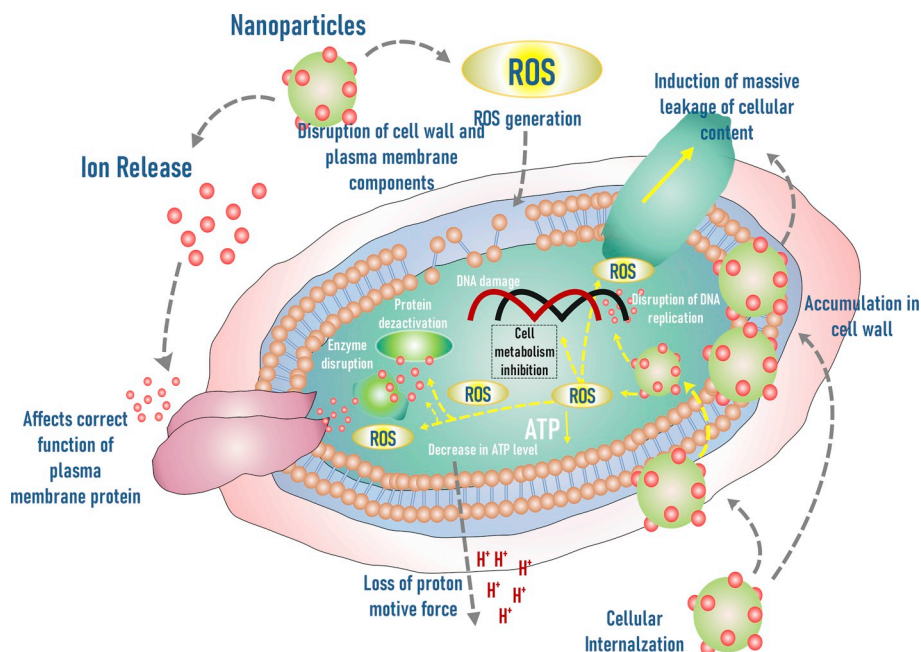


Fig. 6. Antibacterial mechanism of metal and metal oxide containing systems [233]. reproduced with permission from Elsevier.

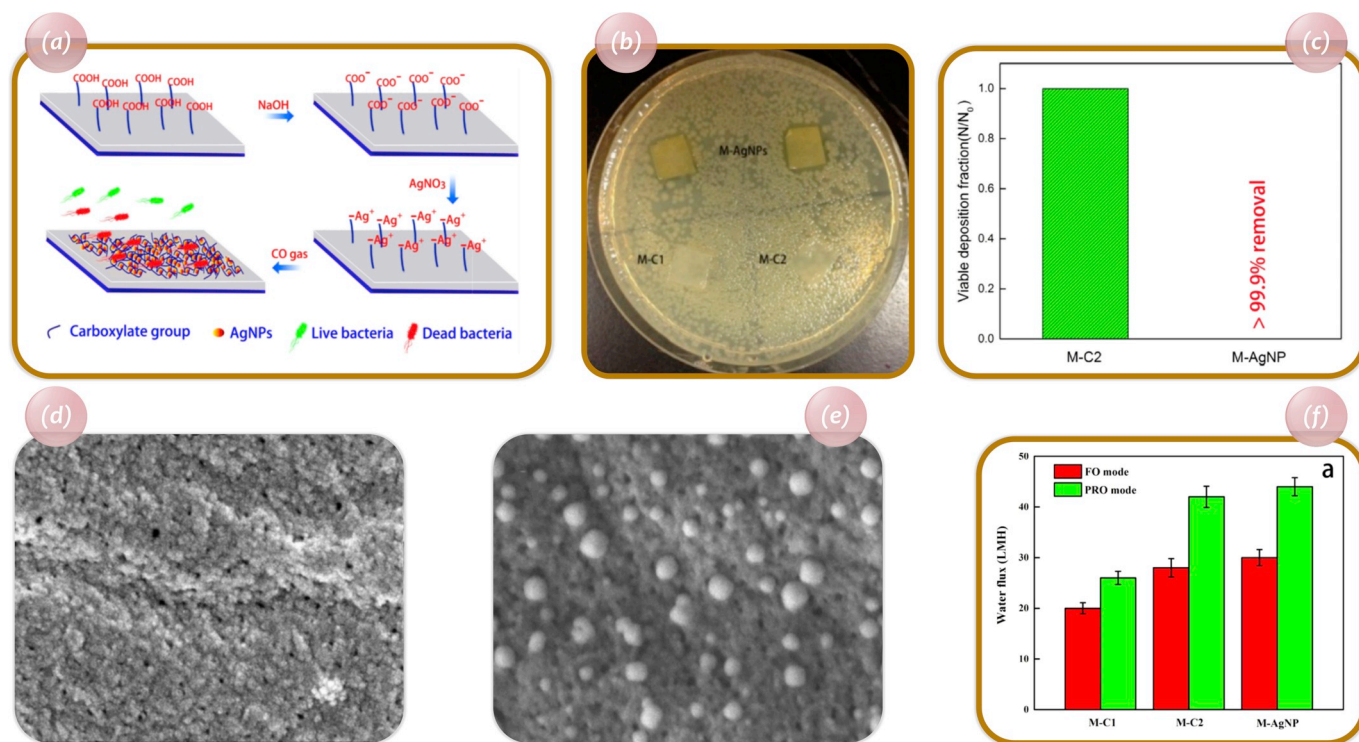


Fig. 7. (a) Schematic illustration of deposited AgNPs on the TFC membrane formed on PAN substrate (M-AgNPs), (b) CFU test of TFC (M-C1), Na-OH-modified TFC (M-C2) and Ag-NPs functionalized (M-AgNPs) membrane, (c) static antibacterial adhesion behavior of M-C2 and M-AgNPs membranes, (d, and e) FE-SEM images of modified-TFC membranes and (f) water flux of membranes [178]. reproduced with permission from ACS Publications.

the feeds and permeates. Therefore, Cu NPs should be stably immobilized on the surface to alleviate their release rate and guarantee long-term antibacterial activity. Liu et al. developed a simple one-pot procedure for the chelation of copper ions on the surface of the TFC PA membrane by soaking the membranes in a dopamine (DOPA) solution containing copper (II) [236]. This cost-effective approach offered a versatile method for the construction of long-term biofouling-resistant surfaces. The PDA-Cu layer affected the transport properties of the

membranes and decreased the water flux of the TFC FO membrane up to 28%. On the other hand, the PDA-Cu functionalized surface demonstrated high antibacterial activity against *S. aureus* (up to 97%) with very little Cu leakage into the permeate. Both findings suggest that the PDA-Cu functionalized membrane can yield interesting long-term antibacterial activity against Gram-positive bacteria [237].

MOFs are an outstanding class of porous nanostructure compounds that have emerged as promising structures compared to traditional

inorganic microporous structures, such as zeolites, due to their hybrid organic/inorganic structure, high surface area, their highly tunable molecular structure and adjustable pore size [238]. MOFs can serve as the reservoir of biocidal metal ions in which the continuous gradual release of the metal ions by (bio)degradation of the framework provides sustainable antibacterial activity with high productivity [239]. Besides, the organic ligand applied for the construction of the framework may itself have antimicrobial activity, and therefore a combined effect may be attained [240]. Recently, Ag-MOFs were used to functionalize TFC PA membranes for FO applications [241]. A two-step deposition procedure was proposed for covalent binding of Ag-MOFs to the TFC PA FO membrane at ambient temperature, applying only two chemical reagents (silver salt and organic ligand) without additional stabilizing/-reducing agent (Fig. 8). Confocal microscopy analyses revealed that Ag-MOFs exerted strong antibacterial activity to the membrane surface,

leading to more than 99% reduction of live bacteria. The functionalization procedure generated a uniform distribution of Ag-MOFs on the active sites of the PA layer, rendering the membrane surface highly antibacterial. As mentioned, surface depletion of metals is considered one of the major drawbacks to the long-term operation of surface functionalized membranes [226]. The regeneration of metal NPs or MOFs after release or leaching has been proposed as a strategy to preserve the antibacterial activity of the membrane surface [223].

Carbon-based nanomaterials (CNTs and GO) also demonstrate antibacterial activity as well as low cytotoxicity towards mammalian cells. When comparing reduced graphene-oxide (rGO), graphite, and graphite-oxide against *E. coli* under similar conditions, graphene-oxide (GO) showed the highest antibacterial activity [242,243]. The generation of reactive oxygen species and penetration of cell membranes are the proposed antibacterial mechanisms of carbon-based nanomaterials

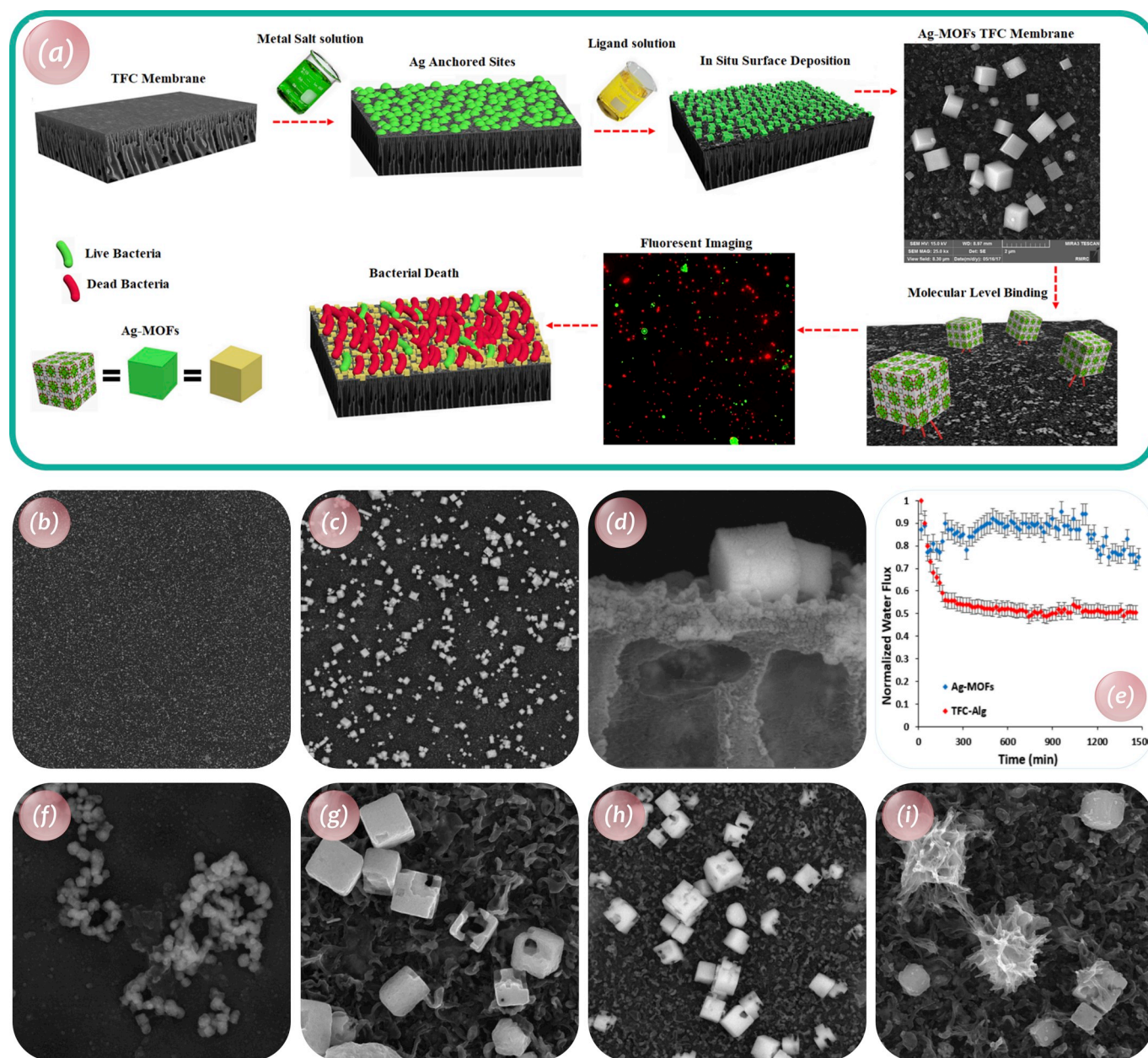


Fig. 8. a) Schematic illustration of Ag-MOFs functionalized TFC membrane, b, c, d) surface and cross sectional FE-SEM images of in-situ functionalized Ag-MOFs TFC membrane, e) flux decline during dynamic biofouling test by *E. coli* of the nascent TFC and Ag-MOFs functionalized membranes, f) FE-SEM images after 6 h FO test, g) after 7 months soaking in water, h) after harsh sonication and after regeneration via the same procedure applied for initial in-situ surface functionalization [241]. reproduced with permission from Elsevier.

[244].

This type of antibacterial activity is stable over time and prolongs the antimicrobial property of the functionalized surface over time without leaching [245]. Under specific conditions, GO has a higher antibacterial ability compared to CNTs [246]. The existence of water transport channels in CNTs and nanochannels between graphene nanosheets [247, 248] may provide enhanced water permeability while exerting anti-biofouling properties.

Although covalent binding of CNTs to PA surfaces is quite rare, Tiraferri et al. proposed an innovative such approach [168]. After the modification of single wall carbon nanotubes (SWNTs) to maximize their cytotoxicity and achieve dispersion in aqueous solution, the EDC/NHS coupling reaction generated covalent amide bonds, activating the carboxyl functional groups of both the membrane and the modified SWNTs to maximize coupling with ethylenediamine. The SWNT-functionalized surface demonstrated high performance in the water separation process and cytotoxicity against *E. coli* bacterial cells (up to 60% inactivation of attached bacteria within 1 h of contact time). This result indicates that covalently bound SWNTs TFC are a potential candidate to delay membrane biofouling.

GO nanosheets are highly efficient antibacterial agents and contain multifunctional oxygen-containing groups such as hydroxyl, carboxyl, carbonyl, and epoxy moieties. The physicochemical properties of GO nanosheets, including sheet morphology, size and size distribution, oxygen-containing functional groups density, electronic mobility, and carbon radicals, can significantly affect their antimicrobial properties. Despite widespread interest, the exact antibacterial mechanisms remain controversial [249]. The main mechanisms proposed for bacteria toxicity can be classified as: sheet attachment, cell membrane penetration, cell wrapping, phospholipid extraction, and oxidative stress [250].

Perreault et al. [245] irreversibly functionalized biocidal GO nanosheets on the TFC PA membrane to decrease the extent of bacterial growth, using a simple amide coupling between carboxyl groups of GO and carboxyl groups of the PA layer (carbodiimide chemistry). The functionalized GO surface resulted in 65% bacterial inactivation after 1 h of contact time without any adverse effect on the intrinsic membrane transport properties. Hegab et al. [251] grafted GO to the surface of the TFC PA layer via a poly L-Lysine (PLL) intermediary using two different approaches of either layer-by-layer (GO/PLL-LBL) or hybrid (GO/PLL-H) grafting to enhance the biofouling resistance of the FO membrane. The GO/PLL-LBL tuned surface presented higher hydrophilicity and smoothness, as well as a 99% reduction in surviving bacteria (Fig. 9). Besides, the hybrid modification (GO/PLL-H) also diminished the reverse salt flux compared to that of the pristine membrane, improving the membrane salt selectivity. In another attempt, Perreault et al. assessed the long-term biofouling tendency of the GO functionalized FO membrane (GO-TFC). The GO-TFC membrane demonstrated increased surface hydrophilicity and improved antimicrobial activity without altering transport properties, and achieved a 36% reduction of *P. aeruginosa* viable cells after 1 h of contact time. The decreased membrane biofouling (50%) after 24 h-long treatment of synthetic wastewater supplemented with *P. aeruginosa* was ascribed to the reduced attachment of the microbial community on GO-TFC surface. The deposited cells on the membrane surface were inactivated and formed a layer of dead cells on GO-TFC that limited biofilm formation.

These findings highlight the potential of GO nanosheets in biofouling mitigation of FO membrane applications, as the design criteria can be desirably optimized to minimize the adverse effect of GO nanosheets on the membrane transport properties. However, there are many challenges to be overcome in the use of graphene-based materials, such as the attainment of the correct GO sheet size, oxidation density, and issues related to long term stability.

Although metal NPs can result in satisfactory antibacterial activity, their instability and the risk of gradual dissolution has greatly restricted their practical application in water treatment. This challenge can potentially be addressed by immobilizing or decorating metal NPs onto

larger support to construct nanocomposites [252]. These carriers enable nanocomposites with multifunctional characteristics such as enhanced antifouling, permeability, and mechanical strength compared to those currently in use [253]. Promising carriers for metal NPs include CNTs [254], GO [255], Zwitterions [256], and PDA [257]. Considering the diversity and complexity of membrane fouling, this strategy provides a new way to combine antiadhesive and antibacterial features for a long-lasting antibiofouling performance.

An example of this approach has been reported for grafting zwitterionic poly(sulfobetaine methacrylate) brushes with the embedment of biocidal silver nanoparticles to attain augmented anti-adhesion and bacterial inactivation [226]. This combined effect has been associated to a more pronounced antibiofouling resistance, also leading to minimum flux reduction during membrane operation. Even though the results of such modifications are very interesting, the complexity of the procedure and equipment involved can be a major obstacle.

GO-based nanocomposites as functional materials are fabricated by immobilizing metal NPs onto the GO nanosheet [258]. The existence of both ionic groups and aromatic sp^2 domains enable GO to serve as a nucleation site and participate in the bonding interactions with biocide metal ions, such as Ag NPs [259]. Additionally, the carboxyl groups present at the edge of the GO nanosheets also combine with the metal ions [258]. Metal ions deposit on the GO nanosheets via the cooperation of π - π interactions, hydrogen bonding, and Ag-O coordination [260].

The metal release rate of GO-based nanocomposites containing metal NPs is considerably lower than for pristine metal NPs. The TFC PA membranes functionalized with GO-based nanocomposites inherit both the effect of nanosheets and that of metal NPs, opening a new pathway for increasing antibiofouling resistance and separation performance without sacrificing the water permeability [261]. Soroush et al. [262] executed pioneer work to covalently bind the Ag NPs immobilized GO nanosheets onto the PA FO surface and imparted enhanced antibacterial activity to the membrane due to the synergetic effect of the metal-GO hybrid. The functionalized TFC surface displayed hydrophilic properties (contact angles below 25°) and significant antibacterial activity (over 96% in a static test) without adverse effect on the membrane transport properties.

In another relevant attempt, Soroush et al. reported [263] the functionalization of TFC FO membranes with Ag-GO nanocomposites to impart biocidal properties to the membranes. The surface of the TFC membrane was initially functionalized with GO nanosheets via EDC/NHS coupling reaction, and then the Ag NPs was formed on the surface using sodium borohydride as the capping agent. The GO nanosheets acted enhanced silver loading with higher stability while decreasing the ion release rate. All these features improved the surface hydrophilicity and antibacterial activity. Besides, the Ag NPs regenerated membrane, following seven days of previous depletion, demonstrated 75% antibacterial activity. In a similar study, Faria et al. [264] functionalized TFC with Ag-GO nanocomposite employing the EDC/NHS approach to prevent bacterial attachment and biofilm growth on the membrane surface. The carboxyl functional groups on the Ag-GO nanocomposite were covalently bound to carboxyl groups of the TFC surface via a crosslinking reaction. The TFC-Ag-GO membrane exhibited 80% antibacterial activity against the attached *P. aeruginosa* microorganisms and showed a promising antibiofouling property during dynamic biofouling experiments (30% water flux decline) without sacrificing the membrane intrinsic transport properties.

Liu et al. [218] developed a simple and facile approach for the in situ generation of Ag NPs on the PDA coated TFC FO membrane with sustainable antibiofouling activity. Ag NPs formed on a PDA layer by a simple dip-coating procedure and displayed long-term silver release with the capacity to be regenerated multiple times after depletion. The functionalized membrane had strong, sustainable bactericidal efficacy against both Gram-negative and Gram-positive bacteria. Although the PDA layer exerted some mass transfer resistance translating into flux loss, the transport properties may be further optimized by controlling

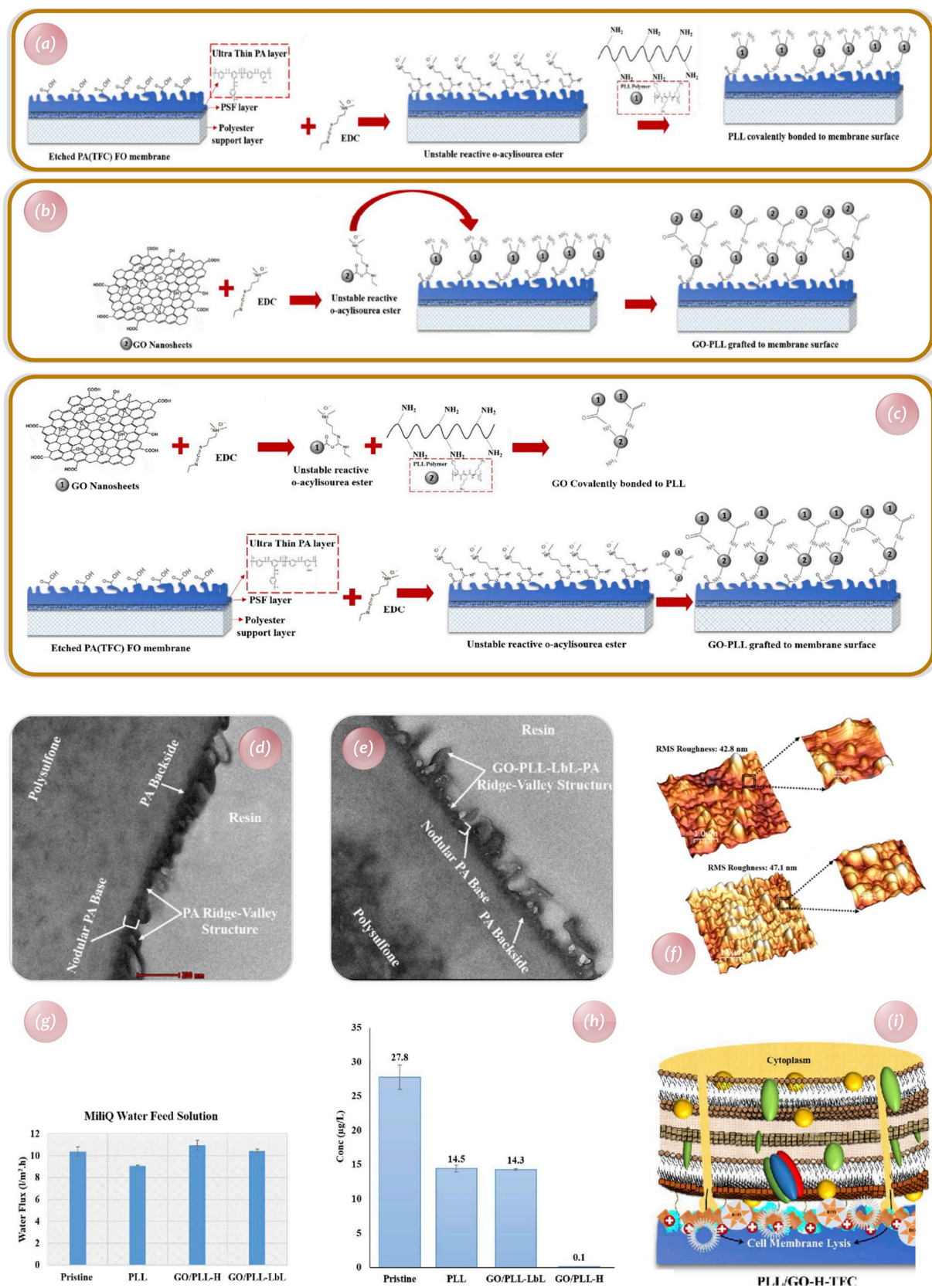


Fig. 9. (a, b, and c) The schematic chemical reaction of PLL, GO/PLL-LbL and GO/PLL-H TFC functionalized membranes, (d, e) TEM and (f) 2D AFM images of pristine and GO/PLL-LbL membranes, (g) Water flux and (h) ATP concentrations of the membranes and (i) Schematic illustrating of the bacterial toxicity mechanism [251]. reproduced with permission from ACS Publications.

the thickness of the PDA layer and the silver growth. Additionally, the facile regeneration of Ag NPs on the PDA surface provides a new direction for the development of sustainable, long-term antibiofouling TFC PA FO membranes. Qi et al. [265] modified the surface of the TFC PA FO membrane with a passive PDA coating and then deposited active antibacterial Ag NPs on the PDA layer (Ag-PDA). The active Ag-PDA membrane demonstrated both excellent antiadhesive and antibacterial properties in static antibiofouling experiments (33.3% and 97.0% of antibacterial activity for PDA and Ag-PDA coated membranes, respectively). More importantly, Ag NPs were stable on the surfaces, and the Ag-PDA membrane revealed 96.1% antimicrobial activity and lower water flux reduction (0.5% water flux decline) after 24 h dynamic cross-flow operation, attributed to biofilm growth prevention.

Integrating highly hydrophilic zwitterionic materials with biocidal agents such as Ag NPs is a promising strategy to alter the membrane surface chemistry in minimizing the attachment of both organics and bio-foulant bacteria. In this technique, the biocidal agent exerts toxicity toward the bacteria while the zwitterionic material shields the surface

from adsorption of organic foulants, overcoming technical challenges faced in treating complex feed solutions. Liu et al. [130] developed a new pathway for the fabrication of Ag NPs-loaded zwitterionic polymers for biofouling mitigation. After grafting zwitterionic poly(sulfobetaine methacrylate) (PSBMA) brushes to the membrane surface via an atom transfer radical polymerization (ATRP) reaction, Ag NPs were formed in situ through chemical reduction of silver. The specific PSBMA-Ag TFC structure in which Ag NPs were present on top of the polymer brush showed superior antibacterial activity (95%) and effectively inhibited the biofilm formation in dynamic biofouling experiments (Fig. 10).

Furthermore, the regeneration of Ag NPs after depletion was more feasible in this PSBMA-Ag TFC architecture. Liu et al. [266] systematically studied the performance of silica nanoparticles (Si NPs) and zwitterionic polymers in improving the antifouling properties of a TFC membrane. A dense layer of Si NPs was grafted via dip-coating with aminosilane-functionalized Si NPs (Si-TFC membrane), while zwitterionic polymer brushes were formed on the membrane surfaces through

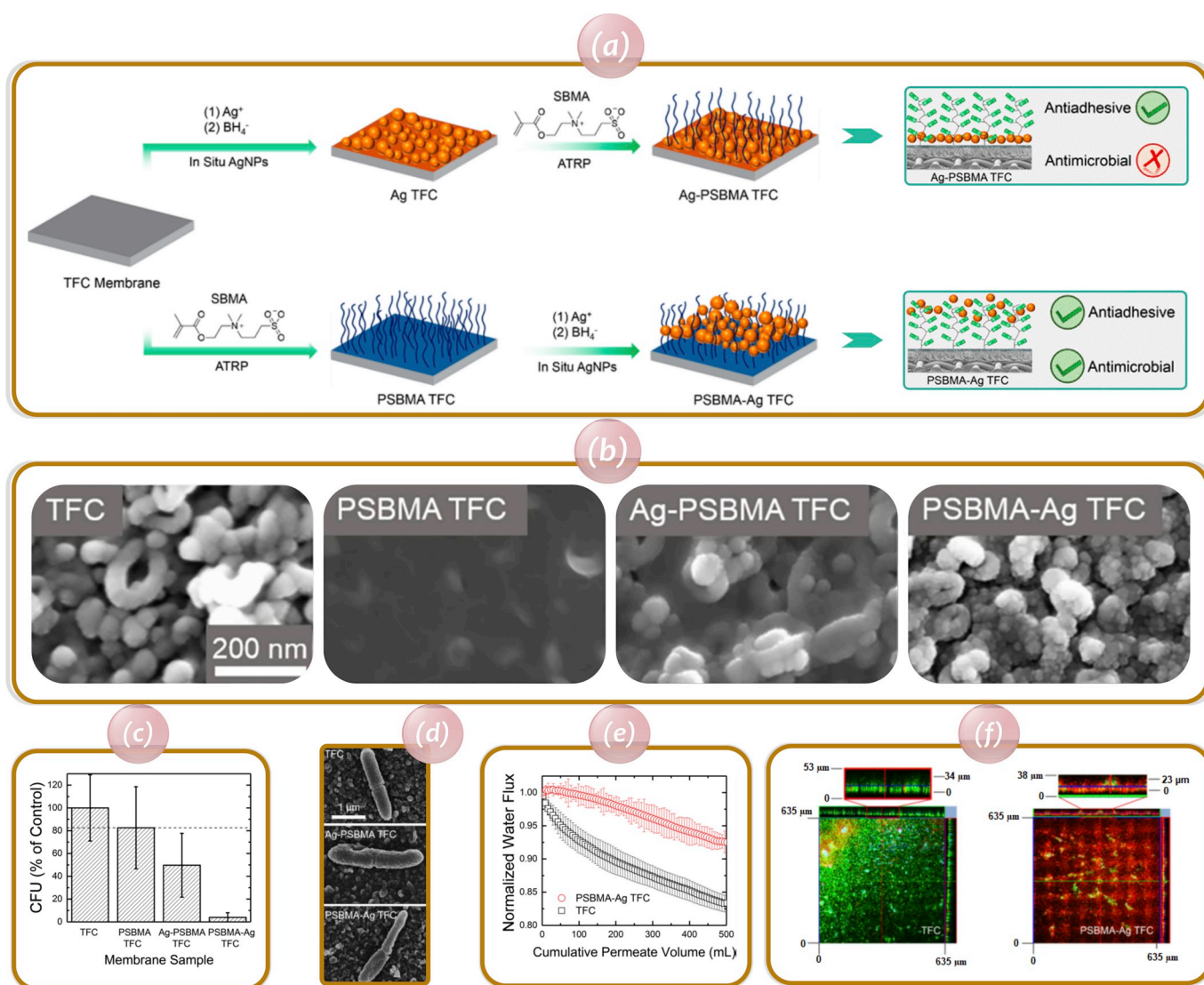


Fig. 10. (a) Schematic illustration of the two different approach for the membrane modification AgNPs formation on the polyamide layer followed by the PSBMA grafting and PSBMA grafting followed by in situ synthesis of AgNPs, (b) SEM images of the pristine TFC, PSBMA, Ag-PSBMA, and PSBMA-Ag TFC membranes, (c) the percentage of colony forming units (CFU) after 3 h exposure to *P. aeruginosa* bacteria, (d) the morphological SEM images of *P. aeruginosa* cells after exposure, (e) normalized permeate water flux during biofouling experiment by *P. aeruginosa*, (f) confocal laser scanning microscopy (CLSM) images of *P. aeruginosa* biofilm developed on pristine TFC and PSBMA-Ag TFC membranes after biofouling test, in which live cells, dead cells, and polysaccharide–EPS were stained with SYTO 9 (green), PI (red), and ConA (blue) dyes, respectively [130]. reproduced with permission from ACS Publications. (For interpretation of the references to color in this figure legend, the reader is referred to the Web version of this article.)

ATRP grafting (PSBMA-TFC membrane). The resulting membrane demonstrated considerably higher fouling resistance compared to that of the Si-TFC membrane against proteins and bacteria during dynamic fouling experiments (Fig. 11).

Qiu et al. [256] reported a strategy to simultaneously improve the water flux and the biofouling resistance via functionalization of Ag NPs

zwitterion nanocomposites on the TFC FO surface. They presented a simple and efficient approach to graft zwitterions via second interfacial polymerization and form Ag NPs by in situ reduction. On account of the simultaneous improvement of antiadhesive and antibacterial activity, the functionalized membrane exhibited significant biofouling resistance with regeneration ability. All the superior results offered by this simple

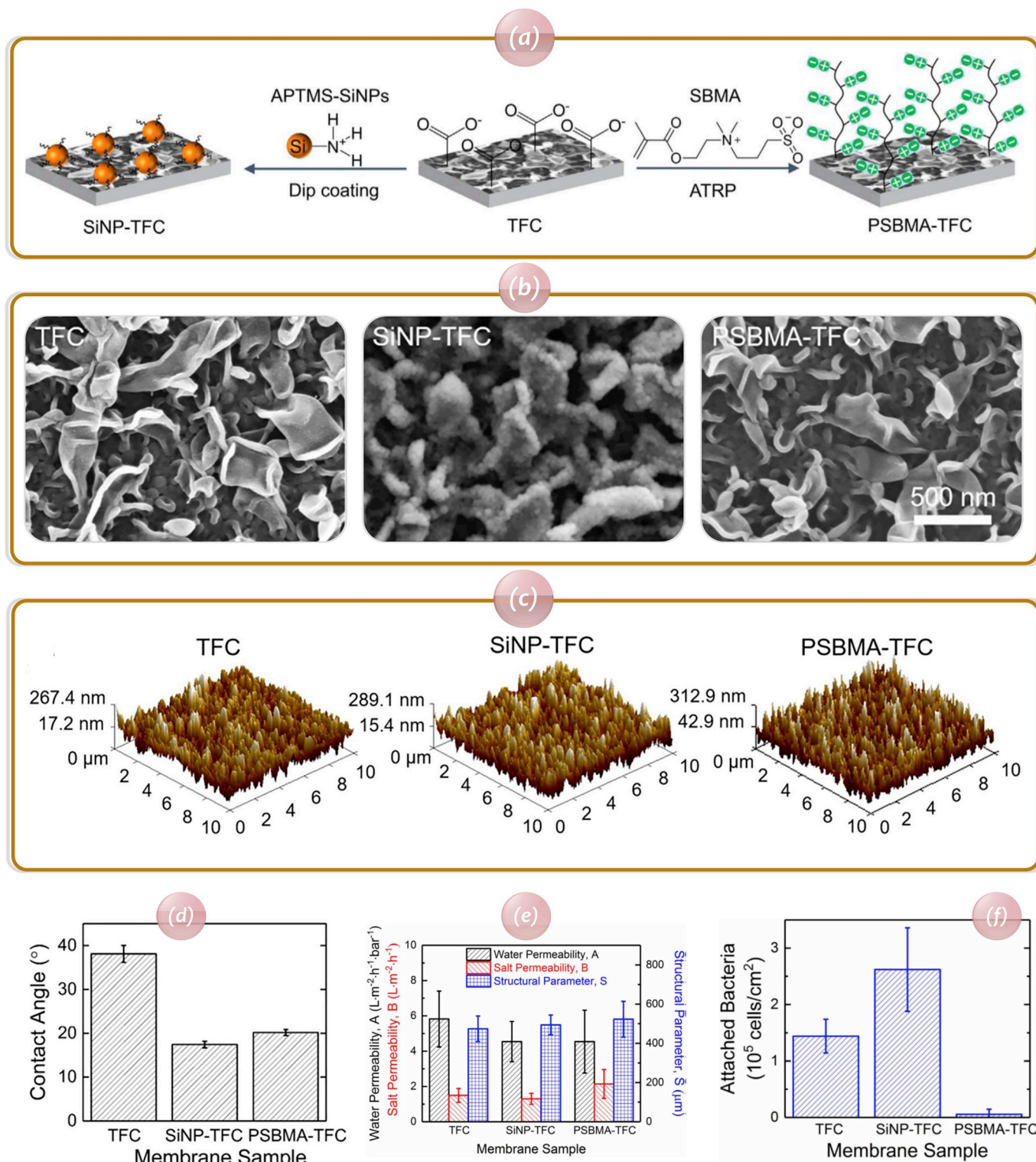


Fig. 11. (a) Schematic illustration of SiNP-modified and zwitterionic polymer PSBMA-modified TFC membranes, (b, c) SEM and 3D AFM images of the pristine TFC, TFC-PDA, and TFC-PSBMA membranes, (d) Water contact angles, (e) transport and structural properties of pristine TFC and TFC-PSBMA membranes, and (f) number of attached live *E. coli* on the surface of pristine and modified membranes after 3 h exposure to bacteria [266]. reproduced with permission from ACS Publications.

surface functionalization approach highlights its potential as an effective way for practical TFC FO applications. Fig. 12 and Table 4 summarize the materials utilized in biofouling mitigation and a comparison of different approaches applied in surface functionalization of FO membranes, respectively.

3.3. Surface coating for membrane biofouling mitigation

Surface coating is a simple technique for membrane surface modification [163]. The coating layer protects the membrane by diminishing the deposition of foulants onto the membrane surface; this technique can often be easily incorporated into existing membrane manufacturing processes [163]. In this method, the membrane surface properties such as roughness, hydrophilicity, and surface charge can be modified to reinforce the antibiofouling properties of the membranes [268]. The main restrictions of this approach are the typical decline in water flux and also issues of delamination of the coating during filtration operation [269]. Surface functionalization via chemical treatment approaches may be difficult and expensive in some cases. Surface coating is a quicker and more convenient physiochemical treatment.

Currently, there are two main surface coating approaches, thin-film coating and self-assembled monolayers for biofouling mitigation of FO membranes [269]. The main goal of these techniques is mostly to form a hydrophilic layer on the PA surface. Like grafting, surface coating involves two distinct procedures: coating-to and coating-from. The former involves post-modification of the membrane surface with hydrophilic polymers (PVA or PEG) or inorganic antibacterial nanomaterials

through dip-coating or spin-coating. In contrast, the latter consists of the formation of antibiofouling polymers or nanomaterials by in situ techniques on the membrane surface. Moreover, both usually include two steps: (i) dipping of the TFC PA surface into a coating solution, and (ii) residual solvent evaporation at ambient or moderate temperature to form the coating layer. Positively charged materials are potential candidates to facilitate stable surface coating with the negatively charged groups on the TFC PA FO membranes without any pre-treatment of the surface. On the other hand, in case of the coating-from technique, the membrane surface is normally pretreated to induce adsorption of the antibiofouling agent via pre-coating. Approaches including sol-gel [270] and bioinspired adhesion [271] have been applied to assist with the in-situ formation of antibacterial coatings.

The coating features, such as the type of solvent and the density of the coating solutions, have important effects on water permeation and salt rejection [272]. In other words, both polymer and solvent can affect water-polymer and polymer-polymer interactions and thereby the transport of water or solute molecules within the selective layer. For example, when using short-chain aliphatic alcohols as a coating layer, the water flux increases considerably due to the higher affinity toward water molecules. Also, solvents that are usually used for coating can swell and remove the residual monomers/additives on the membrane; this eventually results in facilitated water transport and higher permeability. However, a swollen membrane may lose its initial permeability after solvent evaporation by a decrement of water-polymer interactions or pore collapse through capillary forces. The degree of flux reduction after solvent evaporation depends on solvent properties, comprising

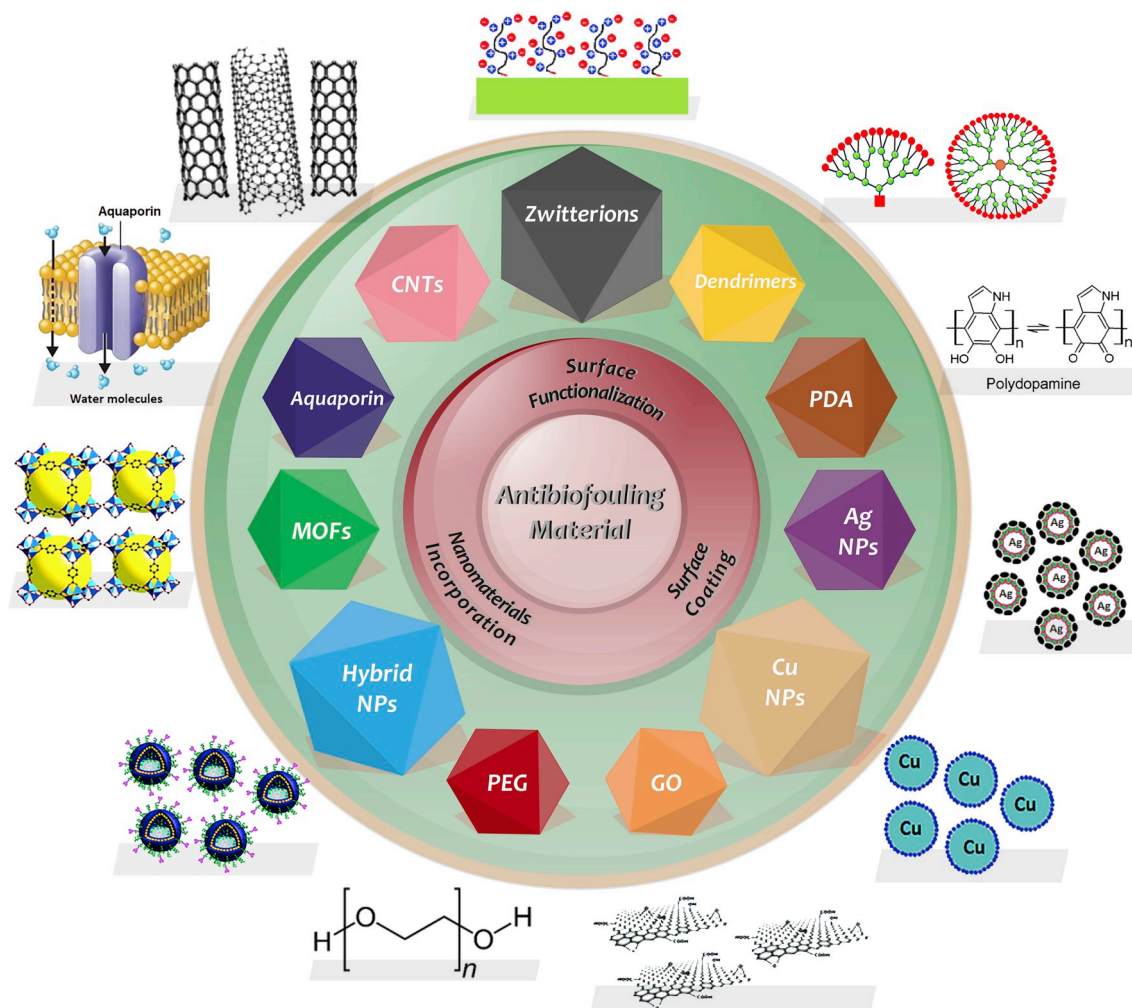


Fig. 12. Different types of materials applied in surface functionalization.

Table 4
A comparison of different approaches and antibacterial agents applied in surface functionalization of the FO membrane.

Year	Substrate Material	FS	DS	Antibiofoulant Constituent	Functionalization Approach	Characterization Method	Biofoulant	A key feature of Functionalization	Ref.
2016	TFC PA	1 M NaCl	DI	Cu NPs on PDA coating	In situ reduction	XPS, SEM, EDX, Ion release	<i>S. aureus</i>	- Simple, rapid, one-step procedure for the chelation of copper ions	[236]
2015	TFC PA	2 M NaCl	DI	Porphyrin	Grafting	DR-UV-vis, ATR-FTIR, AFM, SEM	<i>E. coli</i>	- Simple and efficient grafting integrating sunlight to FO process	[163]
2019	TFC PA	1 M NaCl	DI	PAMAM Dendrimer	Grafting	SEM, AFM, FTIR, XPS, ζ -potential	Domestic wastewater	- Superior ammonia selectivity	[212]
2015	PAN	2 M NaCl	DI	Ag NPs	In situ reduction	XPS, SEM, EDX, ICP-OES	<i>E. coli</i>	- High antifouling performance	[178]
2018	TFC PA	1 M NaCl	DI	BSA capped Ag NPs	Grafting	SEM, AFM, FTIR, XPS, ζ -potential, ICP-MS	<i>E. coli</i>	- Very high antimicrobial activity against <i>E. coli</i> for 14 days under laboratory conditions	[231]
2013	TFC PA	-	DI	GO	EDC/NHS coupling	SEM, Raman	<i>E. coli</i>	- The simple and versatile approach of direct grafting	[245]
2015	TFC PA	2 M NaCl	DI	GO	EDC/PLI grafting	FTIR, ζ -potential, TGA, AFM, TEM	ATP	- Less influence on surface roughness and charge	[251]
2016	TFC PA	0.2–1.2 M NaCl	DI	GO	EDC/NHS coupling	SEM, AFM, Raman, ζ -potential	SW ^a supplemented <i>P. aeruginosa</i>	- Slight improvement in water permeability and salt rejection	[156]
2016	CTA	1 M NaCl	SW	PEG-grafted PDA	Grafting	SEM, ATR-FTIR, AFM	Activated sludge in MBR	- Irreversible binding to the PA surface	[217]
2018	TFC PA	-	DI	Ag NPs on PDA coating	In situ reduction	SEM, ζ -potential, EDX, XPS	SW supplemented <i>P. aeruginosa</i>	- Strong antimicrobial activity	[265]
2015	TFC PA	1 M NaCl	DI	Ag-GO nanocomposite	Click chemistry reaction	SEM, AFM, Raman, ζ -potential, XPS, ATR-FTIR, ICP-MS	<i>E. coli</i>	- easy and stable grafting	[262]
2015	TFC PA	1 M NaCl	DI	Ag-GO nanocomposite	EDC/NHS coupling	SEM, AFM, Raman, ICP-MS, XPS	<i>E. coli</i> , <i>E. faecalis</i>	- enhanced hydrophilicity and smoothness	[263]
2017	TFC PA	NaCl	DI	Ag-GO nanocomposite	EDC/NHS coupling	SEM, AFM, Raman, EDX	<i>P. aeruginosa</i>	- significant salt rejection	[264]

(continued on next page)

Table 4 (continued)

Year	Substrate Material	FS	DS	Antibiofoulant Constituent	Functionalization Approach	Characterization Method	Biofoulant	A key feature of Functionalization	Ref.
2019	TFC PA	NaCl	DI	Ag-MOFs	In situ growth	SEM, AFM, XPS, ζ -potential, ATR-FTIR, ICP-OES	SW supplemented <i>P. aeruginosa</i>	<ul style="list-style-type: none"> - A slight reduction in surface roughness - Facile technique with short reaction time - A two-step deposition procedure of Ag-MOFs without stabilizing/reducing agent - Uniform distribution of Ag-MOFs on the PA layer 	<ul style="list-style-type: none"> 30% water flux decline during dynamic biofouling - No adverse effect on membrane transport properties - Slight reduction in both the salt and water permeability - Nearly 100% antibacterial activity
2017	TFC PA	NaCl	DI	Ag NPs Zwitterionic nanocomposite	ATRP grafting	SEM, AFM	SW supplemented <i>P. aeruginosa</i>	<ul style="list-style-type: none"> - Smoother membrane surface - Remarkable increased hydrophilicity (reduced contact angle from 74° to 21°) - 95% antibacterial activity 	<ul style="list-style-type: none"> - Irreversible binding of Ag-MOFs to the TFC surface - 46% increase in dead cells biovolume - 60% decrease in EPS content
2017	TFC PA	NaCl	DI	Silica NPs Zwitterionic nanocomposite	ATRP grafting	SEM, AFM, ζ -potential	<i>E. coli</i>	<ul style="list-style-type: none"> - High surface hydrophilicity - Reduced surface roughness - Minimized electrostatic attraction of organic foulants - Effective shielding of the carboxylic groups on the TFC surface 	<ul style="list-style-type: none"> 8% water flux decline - 48% reduction in the live cells biovolume - PSBMA-TFC membrane presented 68% decreased fluorescence intensity - Improved antifouling property - Reduced water flux decline (17%)
2018	TFC PA	1 M NaCl	DI	Ag NPs Zwitterionic nanocomposite	Grafting In situ reduction	XPS, ATR-FTIR, SEM, AFM, EDX, ICP-OES	<i>E. coli</i>	<ul style="list-style-type: none"> - Simple and efficient surface modification strategy - Increased structure hydrophilicity without adverse effect on transport properties - High water flux and excellent selectivity 	<ul style="list-style-type: none"> PSBMA-TFC membrane showed a drastic increase in antibiofouling resistance (96% reduction of the number of attached <i>E. coli</i>) - 96% antibacterial activity - Significant biofouling resistance
2016	TFC PA	1 M NaCl	DI	Ag NPs on PDA coating	In situ growth	XPS, AFM, SEM, EDX	<i>E. coli</i> , <i>S. aureus</i>	<ul style="list-style-type: none"> - Enhanced hydrophilicity (contact angle from 68.4° to 40.6°) - Increased roughness 	<ul style="list-style-type: none"> - Simultaneous improvement of antiadhesive property - PDA coating allows long-term Ag NPs release and multiple times regeneration - strong, sustainable antibacterial properties (<i>E. coli</i> 95.6%)
2014	CTA	1 M NaCl	GW ^b	Ag NPs regenerated by TiO ₂	In situ growth	XPS, AFM, SEM, EDX, ζ -potential	ATP	<ul style="list-style-type: none"> - The Ag NPs formation decreased the water flux (8.58%) and the reverse salt flux (20.71%) - Moderately enhanced hydrophilicity - Increased roughness - 11 times less bacterial growth 	<ul style="list-style-type: none"> - TiO₂ regenerates the Ag NPs by decomposing the organic matter - 67–72% initial water flux recovery - effective inhibition of bacteria growth

^a Synthetic Wastewater.^b Ground Water.

surface tension, polarity, water miscibility, and hydrogen bonds [272]. Moreover, it has been observed that long exposure time with the solvent during coating may deteriorate the integrity of the PA selective layer and decrease salt rejection. Coating methods applied to alleviate biofouling of FO membranes can be divided into three methods: bioinspired PDA coating, surface adsorption, and layer-by-layer (LBL) assembly.

3.3.1. Bioinspired coating

Zwitterionic materials may be used for surface coating rather than surface chemical functionalization [273,274]. In a study by Nguyen et al. a commercial FO membrane surface was modified by poly amino acid 3-(3,4-Dihydroxyphenyl)-L-alanine (L-DOPA) coating, a zwitterionic (redox functional amino acid) and inspired by the adhesive proteins found in marine mussels (Fig. 13) [275]. As they reported, L-DOPA can self-polymerize in aqueous solutions and can attach firmly to the various substrates [213]. They theorized that the coated membrane became intensely hydrated and then the accumulation and adsorption of foulants on the membrane surface were prevented by the hydrated layer. Besides, they observed that by increasing the coating time, the water contact angle was reduced from the initial value of 48° for an untreated sample to 44° and 38° for membranes coated in a period of 4 h and 12 h, respectively. The L-DOPA coated membranes revealed higher negative Zeta potential than that of uncoated membranes. While L-DOPA has the positively charged NH_3^+ groups and the negatively charged carboxylic acid (COO^-) groups, coating of L-DOPA imparted a negative charge on modified membrane surfaces as the COO^- contributed more to the overall charges.

In other studies, layered double hydroxide (LDH) nanoparticles were deposited on the TFC membranes by a polydopamine (PDA)-induced coating process [257]. Layered double hydroxides (LDHs) are ionic lamellar compounds fabricated by positively charged Brucite-like layers with an interlayer region comprising charge compensating anions and water molecules [276,277]. The TFC membrane prepared via an interfacial polymerization was dip-coated into the LDHs suspension for 1 h. Another type of modified TFC PA membrane was obtained by immersing the TFC membrane in the suspension of LDHs until desiccation at 30 °C. Only a small quantity of LDH nanoparticles appeared on membrane surfaces following the dip-coating method while a more consistent LDH nanoparticle layer was prepared successfully on the membrane surface following desiccation. The latter surface showed the least coverage of *E. coli*. The enhanced antibiofouling performance of the modified membranes was attributed to the considerable increase of surface hydrophilicity due to the surface coating of LDHs nanoparticles, which was

consistent with the water contact angles [278].

3.3.2. Surface adsorption

Physical adsorption is a convenient approach for membrane surface modification, and some researchers have adopted this technique to modify the surface properties of FO membranes. In this approach, the hydrophobic portion of the antibiofouling agent has favorable free energy of attraction to the TFC PA surface, altering the membrane surface morphology, physiochemical properties, and performance. Generally, these types of coatings are relatively simple to use in commercial applications. Ahn et al. have fabricated polyvinyl alcohol (PVA)-coated cellulose acetate (CA)-based flat-sheet membranes and applied them in the forward osmosis (FO) process [279]. The CA-based membranes were prepared via conventional immersion precipitation and then coated by PVA. The membranes were then immersed in the 0.01% glutaraldehyde solution for a cross-linking process. It was reported that the hydrophilicity of CA membrane surface increased and hence water contact angle was diminished (from 73.5° to 39.3°). In another study conducted by Zhang, surface modification of the TFC PA membranes was carried out by different functionalities [279]. For instance, the TFC-based membrane was in contact with a 2.0 wt% poly (amido amine) (PAMAM) aqueous solution for 5 min, imparting amine groups on the surface of the membrane. It was revealed that the nitrogen content increased from 8.93% to 11.27% after dendrimer attachment as evidence for the existence of the PAMAM layer on the modified membrane. After this step, the modified membrane was coated by 5% (w/v) poly(ethylene glycol) methyl ether acrylate (PEGMEA) solution in 80/20 (v/v) water/isopropanol (IPA) for 24 h to obtain the final TFC membrane. The water contact angle of the pristine TFC membrane was around 68°, while all the modified membranes demonstrated water contact angles of less than 50°. Experiments of bacteria attachment on membrane surfaces showed significant improvements in antibacterial properties after modification. The live bacteria attachment on pristine TFC membranes was 6.5% per membrane area, while it was only 0.16% and 0.05% for TFC membranes modified by PAMAM and PEGMEA, respectively. The results in comparing two modified membranes showed that, although both hydrophilicity and electrostatic parameters were contributing to bacterial adhesion, hydrophilicity seemed to play a more significant role in the process of live cell attachment. The hydrophilic PAMAM-modified membrane was much less prone to live *E. coli* adhesion even though positively charged. Fig. 14 illustrates the evaluation of the live and dead *E. coli* attachment on membrane surfaces along with statistical comparisons.

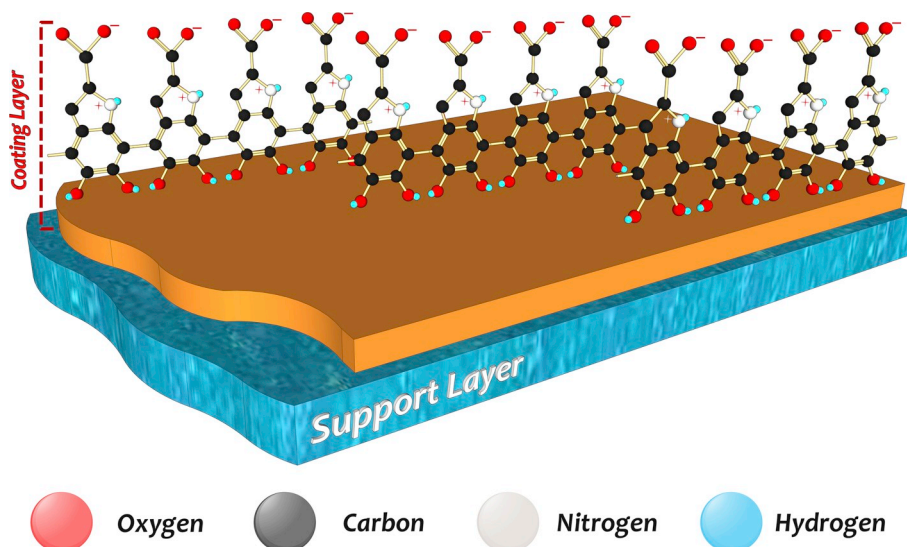


Fig. 13. Schematic diagram of the structure of zwitterionic poly (L-DOPA) attached on the FO membrane.

In a similar study by Li et al. [217], the biofouling behavior of CTA FO membrane with surface modification via PDA coating in a submerged osmotic OMBR was investigated. The hydrophilicity of the membranes improved considerably whereby the contact angle of CTA membranes decreased from 91.4° to 79.8° after PDA coating modification. It was also reported that the amount of β -polysaccharides adsorbed onto the surface of the modified membrane was considerably less than that on the pristine membrane, indicating antiadhesion properties of the PDA-modified membrane for polysaccharides. This is consistent with the findings of McCloskey et al., who reported that the presence of a PDA-coated layer on the membrane surface diminished the protein adhesion [281]. Improvement of antibacterial properties is possibly the main reason for improving the antibiofouling of the PDA-modified membrane [281]. The protonated amine groups in the PDA layer were also responsible of bacteria inactivation by contacting the cells [282]. Therefore, this modification provided ideal antibacterial properties and decreased the adhesion of biopolymers, leading to a lower resistance in OMBR operation.

3.3.3. Layer-by-layer assembly

LBL offers an efficient and facile technique for fabrication of ultra-thin films and has been developed for different membrane separation processes, such as nanofiltration [283], reverse osmosis [284], and forward osmosis [285]. It is performed by alternate electrostatic-based deposition of oppositely charged polyelectrolytes on a porous substrate [286,287]. Compared to the other modification approaches, one of the main advantages of LBL coating is that it provides the potential to accurately control the thickness of the coating layer to fall in the nanoscale range. Consequently, the water flux reduction resulting from the formation of extra mass transfer resistance can be potentially minimized. Moreover, the fouling tendency can be diminished by control over the selection of the outermost polyelectrolyte by exploiting electrostatic repulsions with the foulants. Additionally, these multilayer coatings can be prepared by cost-effective, water soluble, and commercially available polymers. Despite relatively simple fabrication procedures, there are a few possible variations in the implementation of LBL membranes. The main controlling parameters affecting the final properties of the membrane structure comprise polymer compositions and concentrations, number of layers, and fabrication conditions/methods.

Natural electrolytes can also be used in LBL assembly. Phytic acid (PhA) is a nontoxic, biocompatible electrolyte with phosphate acid groups attached to a cyclohexanhexol ring. This compound exhibits high affinity toward water molecules and strong chelation capacity with

mono/divalent metal ions. Accordingly, superhydrophilic complexes of PhA-metals can be prepared by spontaneous chelation of PhA and metal ions. Xiong et al. [288] conducted LBL assembly of PhA and antibacterial metal ions ($M = Ag, Cu$) for surface modification of PhA-TFC membrane to simultaneously enhance the bio/organic fouling resistance and separation performance of the final membrane. Since the PA molecule carries six phosphonic acid functional groups with chelation ability, it strongly reacts with the metal ions as an inorganic cross-linker and induces the LBL assembly of the PA- Ag, and PhA- Cu complexes on the TFC PA surface (Fig. 15). The FO performance of both the modified TFC-PhA-Ag and TFC-PhA-Cu TFC membranes was systematically regulated by manipulating the PhA concentration and assembly cycles. Due to increased hydrophilicity, the water flux of the modified TFC-PhA-Ag and TFC-PhA-Cu TFC membranes was enhanced up to 57% and 68%, respectively, without sacrificing the membrane selectivity. Additionally, both modified membranes demonstrated excellent stability against metal release owing to the strong attachment on the surface. The released concentrations of Ag and Cu ions in the water product were reported in the ranges of 0.001–0.013 and 0.001–0.010 mg/L, respectively. This result indicates the safety of both the TFC-PhA-Ag and TFC-PhA-Cu membranes for water treatment applications, as the released content of both ions were very far below those suggested by WHO guidelines (0.1 and 2 mg/L for Ag and Cu ions, respectively).

Liu et al. fabricated silver nanocomposite LBL-Ag FO membranes using the LBL assembly method for membrane biofouling control [289]. To prepare nanocomposite membranes, they controlled the amount of Ag NPs incorporated into the polyanion poly(sodium 4-styrene-sulfonate) (PSS) solution, and polycation poly(allylamine hydrochloride) (PAH) solution, following by crosslinking with 0.1 wt% glutaraldehyde solution at room temperature. The membrane surface hydrophilicity was enhanced, which could considerably reduce bacterial adhesion and hence alleviate membrane biofouling [290]. The antibacterial activities of modified membranes against *B. subtilis* and *E. coli* were tested using the CFU method, and the results showed that the number of live bacteria decreased after 24 h exposure to the nanocomposite membranes. It was also observed that nanocomposite membranes showed slightly better bactericidal activities against *B. subtilis* compared to *E. coli*, owing to the existence of a lipopolysaccharide (LPS) layer outside the latter Gram-negative bacterial cell membrane. The presence of this LPS layer around the cell wall of Gram-negative bacteria makes them more resistant to Ag toxicity [291,292]. Additionally, the results of bacterial diffusion tests were consistent with the results of the CFU test.

Salehi et al. used LBL assembly of positive chitosan (CS) and negative GO nanosheets on a support layer which was prepared by blending

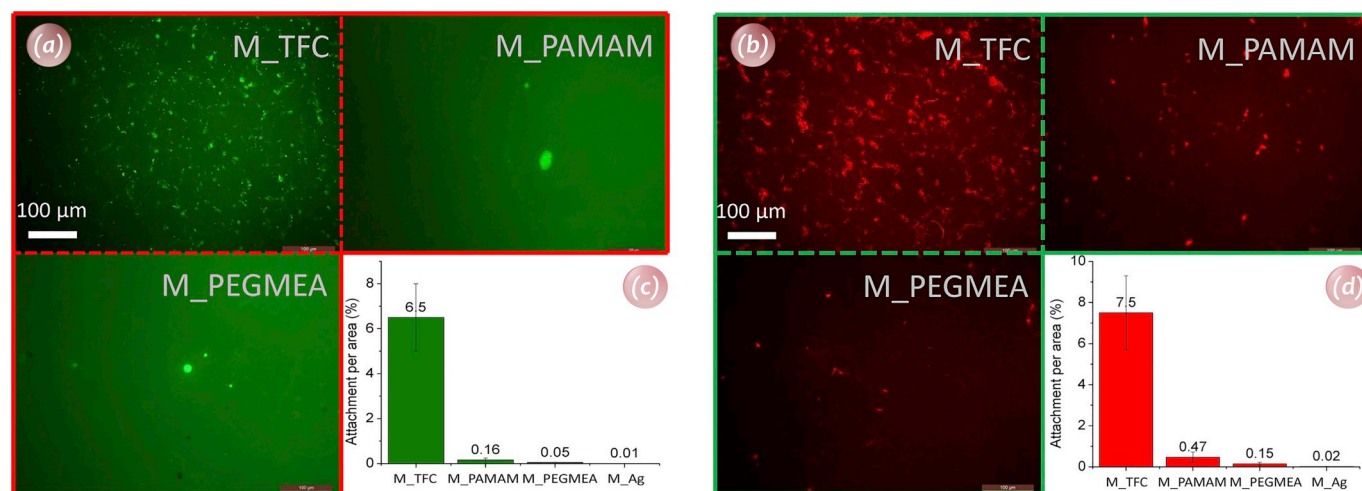


Fig. 14. Representative fluorescent microscope graphs of the (a) live and (b) dead *E. coli* attachment on the surfaces of the TFC, PAMAM, PEGMEA membranes, and statistical comparisons of the (c) live and (d) dead cell attachment per area of the membranes [280]. reproduced with permission from Elsevier.

hydrophilic sulfonated polyethersulfone (SPES) with traditional PES [293–295]. Water contact angle values obtained from the surface of the membranes revealed that this process effectively improved the hydrophilicity of the membrane surface. This result was mostly attributed to the existence of hydrophilic oxygen-containing functional groups on the surface of LBL modified membrane.

In another study, a GO-OCNTs-LBL membrane was fabricated by GO and oxidized carbon nanotubes (OCNTs) with five bilayers on a PES support. It was reported that the contact angle of GO-OCNTs-LBL membrane was lower than that of GO-LBL membrane due to the presence of hydroxyl groups. Due to the presence of higher content of oxygen-containing groups in the GO-OCNTs-LBL membrane (9.5%, detected by XPS) compared to the GO-LBL membrane (7.6%), hydrophilicity increased considerably. Although covalent attachment and LBL self-assembly are two successful techniques in the use of GO nanosheets to improve the properties of membrane separation processes, industrial application is less appealing due to the complexity of multiple steps [247]. In this context, Hegab et al. [296] introduced an industrially favorable methodology for immobilizing GO nanosheets onto the surface of TFC FO membranes using PDA bio-adhesion as a single pot technique of surface modification. They used PDA as an adhesive agent to bind the GO nanosheets onto the surface of TFC FO membranes. The membrane presented a smoother surface with improved antibiofouling performance. Although all membranes demonstrated a decline in the water flux because of the growth of bacterial cells, the rate of flux decline for GO-PDA-modified membrane was the least. Additionally, ATP bioluminescence assay revealed the antibacterial activity of the GO-PDA-modified membrane compared to that of other membranes, attributed to the effect of the smoother, more compact, and hydrophilic surface [262]. Furthermore, the active sites of defective edges on the GO nanosheets generated ROS [297] that caused stress, leading to the destruction of the bacterial cell membrane [298]. Additionally, on the direct contact of bacterial cells with the surface of modified membranes, cell lysis occurred and this phenomenon suppressed biofilm establishment, resulting in considerable reduction of biofouling and its associated complications [299].

LBL membranes are sensitive to operating conditions. In the presence of chlorine, at high ionic strength conditions, and at very low or high pH, the LBL layer may swell or even desorb. To improve stability, a number of efforts applied crosslinking steps to stabilize the layers, as also mentioned in several occasions in the discussion above [300]. However, the application of such LBL FO membranes is still restricted only to the use of low concentration draw solution and neutral draw solutes for long-term operation [301].

Hegab et al. adopted a surface modification approach of forward osmosis membranes by the single-step assembly of multifunctional poly tannic acid-graphene oxide coating to reduce biofouling [302]. In this study, they used a versatile platform to immobilize GO nanosheets on the surface of TFC PA FO membranes resulting in smoother and higher hydrophilicity. This hydrophilic layer provided a barrier with the aim of short-term inhibition of foulant adhesion on the membrane surface and hence 99% antifouling resistance (33% increase compared to the pristine membrane) [303]. Additionally, the sharp edges of GO nanosheets imparted biocidal activity [304,305] and acted as reactive sites for the production of ROS [297,298].

Despite its simplicity for antibiofouling application, surface coating suffers from some challenges. Although it has been effectively applied for FO membrane surface modification, the interaction of the coating layer with a PA layer sublayer is of relatively weak noncovalent type, such as van der Waals attraction, electrostatic interaction, or hydrogen bonding; this increases the possibility of delamination or detachment during continuous operation or cleaning processes. In other words, this susceptibility to damage offers short-term resistance to biofouling, and long-term biofouling formation may still occur. Additionally, in this process an additional thin layer forms on the membrane surface with typical thickness in the range of hundreds of nanometers. The presence

of this layer may provide an extra barrier for water molecule permeation and a decrease of the accessible sites for water flow [306]. The coverage may lead to a certain degree (nearly 10% or greater) of water permeability reduction, even in case of initially dense TFC PA FO membranes. As a result, the coating layer should have an intrinsically high water permeability and be satisfactorily thin to obtain the maximum water flux for practical applications.

3.4. Nanomaterials incorporation to achieve biofouling mitigation

The incorporation of nanomaterials into the PA layer during the IP process is another approach for further modification of the final FO membrane properties [307]. The key advantage of TFC membranes is the flexibility of tuning the chemistry of the PA layer and that of the porous support independently to adjust the membrane performance [308]. For optimizing the selective layer properties, thin film nanocomposite (TFN) membranes have been prepared by the incorporation of different types of nanomaterials within the PA layer, including zeolites, silica particles, CNTs, and metal oxides [309–311].

Despite considerable improvement in membrane performance and fouling resistance by incorporating hydrophilic nanoparticles, the main bottleneck is still the alleviation of biofouling implications [312]. Compared to surface-coated or functionalized TFC membranes, antibacterial TFN PA membranes have the advantages of ease of manufacture and fewer preparation steps, and generally lower cost. In this context, different types of antibacterial nanomaterials such as Ag nanoparticles and Ag-based nanocomposites [313], Cu nanoparticles [314,315], titanium dioxides (TiO₂) [316,317], and zinc oxides (ZnO) [318] have been applied to prepare antimicrobial TFN membranes. The physiochemical aspects of the direct incorporation of nanomaterials into the selective layer can be summarized as follows:

- The weak interaction of the filler with the membrane matrix may result in an easy release, and thus a decrease in the antibacterial activity over time [319].
- The low compatibility between the inorganic nanomaterials and polyamide chain adversely affects the membrane performance and especially selectivity [319,320]. In contrast, the higher affinity of the nanomaterials to the PA matrix increases the compatibility and connection between the two phases via hydrogen or covalent bonds, and Van der Waals forces reduce or ideally prevent nanoparticle depletion during membrane operation.
- Another drawback of the incorporation of nanoparticles into the polymeric matrix of the PA layer is related to size. Since the thickness of the PA layer is in the range of 100–300 nm, nanomaterials of larger size deteriorate the membrane surface integrity and adversely affect salt rejection. On the other hand, if appropriately incorporated, these nanoparticles can have a substantial positive impact on mechanical strength and solute selectivity. However, it has been observed that nanoparticles of small size (less than 100 nm in diameter) have a great tendency to agglomeration due to surface interactions [321, 322].
- The incorporation of nanomaterials may alter the penetration rate of MPD/TMC monomers during interfacial polymerization, and possibly a much thinner polyamide layer forms. The functional groups on the surface of nanomaterials may react with monomers during the IP process, and this interaction affects the rate of the polycondensation reaction between amine and acyl chloride monomers [323,324]. Although this behavior may decrease the cross-linking degree of the final selective layer, it protects the nanoparticle release rate and provides relatively higher long term stability.
- The performance of the nanocomposite membranes is closely related to the morphology of the nanoparticles [325]. It is generally accepted that nanomaterials with spherical morphology have better compatibility with the polymeric matrix, resulting in a less destructive effect of the mechanical and transport properties. Although

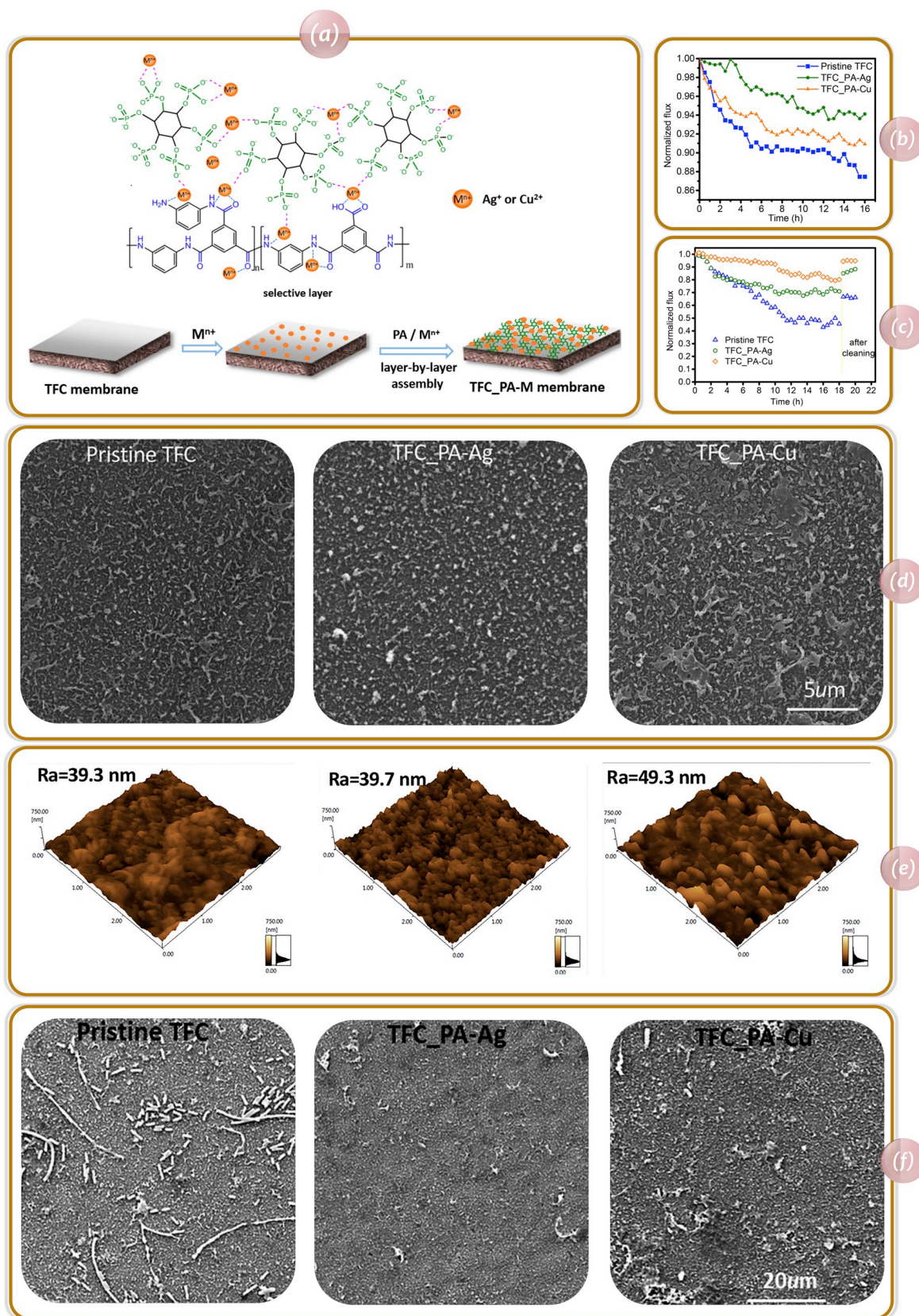


Fig. 15. (a) Schematic diagram of TFC-PA-Ag and TFC-PA-Cu membranes fabrication procedure, water flux reduction during dynamic b) biofouling, (c) organic fouling tests with *E. coli* and sodium alginate as model foulants (d, e) SEM and AFM surface morphologies of pristine TFC and surface-modified membranes, and (f) SEM micrographs of the pristine TFC and surface-modified membranes (4 cycles of LBL assemblies) after incubation in *E. coli* suspension [288]. reproduced with permission from ACS Publications.

numerous studies were carried out to elucidate the effects of nanoparticles with different sizes on the properties of the nanocomposite membranes [326,327], further investigation is still required to assess the effect of the nanoparticle size on the physicochemical properties of the nanocomposite membranes.

- The presence of nanomaterials may increase the gaps between chains; therefore, the fractional volume usually increases and hence additional nanochannels are created for rapid transport of water molecules, which should result in increased water flux [328]. Additionally, the presence of nanomaterials may lengthen the permeation path of solute molecules by unique zigzag passage [329]. However, the typical trade-off between permeance and selectivity is usually observed also for TFN membranes.
- Although the nanomaterials offer numerous advantages even at low concentration, the uniform dispersion of the nanoparticles in the aqueous or organic phase during IP is a limiting factor [330]. The incorporation of small-size biocidal nanoparticles has led to the fabrication of membranes with improved biofouling resistance, smoother surface, enhanced hydrophilicity, increased water flux as well as higher antibacterial activity [331]. However, it has been suggested that excessive loading of nanoparticles may lead to severe aggregation, a considerable reduction in the effective surface area of the nanoparticles, defect formation, and a decrease in the mechanical strength of the resultant membrane. Consequently, exploring the optimum loading concentration of nanoparticles is of critical importance to attain a uniform dispersion of nanoparticles [332, 333].

To address these challenges, some alternative materials have been proposed, such as MOFs [334,335]. Compared to the fully inorganic compounds, MOFs offer a better affinity to the organic nature of the PA chain because of the presence of the organic linker in their structures. The proper compatibility between the two phases can be attributed to the higher interaction with the PA matrix at the MOF-polymer interface in terms of covalent, non-covalent, and hydrogen bonds, also preventing the formation of non-selective voids between the phases [336–339]. Moreover, MOFs are frequently reported in the literature for their potential antibacterial activity [340,341] as they serve as a reservoir of metal ions for the controlled release of bactericidal agents [342,343]. Another advantage of MOFs is the uniform distribution of active metal sites in their structures, providing a long-lasting bactericidal activity without agglomeration or oxidation [344,345]. These outstanding features provide a prospect for improving the selective layer characteristics without deteriorating its performance and especially selectivity.

Although MOFs were widely reported in the fabrication of TFN membranes by incorporating ZIF-11 [346], ZIF-8 [347], MIL-68 [346], MIL-53 [160], or MIL-101 [346], few studies considered the incorporation of MOFs for biofouling mitigation of FO membranes. Besides, almost all the TFN FO membranes were prepared via dispersion of MOFs, whether in TMC organic or MPD aqueous solutions. Based on the literature survey, the influence of MOFs incorporation to the TFN membranes can be generally divided to five domains: (i) the relation between the membrane transport performance and structure change, (ii) MOF type and concentration during the IP process, (iii) catalytic effect on the IP process, (iv) MOFs-polyamide chain interaction at interface, and (v) the hydrolysis degree of the acid chlorides.

To simultaneously improve the desalination performance and antibacterial activity of the TFC FO membrane without detrimental effects on the selectivity, Zirehpour et al. [348], incorporated Ag-BTC MOF nanocrystals into the PA layer during IP reaction. Although low content of Ag-MOF (0.02 wt%) was incorporated, the water permeance of TFN membrane increased of about 55% following modification. Meanwhile, the TFN membrane presented bacterial inactivation of 96% and 90% for *E. coli* and *S. aureus*, respectively. This antibacterial activity was attributed to the minimal aggregation, improved compatibility, and uniform distribution of the MOF crystals in the PA layer, while a low

release rate of the Ag ions (0.04 $\mu\text{g}/\text{cm}^2\cdot\text{day}$) was observed after 24 days. All these features resulted in 20–24% and 6–10% water flux decline for TFC and for TFN membrane during biofouling experiment, respectively [349]. In a similar study by Firouzjahi et al. [184], the synergetic biocidal activity of graphene-oxide and silver metal-organic framework (GO-Ag-MOF) was investigated for TFN FO membranes. GO, Ag-MOF, and GO-Ag-MOF were incorporated into the PA layer during IP process (Fig. 16). The GO-Ag-MOF TFN membrane showed about 95% antibacterial activity against *E. coli*, while the bacteria inactivation of Ag-MOF TFN, GO TFN, and TFC FO membranes were reported to be 80%, 66%, and 2%, respectively. After a 24-h biofouling experiment, no significant water flux drop was reported for the GO-Ag-MOF TFN membrane, while 38%, 54%, and 70% water flux reduction was observed for Ag-MOF TFN, GO TFN, and TFC membranes, respectively. In a different study conducted by Rahimpour et al. [185], two different concentrations (0.01 and 0.02 wt%) of polyrhodanine (PRh) NPs were incorporated into the PA layer and provided a more hydrophilic layer with lower thickness compared to the pristine membrane. The modification translated into a 45% water permeance increment, and into a 40% decline in salt permeability. High antifouling/antibio-fouling properties and bacterial inactivation (89% for *E. coli* and 92% for *S. aureus*) were observed for the PRh-TFN membrane.

In other studies, He et al. [350], added different amounts of graphene-oxide in the MPD solution during IP process for the PA layer formation. Antibacterial activity of about 22%, and viability of 87 and 60% were observed for pure GO, TFC, and GO-TFC PA membranes, respectively. Physical damage to the cell membranes through increased cellular oxidative stress was reported as the main mechanism for the antibacterial activity of GO nanosheets. Wang et al. [351] fabricated free-standing cellulose triacetate (F-CTA)-graphene oxide (GO) FO membrane by addition of GO nanosheets to the CTA/methanol/lactic acid solution. Ascribed to the hydrophilic nature of GO nanosheets, the biofilm thickness of *E. coli* reduced from 18.5 μm to 4 μm as a consequence of GO incorporation. In another study, different concentrations of graphene quantum dots (GQDs) were added to the MPD solution during IP process and the results showed a contact angle reduction from 72.9° to 51°, for the neat TFC and TFN membranes, respectively. In addition, bacterial inactivation of 90% and 95%, against *E. coli* and *S. aureus*, was reported after 1-h incubation, attributed to the uniform distribution of GQDs within PA layer (Fig. 17a–d). The SEM images of the membrane after incubation with bacteria showed physically disrupted bacteria cells on the TFN membrane, while such behavior was not observed for neat TFC surfaces (Fig. 17e and f). The significant antibacterial activity of GQDs modified TFC membranes was also confirmed by fluorescent images (Fig. 17g and h) [351].

4. Future perspective

Despite increasing attention to the FO process during the last decade, an ideal and broadly applicable membrane has yet to emerge. TFC polyamide FO membranes have numerous advantages such as high salt rejection and water permeability, as well as safe operation in a wide range of temperature and pH, but they also have two main drawbacks: the susceptibility to fouling and the sensitivity to chlorine disinfectants and other oxidants. Membrane biofouling is specifically one of the most significant bottlenecks of the FO technology. The increase in energy needs and hence operational costs, increased chemical cleaning frequency, and shorter membrane lifespan are some of its adverse consequences. Biofouling cannot be completely eradicated, but membrane modification approaches have been established that minimize or delay it. The fabrication of antibiofouling membranes is an important research topic in FO technology and has attracted wide attention in recent years. The strategies reviewed here are related to the post-fabrication surface modification of TFC PA FO membranes, and to the incorporation of nanoparticles or nanostructures during interfacial polymerization

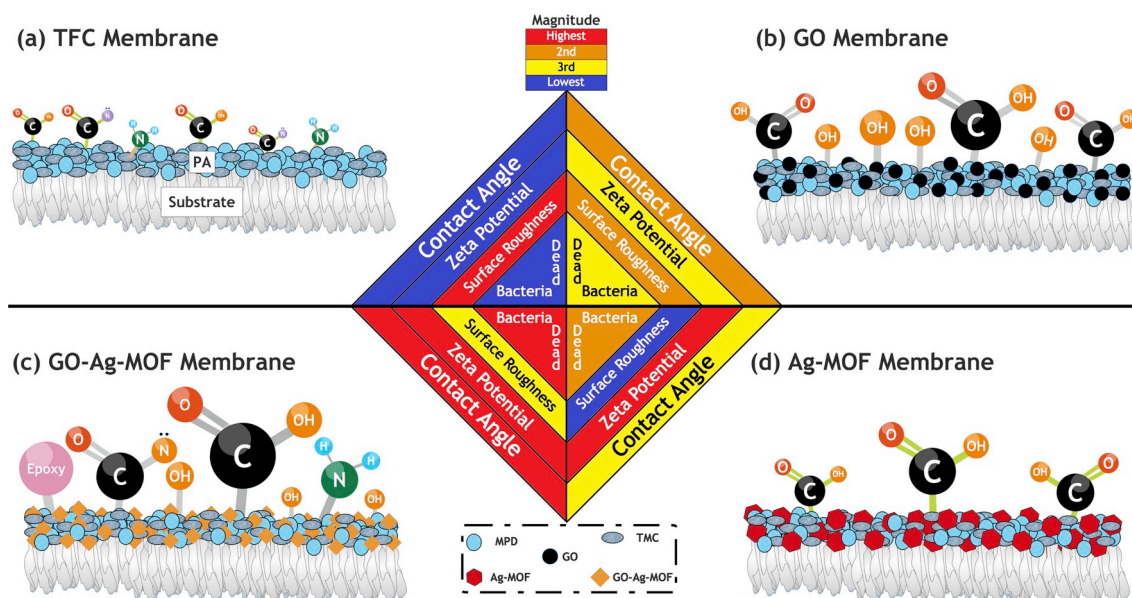


Fig. 16. Different structural membrane's parameters effecting biofouling of incorporated TFC FO membrane [184]. reproduced with permission from ACS Publications.

processes.

It is generally accepted that membrane surface characteristics exert a critical influence on all types of fouling (including organic, inorganic and biofouling). Surface hydrophilicity, roughness, charge density, and functional groups of the membrane active layer play a key role in biofouling. Several researchers have demonstrated that the increase of the hydrophilicity, alteration of the roughness, increasing the density of charged functional groups, and functionalization of the surface with antibacterial agents can improve the membrane resistance to biofouling. Early attempts in this direction focused on the development of hydrophilic surfaces to mitigate the membrane biofouling via anti-adhesion strategies, while their biofouling control was not satisfactory to provide long-term operation owing to continuous adsorption, multiplication, and regeneration of bacteria on the membrane surface. However, current research has shifted towards the design of bactericidal membranes via the integration of antibacterial nanomaterials, whether through incorporation or surface functionalization of the active

membrane layer. To achieve this goal, nanotechnology and bioinspired engineering have expanded the opportunities for the development of antifouling membranes. Various types of nanomaterials, including metal NPs, CNTs, graphene-based materials, zwitterionic materials, and hybrid nanomaterials, are promising candidates for next-generation FO desalination membranes and can be used to tailor the FO water flux, selectivity, biofouling-resistance, and mechanochemical stability. Generally, inorganic metal-based nanoparticles and carbon-based nanomaterials are considered promising antibacterial agents due to their strong biocidal properties, high stability, and commercial availability. Even though metal NPs offer desired antibacterial activity, the poor compatibility with the organic membranes still represents a technical limitation. Additionally, the release of metal NPs from the antibacterial membrane should be carefully considered as it will possibly trigger a risk to the aquatic ecosystem. Meanwhile, further research should focus on controlling the release rate to efficiently inactivate the bacteria for the longest possible time, ideally in the same order of

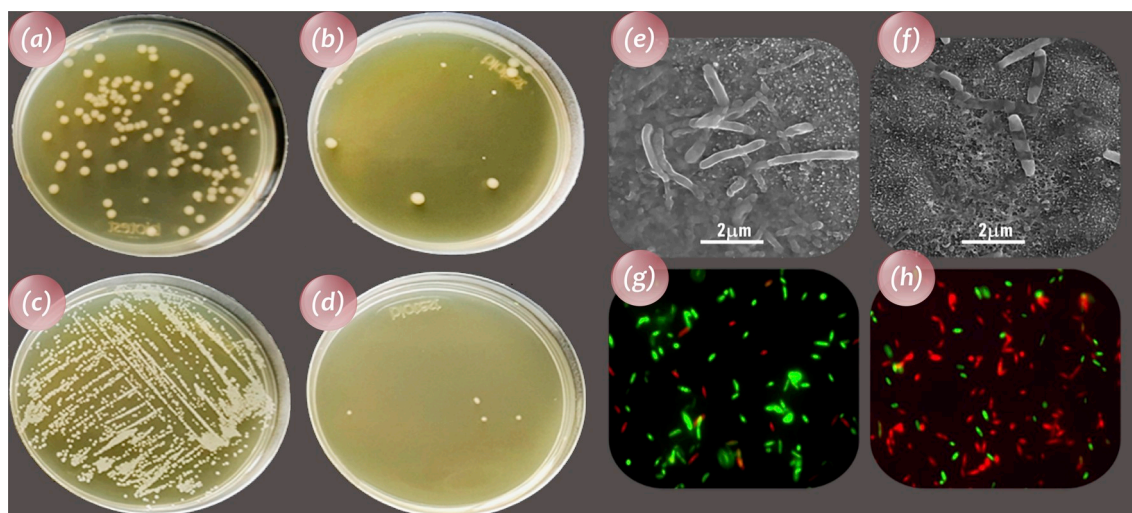


Fig. 17. CFU of TFC and TFN-0.5 membranes obtained by the standard plate count method. (a, b-left) *E. coli* and (c, d-left) *S. aureus* cells colonies. Bacterial inactivation on the membrane surface (right a-d images); (a) and (b) SEM micrographs, (c) and (d) fluorescent images of *E. coli* cells on the membrane surface, stained with SYTO 9 (green), and PI (red) for viable and dead cells, respectively; (a) and (c) TFC, and (b) and (d) TFN membranes [129]. reproduced with permission from Elsevier. (For interpretation of the references to color in this figure legend, the reader is referred to the Web version of this article.)

magnitude as the membrane lifespan. Despite extensive attempts to tailor the membrane surface properties by the selection of an appropriate antibiofouling strategy, a challenge still exists to attain efficient, long-lasting antibiofouling properties. This obstacle is not surprising due to the more complicated mechanism of biofouling than other membrane fouling phenomena.

With ongoing progress in nanotechnology, advanced approaches have directed the design of multifunctional hybrid nanocomposite FO membranes. Introducing nanomaterials has resulted in novel FO membranes with excellent separation performance and antibiofouling properties that have great potential to overcome the biofouling bottleneck. However, despite exceptional enhancements in antibiofouling performance, the applications of these FO membranes have yet to expand beyond the lab scale, and successful commercialization for large-scale industrial applications has been rare. The main challenges of incorporating nanomaterials into the hybrid nanocomposite FO membranes include NPs agglomeration, depletion due to NP release, and the high cost of nanomaterials. Agglomeration leads to heterogeneous distribution in or on the active layer, and this result usually translates into an alteration of the separation performance. In the last decade, many efforts have been made to mitigate this phenomenon, and among them, surface modification technology and optimizing the fabrication conditions have become more popular in membrane preparation due to the various options and multifunctional characteristics. A lot of researchers reported membrane modification via physical and chemical treatments to improve the membrane surface properties. In comparison to physical surface coating, chemical modification would be a more favorable approach to assure strong adhesion between the membrane surface and the modifying agent via covalent linkage; covalent linkage considerably decreases concerns related to antifouling agent release particularly during long-term operation. Although still in its infancy stage, surface functionalization is a promising approach to anchor biocides on the membrane surface and provides a high level of bacterial inactivation. Moreover, surface functionalization benefits from the dual synergic effect of both contact killing and release killing. Nonetheless, the choice of antibacterial agent largely depends on the cost and on laboratory criteria. Significant progress in bioinspired adhesion of PDA, supramolecular chemistry via in situ self-assembly, coupling chemistry have allowed enormous structural diversity (e.g., brush-like structure, hybrid nanocomposite surface and ultra-thin layer) for the facile and rational engineering design of membrane surfaces with functional antibiofouling properties. Although surface modification, either physical or chemical, is an effective strategy in tailoring surface properties, there are major challenges that may be classified as follows:

- (1) Decreased water flux due to the extra mass transfer resistance originated from the additional layer formed after modification; a trade-off between water flux reduction and antibiofouling property should be optimized.
- (2) Surface defects or increased roughness
- (3) Use of hazardous chemicals for the modification, increased fabrication cost, and safety and environmental issues.
- (4) Long-term chemical and mechanical stability of the surface-modified FO membranes.
- (5) Complex operation or difficult scale up: many approaches for surface modification are currently limited to the lab scale, and only a few methods are ready for commercial application.
- (6) Compatibility of the modifying agents with acid, alkaline or other cleaning materials.

Therefore, the following features should be taken into account when developing novel FO membranes through chemical modifications in the future: (i) simplifying functionalization approaches and optimizing reaction conditions to obtain a membrane with a thin, robust, and highly selective active layer, (ii) identifying more effective procedures and proper reagents for chemical modification, and (iii) targeting designed

functionalization schemes for each specific FO application.

Biomimetic adhesion approaches and bioinspired chemistries may offer fascinating breakthroughs. Living creatures provide unique biological materials that involve highly organized architectures with self-cleaning behavior. The progress in this interdisciplinary area will undoubtedly open up broad prospects and new eras in integrating biomimetic science and membrane science for sustainable water purification.

Finally, it is worth underlining that the goal of constructing anti-biofouling membrane surfaces is to attain long-lasting performance to meet the requirements of real applications. There still exists a noticeable gap between preliminary research and large-scale industrial applications. A vast majority of lab-scale experiments are implemented under very specific and narrow conditions, such as evaluating the anti-biofouling activity with only a few model strains, mainly *E. coli* and *P. aeruginosa*, and applying hydrodynamics that are rather different from those prevailing in most real-scale operations. As a result, there is a lack of understanding of the antibiofouling properties of surface-modified membranes with complex and multi-component feed streams, especially for their long-term applications.

As a final note, it is interesting to note that early work in FO long touted its resiliency under high fouling conditions. While FO was found to have some quantifiable benefits when it came to reduced fouling or improved cleanability, it is clear that fouling, and especially biofouling, still occur with regularity and the prevention of such events are challenging. While a number of approaches to countering biofouling have been considered (and reviewed here), given the lack of commercialized mixed matrix membranes with biocidal characteristics or otherwise modified membranes with similar properties, it might be time to ask ourselves as scientists if we are going down the right path. Do we continue to make “fancy” membranes which offer some degree of biofouling resistance, while at least temporarily avoiding the questions of scalable membrane production, or do we consider an alternative path.

As with organic and inorganic fouling, we might consider the approach taken by the RO industry today: prevent it ever from happening. However, biofouling is more challenging to prevent because of the fouling growth after the pretreatment. Accordingly, the only way to really prevent biofouling is to prevent this growth. The easiest way to do that is to allow for a disinfection residual to exist in the solution. That may mean that our efforts and resources are better directed toward chlorine tolerance rather than biocidal modification or in both domain simultaneously.

Declaration of competing interestCOI

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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References

- [1] C.D. Moody, J.O. Kessler, Forward osmosis extractors, *Desalination* 18 (1976) 283–295.
- [2] A. Subramani, J.G. Jacangelo, Emerging desalination technologies for water treatment: a critical review, *Water Res.* 75 (2015) 164–187.
- [3] R.V. Linares, Z. Li, S. Sarp, S.S. Bucs, G. Amy, J.S. Vrouwenvelder, Forward osmosis niches in seawater desalination and wastewater reuse, *Water Res.* 66 (2014) 122–139.
- [4] B. Marrot, A. Barrios-Martinez, P. Moulin, N. Roche, Industrial wastewater treatment in a membrane bioreactor: a review, *Environ. Prog.* 23 (2004) 59–68.

- [5] T.-S. Chung, L. Luo, C.F. Wan, Y. Cui, G. Amy, What is next for forward osmosis (FO) and pressure retarded osmosis (PRO), *Separ. Purif. Technol.* 156 (2015) 856–860.
- [6] C. Bhattacharjee, V.K. Saxena, S. Dutta, Fruit juice processing using membrane technology: a review, *Innov. Food Sci. Emerg. Technol.* 43 (2017) 136–153.
- [7] C. Zhao, J. Xue, F. Ran, S. Sun, Modification of polyethersulfone membranes – a review of methods, *Prog. Mater. Sci.* 58 (2013) 76–150.
- [8] M. Qasim, N.A. Darwish, S. Sarp, N. Hilal, Water desalination by forward (direct) osmosis phenomenon: a comprehensive review, *Desalination* 374 (2015) 47–69.
- [9] A. Mollahosseini, A. Rahimpour, A new concept in polymeric thin-film composite nanofiltration membranes with antibacterial properties, *Biofouling* 29 (2013) 537–548.
- [10] H. Maddah, A. Chogle, Biofouling in reverse osmosis: phenomena, monitoring, controlling and remediation, *Appl. Water Sci.* 7 (2017) 2637–2651.
- [11] W.L. Ang, A. Wahab Mohammad, D. Johnson, N. Hilal, Forward osmosis research trends in desalination and wastewater treatment: a review of research trends over the past decade, *J. Water Process Eng.* 31 (2019) 100886.
- [12] P. Lu, Y. Gao, A. Umar, T. Zhou, J. Wang, Z. Zhang, L. Huang, Q. Wang, Recent advances in cellulose-based forward osmosis membrane, *Sci. Adv. Mater.* 7 (2015) 2182–2192.
- [13] N. Akther, S. Phuntsho, Y. Chen, N. Ghaffour, H.K. Shon, Recent advances in nanomaterial-modified polyamide thin-film composite membranes for forward osmosis processes, *J. Membr. Sci.* 584 (2019) 20–45.
- [14] N. Akther, A. Sadiq, A. Giwa, S. Daer, H.A. Arafat, S.W. Hasan, Recent advancements in forward osmosis desalination: a review, *Chem. Eng. J.* 281 (2015) 502–522.
- [15] T. Nguyen, F.A. Roddick, L. Fan, Biofouling of water treatment membranes: a review of the underlying causes, monitoring techniques and control measures, *Membranes* 2 (2012) 804–840.
- [16] S. Phuntsho, S. Hong, M. Elimelech, H.K. Shon, Osmotic equilibrium in the forward osmosis process: modelling, experiments and implications for process performance, *J. Membr. Sci.* 453 (2014) 240–252.
- [17] B. Van der Bruggen, P. Luis, Forward osmosis: understanding the hype, *Rev. Chem. Eng.* 31 (2015) 1–12.
- [18] C.R. Martinetti, A.E. Childress, T.Y. Cath, High recovery of concentrated RO brines using forward osmosis and membrane distillation, *J. Membr. Sci.* 331 (2009) 31–39.
- [19] W. Xu, Q. Chen, Q. Ge, Recent advances in forward osmosis (FO) membrane: chemical modifications on membranes for FO processes, *Desalination* 419 (2017) 101–116.
- [20] M. Elimelech, W.A. Phillip, The future of seawater desalination: energy, technology, and the environment, *Science* 333 (2011) 712–717.
- [21] D. Li, Y. Yan, H. Wang, Recent advances in polymer and polymer composite membranes for reverse and forward osmosis processes, *Prog. Polym. Sci.* 61 (2016) 104–155.
- [22] Y. Cai, A critical review on draw solutes development for forward osmosis, *Desalination* 391 (2016) 16–29.
- [23] H. Luo, Q. Wang, T.C. Zhang, T. Tao, A. Zhou, L. Chen, X. Bie, A review on the recovery methods of draw solutes in forward osmosis, *J. Water Process Eng.* 4 (2014) 212–223.
- [24] A. Zirehpour, A. Rahimpour, M. Jahanshahi, M. Peyravi, Mixed matrix membrane application for olive oil wastewater treatment: process optimization based on Taguchi design method, *J. Environ. Manag.* 132 (2014) 113–120.
- [25] A.D. Kiadehi, A. Rahimpour, M. Jahanshahi, A.A. Ghoreyshi, Novel carbon nano-fibers (CNF)/polysulfone (PSF) mixed matrix membranes for gas separation, *J. Ind. Eng. Chem.* 22 (2015) 199–207.
- [26] I. Alsvik, M.-B. Hägg, Pressure retarded osmosis and forward osmosis membranes: materials and methods, *Polymers* 5 (2013) 303–327.
- [27] S. Pourjafar, M. Jahanshahi, A. Rahimpour, Optimization of TiO₂ modified poly (vinyl alcohol) thin film composite nanofiltration membranes using Taguchi method, *Desalination* 315 (2013) 107–114.
- [28] A. Rahimpour, B. Rajaeian, A. Hosienzadeh, S.S. Madaeni, F. Ghoreishi, Treatment of oily wastewater produced by washing of gasoline reserving tanks using self-made and commercial nanofiltration membranes, *Desalination* 265 (2011) 190–198.
- [29] S. Loeb, Sea Water Demineralization by Means of A Semipermeable Membrane, UCLA Dept. of Engineering Report, 1960.
- [30] A. Rahimpour, S. Madaeni, Y. Mansourpanah, High performance polyethersulfone UF membrane for manufacturing spiral wound module: preparation, morphology, performance, and chemical cleaning, *Polym. Adv. Technol.* 18 (2007) 403–410.
- [31] Z.-Y. Li, V. Yangali-Quintanilla, R. Valladares-Linares, Q. Li, T. Zhan, G. Amy, Flux patterns and membrane fouling propensity during desalination of seawater by forward osmosis, *Water Res.* 46 (2012) 195–204.
- [32] Y. Mansourpanah, S. Madaeni, A. Rahimpour, Preparation and investigation of separation properties of polyethersulfone supported poly (piperazineamide) nanofiltration membrane using microwave-assisted polymerization, *Separ. Purif. Technol.* 69 (2009) 234–242.
- [33] A. Zirehpour, A. Rahimpour, S. Khoshhal, M.D. Firouzjaei, A.A. Ghoreyshi, The impact of MOF feasibility to improve the desalination performance and antifouling properties of FO membranes, *RSC Adv.* 6 (2016) 70174–70185.
- [34] H. Wang, L. Li, X. Zhang, S. Zhang, Polyamide thin-film composite membranes prepared from a novel triamine 3, 5-diamino-N-(4-aminophenyl)-benzamide monomer and m-phenylenediamine, *J. Membr. Sci.* 353 (2010) 78–84.
- [35] S. Yu, M. Liu, Z. Lü, Y. Zhou, C. Gao, Aromatic-cycloaliphatic polyamide thin-film composite membrane with improved chlorine resistance prepared from m-phenylenediamine-4-methyl and cyclohexane-1, 3, 5-tricarboxyl chloride, *J. Membr. Sci.* 344 (2009) 155–164.
- [36] J. Zhao, Y. Su, X. He, X. Zhao, Y. Li, R. Zhang, Z. Jiang, Dopamine composite nanofiltration membranes prepared by self-polymerization and interfacial polymerization, *J. Membr. Sci.* 465 (2014) 41–48.
- [37] M.N.A. Seman, M. Khayet, N. Hilal, Nanofiltration thin-film composite polyester polyethersulfone-based membranes prepared by interfacial polymerization, *J. Membr. Sci.* 348 (2010) 109–116.
- [38] T. Wang, L. Dai, Q. Zhang, A. Li, S. Zhang, Effects of acyl chloride monomer functionality on the properties of polyamide reverse osmosis (RO) membrane, *J. Membr. Sci.* 440 (2013) 48–57.
- [39] L. Zhao, P.C.-Y. Chang, C. Yen, W.S.W. Ho, High-flux and fouling-resistant membranes for brackish water desalination, *J. Membr. Sci.* 425 (2013) 1–10.
- [40] H. Wang, Q. Zhang, S. Zhang, Positively charged nanofiltration membrane formed by interfacial polymerization of 3, 3', 5, 5'-biphenyl tetraacyl chloride and piperazine on a poly (acrylonitrile)(PAN) support, *J. Membr. Sci.* 378 (2011) 243–249.
- [41] G.N.B. Baroña, J. Lim, B. Jung, High performance thin film composite polyamide reverse osmosis membrane prepared via m-phenylenediamine and 2, 2'-benzidinedisulfonic acid, *Desalination* 291 (2012) 69–77.
- [42] Q.-F. An, W.-D. Sun, Q. Zhao, Y.-L. Ji, C.-J. Gao, Study on a novel nanofiltration membrane prepared by interfacial polymerization with zwitterionic amine monomers, *J. Membr. Sci.* 431 (2013) 171–179.
- [43] W. Xie, G.M. Geise, B.D. Freeman, H.-S. Lee, G. Byun, J.E. McGrath, Polyamide interfacial composite membranes prepared from m-phenylene diamine, trimesoyl chloride and a new disulfonated diamine, *J. Membr. Sci.* 403 (2012) 152–161.
- [44] B. Tang, C. Zou, P. Wu, Study on a novel polyester composite nanofiltration membrane by interfacial polymerization. II. The role of lithium bromide in the performance and formation of composite membrane, *J. Membr. Sci.* 365 (2010) 276–285.
- [45] Y.-C. Chiang, Y.-Z. Hsueh, R.-C. Ruaan, C.-J. Chuang, K.-L. Tung, Nanofiltration membranes synthesized from hyperbranched polyethyleneimine, *J. Membr. Sci.* 326 (2009) 19–26.
- [46] S. Yu, M. Ma, J. Liu, J. Tao, M. Liu, C. Gao, Study on polyamide thin-film composite nanofiltration membrane by interfacial polymerization of polyvinylamine (PVAm) and isophthaloyl chloride (IPC), *J. Membr. Sci.* 379 (2011) 164–173.
- [47] M. Liu, D. Wu, S. Yu, C. Gao, Influence of the polyacyl chloride structure on the reverse osmosis performance, surface properties and chlorine stability of the thin-film composite polyamide membranes, *J. Membr. Sci.* 326 (2009) 205–214.
- [48] L. Wang, D. Li, L. Cheng, L. Zhang, H. Chen, Synthesis of 4-aminobenzoylpiperazine for preparing the thin film composite nanofiltration membrane by interfacial polymerization with TMC, *Separ. Sci. Technol.* 48 (2013) 466–472.
- [49] S. Yu, M. Liu, X. Liu, C. Gao, Performance enhancement in interfacially synthesized thin-film composite polyamide-urethane reverse osmosis membrane for seawater desalination, *J. Membr. Sci.* 342 (2009) 313–320.
- [50] S.K. Jewrajka, A.V.R. Reddy, H.H. Rana, S. Mandal, S. Khullar, S. Haldar, N. Joshi, P.K. Ghosh, Use of 2, 4, 6-pyridinetri-carboxylic acid chloride as a novel co-monomer for the preparation of thin film composite polyamide membrane with improved bacterial resistance, *J. Membr. Sci.* 439 (2013) 87–95.
- [51] L.-F. Liu, Z.-B. Cai, J.-N. Shen, L.-X. Wu, E.M.V. Hoek, C.-J. Gao, Fabrication and characterization of a novel poly (amide-urethane@ imide) TFC reverse osmosis membrane with chlorine-tolerant property, *J. Membr. Sci.* 469 (2014) 397–409.
- [52] D.H.N. Perera, Q. Song, H. Qiblawey, E. Sivanah, Regulating the aqueous phase monomer balance for flux improvement in polyamide thin film composite membranes, *J. Membr. Sci.* 487 (2015) 74–82.
- [53] S. Hong, I.-C. Kim, T. Tak, Y.-N. Kwon, Interfacially synthesized chlorine-resistant polyimide thin film composite (TFC) reverse osmosis (RO) membranes, *Desalination* 309 (2013) 18–26.
- [54] Z. Zhang, S. Wang, H. Chen, Q. Liu, J. Wang, T. Wang, Preparation of polyamide membranes with improved chlorine resistance by bis-2, 6-N, N-(2-hydroxyethyl) diaminotoluene and trimesoyl chloride, *Desalination* 331 (2013) 16–25.
- [55] J.-b. Jin, D.-q. Liu, D.-d. Zhang, Y.-h. Yin, X.-y. Zhao, Y.-f. Zhang, Preparation of thin-film composite nanofiltration membranes with improved antifouling property and flux using 2, 2'-oxybis-ethylamine, *Desalination* 355 (2015) 141–146.
- [56] A. Bera, R.M. Gol, S. Chatterjee, S.K. Jewrajka, PEGylation and incorporation of triazine ring into thin film composite reverse osmosis membranes for enhancement of anti-organic and anti-biofouling properties, *Desalination* 360 (2015) 108–117.
- [57] J. Wei, C. Qiu, C.Y. Tang, R. Wang, A.G. Fane, Synthesis and characterization of flat-sheet thin film composite forward osmosis membranes, *J. Membr. Sci.* 372 (2011) 292–302.
- [58] X. Liu, H.Y. Ng, Double-blade casting technique for optimizing substrate membrane in thin-film composite forward osmosis membrane fabrication, *J. Membr. Sci.* 469 (2014) 112–126.
- [59] P. Xiao, L.D. Nghiem, Y. Yin, X.-M. Li, M. Zhang, G. Chen, J. Song, T. He, A sacrificial-layer approach to fabricate polysulfone support for forward osmosis thin-film composite membranes with reduced internal concentration polarisation, *J. Membr. Sci.* 481 (2015) 106–114.
- [60] X. Liu, H.Y. Ng, Fabrication of layered silica-polysulfone mixed matrix substrate membrane for enhancing performance of thin-film composite forward osmosis membrane, *J. Membr. Sci.* 481 (2015) 148–163.
- [61] G. Han, T.-S. Chung, M. Torida, S. Tamai, Thin-film composite forward osmosis membranes with novel hydrophilic supports for desalination, *J. Membr. Sci.* 423 (2012) 543–555.

- [62] D. Emadzadeh, W.J. Lau, T. Matsuura, M. Rahbari-Sisakht, A.F. Ismail, A novel thin film composite forward osmosis membrane prepared from PSF-TiO₂ nanocomposite substrate for water desalination, *Chem. Eng. J.* 237 (2014) 70–80.
- [63] N. Widojo, T.-S. Chung, M. Weber, C. Maletzko, V. Warzelhan, The role of sulphonated polymer and macrovoid-free structure in the support layer for thin-film composite (TFC) forward osmosis (FO) membranes, *J. Membr. Sci.* 383 (2011) 214–223.
- [64] Y. Wang, R. Ou, H. Wang, T. Xu, Graphene oxide modified graphitic carbon nitride as a modifier for thin film composite forward osmosis membrane, *J. Membr. Sci.* 475 (2015) 281–289.
- [65] N. Widojo, T.-S. Chung, M. Weber, C. Maletzko, V. Warzelhan, A sulfonated polyphenylenesulfone (sPPSU) as the supporting substrate in thin film composite (TFC) membranes with enhanced performance for forward osmosis (FO), *Chem. Eng. J.* 220 (2013) 15–23.
- [66] N. Ma, J. Wei, S. Qi, Y. Zhao, Y. Gao, C.Y. Tang, Nanocomposite substrates for controlling internal concentration polarization in forward osmosis membranes, *J. Membr. Sci.* 441 (2013) 54–62.
- [67] I.L. Alsvik, K.R. Zodrow, M. Elimelech, M.-B. Hägg, Polyamide formation on a cellulose triacetate support for osmotic membranes: effect of linking molecules on membrane performance, *Desalination* 312 (2013) 2–9.
- [68] Y. Wang, R. Ou, Q. Ge, H. Wang, T. Xu, Preparation of polyethersulfone/carbon nanotube substrate for high-performance forward osmosis membrane, *Desalination* 330 (2013) 70–78.
- [69] G. Han, S. Zhang, X. Li, N. Widojo, T.-S. Chung, Thin film composite forward osmosis membranes based on polydopamine modified polysulfone substrates with enhancements in both water flux and salt rejection, *Chem. Eng. Sci.* 80 (2012) 219–231.
- [70] M. Tian, C. Qiu, Y. Liao, S. Chou, R. Wang, Preparation of polyamide thin film composite forward osmosis membranes using electrospun polyvinylidene fluoride (PVDF) nanofibers as substrates, *Separ. Purif. Technol.* 118 (2013) 727–736.
- [71] Y. Sun, L. Xue, Y. Zhang, X. Zhao, Y. Huang, X. Du, High flux polyamide thin film composite forward osmosis membranes prepared from porous substrates made of polysulfone and polyethersulfone blends, *Desalination* 336 (2014) 72–79.
- [72] M. Yasukawa, S. Mishima, M. Shibuya, D. Saeki, T. Takahashi, T. Miyoshi, H. Matsuyama, Preparation of a forward osmosis membrane using a highly porous polyketone microfiltration membrane as a novel support, *J. Membr. Sci.* 487 (2015) 51–59.
- [73] K.Y. Wang, T.S. Chung, G. Amy, Developing thin-film-composite forward osmosis membranes on the PES/SPSF substrate through interfacial polymerization, *AIChE J.* 58 (2012) 770–781.
- [74] C. Klayson, S. Hermans, A. Gahlaut, S. Van Craenenbroeck, I.F.J. Vankelecom, Polyamide/Polyacrylonitrile (PA/PAN) thin film composite osmosis membranes: film optimization, characterization and performance evaluation, *J. Membr. Sci.* 445 (2013) 25–33.
- [75] Y.H. Cho, J. Han, S. Han, M.D. Guiver, H.B. Park, Polyamide thin-film composite membranes based on carboxylated polysulfone microporous support membranes for forward osmosis, *J. Membr. Sci.* 445 (2013) 220–227.
- [76] L. Huang, J.R. McCutcheon, Impact of support layer pore size on performance of thin film composite membranes for forward osmosis, *J. Membr. Sci.* 483 (2015) 25–33.
- [77] S.-B. Kwon, J.S. Lee, S.J. Kwon, S.-T. Yun, S. Lee, J.-H. Lee, Molecular layer-by-layer assembled forward osmosis membranes, *J. Membr. Sci.* 488 (2015) 111–120.
- [78] R.C. Ong, T.-S. Chung, J.S. de Wit, B.J. Helmer, Novel cellulose ester substrates for high performance flat-sheet thin-film composite (TFC) forward osmosis (FO) membranes, *J. Membr. Sci.* 473 (2015) 63–71.
- [79] N.Y. Yip, A. Tiraferri, W.A. Phillip, J.D. Schiffman, M. Elimelech, High performance thin-film composite forward osmosis membrane, *Environ. Sci. Technol.* 44 (2010) 3812–3818.
- [80] X. Li, K.Y. Wang, B. Helmer, T.-S. Chung, Thin-film composite membranes and formation mechanism of thin-film layers on hydrophilic cellulose acetate propionate substrates for forward osmosis processes, *Ind. Eng. Chem. Res.* 51 (2012) 10039–10050.
- [81] R. Wang, L. Shi, C.Y. Tang, S. Chou, C. Qiu, A.G. Fane, Characterization of novel forward osmosis hollow fiber membranes, *J. Membr. Sci.* 355 (2010) 158–167.
- [82] P.H.H. Duong, S. Chisca, P.-Y. Hong, H. Cheng, S.P. Nunes, T.-S. Chung, Hydroxyl functionalized polytriazole-co-polyoxadiazole as substrates for forward osmosis membranes, *ACS Appl. Mater. Interfaces* 7 (2015) 3960–3973.
- [83] P. Sukitpaneevit, T.-S. Chung, High performance thin-film composite forward osmosis hollow fiber membranes with macrovoid-free and highly porous structure for sustainable water production, *Environ. Sci. Technol.* 46 (2012) 7358–7365.
- [84] L. Shi, S.R. Chou, R. Wang, W.X. Fang, C.Y. Tang, A.G. Fane, Effect of substrate structure on the performance of thin-film composite forward osmosis hollow fiber membranes, *J. Membr. Sci.* 382 (2011) 116–123.
- [85] W. Fang, C. Liu, L. Shi, R. Wang, Composite forward osmosis hollow fiber membranes: integration of RO-and NF-like selective layers for enhanced organic fouling resistance, *J. Membr. Sci.* 492 (2015) 147–155.
- [86] S. Chou, L. Shi, R. Wang, C.Y. Tang, C. Qiu, A.G. Fane, Characteristics and potential applications of a novel forward osmosis hollow fiber membrane, *Desalination* 261 (2010) 365–372.
- [87] W. Fang, R. Wang, S. Chou, L. Setiawan, A.G. Fane, Composite forward osmosis hollow fiber membranes: integration of RO-and NF-like selective layers to enhance membrane properties of anti-scaling and anti-internal concentration polarization, *J. Membr. Sci.* 394 (2012) 140–150.
- [88] M. Obaid, Z.K. Ghouri, O.A. Fadali, K.A. Khalil, A.A. Almajid, N.A. Barakat, Amorphous SiO₂ NP-incorporated poly (vinylidene fluoride) electrospun nanofiber membrane for high flux forward osmosis desalination, *ACS Appl. Mater. Interfaces* 8 (2016) 4561–4574.
- [89] Q. Zhang, Y.W. Jie, W.L.C. Loong, J. Zhang, A.G. Fane, S. Kjelleberg, S.A. Rice, D. McDougald, Characterization of biofouling in a lab-scale forward osmosis membrane bioreactor (FOMBR), *Water Res.* 58 (2014) 141–151.
- [90] L.V.L.V. Evans, *Biofilms: Recent Advances in Their Study and Control*, Harwood Academic Publishers, 2003.
- [91] J.S. Baker, L.Y. Dudley, Biofouling in membrane systems - a review, *Desalination* 118 (1998) 81–89.
- [92] M. Herzberg, S. Kang, M. Elimelech, Role of extracellular polymeric substances (EPS) in biofouling of reverse osmosis membranes, *Environ. Sci. Technol.* 43 (2009) 4393–4398.
- [93] S.T. Kang, A. Subramani, E.M.V. Hoek, M.A. Deshusses, M.R. Matsumoto, Direct observation of biofouling in cross-flow microfiltration: mechanisms of deposition and release, *J. Membr. Sci.* 244 (2004) 151–165.
- [94] M.F.A. Goosen, S.S. Sablani, H. Al-Hinai, S. Al-Obeidani, R. Al-Belushi, D. Jackson, Fouling of reverse osmosis and ultrafiltration membranes: a critical review, *Sep. Sci. Technol.* 39 (2004) 2261–2297.
- [95] Q. She, R. Wang, A.G. Fane, C.Y. Tang, Membrane fouling in osmotically driven membrane processes: a review, *J. Membr. Sci.* 499 (2016) 201–233.
- [96] M. Peyravi, A. Rahimpour, M. Jahanshahi, Developing nanocomposite PI membranes: morphology and performance to glycerol removal at the downstream processing of biodiesel production, *J. Membr. Sci.* 473 (2015) 72–84.
- [97] J.A. Redman, S.L. Walker, M. Elimelech, Bacterial adhesion and transport in porous media: role of the secondary energy minimum, *Environ. Sci. Technol.* 38 (2004) 1777–1785.
- [98] A.C. Olofsson, M. Hermansson, H. Elwing, N-acetyl-L-cysteine affects growth, extracellular polysaccharide production, and bacterial biofilm formation on solid surfaces, *Appl. Environ. Microbiol.* 69 (2003) 4814–4822.
- [99] U. Passow, Transparent exopolymer particles (TEP) in aquatic environments, *Prog. Oceanogr.* 55 (2002) 287–333.
- [100] E. Bar-Zeev, U. Passow, S. Romero-Vargas Castrillón, M. Elimelech, Transparent exopolymer particles: from aquatic environments and engineered systems to membrane biofouling, *Environ. Sci. Technol.* 49 (2015) 691–707.
- [101] G.A. O'Toole, R. Kolter, Flagellar and twitching motility are necessary for *Pseudomonas aeruginosa* biofilm development, *Mol. Microbiol.* 30 (1998) 295–304.
- [102] Y. Liu, H.H.P. Fang, Influences of extracellular polymeric substances (EPS) on flocculation, settling, and dewatering of activated sludge, *Crit. Rev. Environ. Sci. Technol.* 33 (2003) 237–373.
- [103] H. Lin, M. Zhang, F. Wang, F. Meng, B.Q. Liao, H. Hong, J. Chen, W. Gao, A critical review of extracellular polymeric substances (EPSs) in membrane bioreactors: characteristics, roles in membrane fouling and control strategies, *J. Membr. Sci.* 460 (2014) 110–125.
- [104] U. Metzger, P. Le-Clech, R.M. Stuetz, F.H. Frimmel, V. Chen, Characterisation of polymeric fouling in membrane bioreactors and the effect of different filtration modes, *J. Membr. Sci.* 301 (2007) 180–189.
- [105] J. Mansouri, S. Harrison, V. Chen, Strategies for controlling biofouling in membrane filtration systems: challenges and opportunities, *J. Mater. Chem.* 20 (2010) 4567–4586.
- [106] H.C. Flemming, Reverse osmosis membrane biofouling, *Exp. Thermal Fluid Sci.* 14 (1997) 382–391.
- [107] W. Guo, H.H. Ngo, J. Li, A mini-review on membrane fouling, *Bioresour. Technol.* 122 (2012) 27–34.
- [108] H.C. Flemming, Biofouling in water systems - cases, causes and countermeasures, *Appl. Microbiol. Biotechnol.* 59 (2002) 629–640.
- [109] D.L. Shaffer, N.Y. Yip, J. Giron, M. Elimelech, Seawater desalination for agriculture by integrated forward and reverse osmosis: improved product water quality for potentially less energy, *J. Membr. Sci.* 415 (2012) 1–8.
- [110] C. Boo, S. Lee, M. Elimelech, Z. Meng, S. Hong, Colloidal fouling in forward osmosis: role of reverse salt diffusion, *J. Membr. Sci.* 390 (2012) 277–284.
- [111] S. Zou, Y. Gu, D. Xiao, C.Y. Tang, The role of physical and chemical parameters on forward osmosis membrane fouling during algae separation, *J. Membr. Sci.* 366 (2011) 356–362.
- [112] S. Phuntsho, H.K. Shon, S. Hong, S. Lee, S. Vigneswaran, A novel low energy fertilizer driven forward osmosis desalination for direct fertigation: evaluating the performance of fertilizer draw solutions, *J. Membr. Sci.* 375 (2011) 172–181.
- [113] S. Zou, Y.N. Wang, F. Wicaksana, T. Aung, P.C.Y. Wong, A.G. Fane, C.Y. Tang, Direct microscopic observation of forward osmosis membrane fouling by microalgae: critical flux and the role of operational conditions, *J. Membr. Sci.* 436 (2013) 174–185.
- [114] L. Chen, Y. Gu, C. Cao, J. Zhang, J.W. Ng, C. Tang, Performance of a submerged anaerobic membrane bioreactor with forward osmosis membrane for low-strength wastewater treatment, *Water Res.* 50 (2014) 114–123.
- [115] R. Valladares Linares, S.S. Buys, Z. Li, M. AbuGhdeeb, G. Amy, J. S. Vrouwenvelder, Impact of spacer thickness on biofouling in forward osmosis, *Water Res.* 57 (2014) 223–233.
- [116] J.S. Vrouwenvelder, D.A. Graf von der Schulenburg, J.C. Kruijthof, M.L. Johns, M. C.M. van Loosdrecht, Biofouling of spiral-wound nanofiltration and reverse osmosis membranes: a feed spacer problem, *Water Res.* 43 (2009) 583–594.
- [117] H. Yoon, Y. Baek, J. Yu, J. Yoon, Biofouling occurrence process and its control in the forward osmosis, *Desalination* 325 (2013) 30–36.
- [118] S.E. Kwan, E. Bar-Zeev, M. Elimelech, Biofouling in forward osmosis and reverse osmosis: measurements and mechanisms, *J. Membr. Sci.* 493 (2015) 703–708.

- [119] A. Achilli, T.Y. Cath, E.A. Marchand, A.E. Childress, The forward osmosis membrane bioreactor: a low fouling alternative to mbr processes, *Desalination* 239 (2009) 10–21.
- [120] E.R. Cornelissen, D. Harmsen, K.F. de Korte, C.J. Ruiken, J.J. Qin, H. Oo, L. P. Wessels, Membrane fouling and process performance of forward osmosis membranes on activated sludge, *J. Membr. Sci.* 319 (2008) 158–168.
- [121] S. Liang, C. Liu, L. Song, Soluble microbial products in membrane bioreactor operation: behaviors, characteristics, and fouling potential, *Water Res.* 41 (2007) 95–101.
- [122] N. Park, B. Kwon, I.S. Kim, J. Cho, Biofouling potential of various NF membranes with respect to bacteria and their soluble microbial products (SMP): characterizations, flux decline, and transport parameters, *J. Membr. Sci.* 258 (2005) 43–54.
- [123] A. Ramesh, D.J. Lee, J.Y. Lai, Membrane biofouling by extracellular polymeric substances or soluble microbial products from membrane bioreactor sludge, *Appl. Microbiol. Biotechnol.* 74 (2007) 699–707.
- [124] Q. Wang, M. Hu, Z. Wang, W. Hu, J. Cao, Z.-C. Wu, Uniqueness of biofouling in forward osmosis systems: mechanisms and control, *Crit. Rev. Environ. Sci. Technol.* 48 (2018) 1031–1066.
- [125] R.A. Al-Juboori, T. Yusaf, Biofouling in RO system: mechanisms, monitoring and controlling, *Desalination* 302 (2012) 1–23.
- [126] Y.-N. Wang, J. Wei, Q. She, F. Pacheco, C.Y. Tang, Microscopic characterization of FO/PRO membranes—a comparative study of CLSM, TEM and SEM, *Environ. Sci. Technol.* 46 (2012) 9995–10003.
- [127] Q. Wang, M. Hu, Z. Wang, W. Hu, J. Cao, Z.-C. Wu, Uniqueness of biofouling in forward osmosis systems: mechanisms and control, *Crit. Rev. Environ. Sci. Technol.* 48 (2018) 1031–1066.
- [128] D. Nivens, R. Palmer, D. White, Continuous nondestructive monitoring of microbial biofilms: a review of analytical techniques, *J. Ind. Microbiol.* 15 (1995) 263–276.
- [129] S.F. Seyedpour, A. Rahimpour, A.A. Shamsabadi, M. Soroush, Improved performance and antifouling properties of thin-film composite polyamide membranes modified with nano-sized bactericidal graphene quantum dots for forward osmosis, *Chem. Eng. Res. Des.* 139 (2018) 321–334.
- [130] C. Liu, A.F. Faria, J. Ma, M. Elimelech, Mitigation of biofilm development on thin-film composite membranes functionalized with zwitterionic polymers and silver nanoparticles, *Environ. Sci. Technol.* 51 (2016) 182–191.
- [131] C.-M. Kim, S.-J. Kim, L.H. Kim, M.S. Shin, H.-W. Yu, I.S. Kim, Effects of phosphate limitation in feed water on biofouling in forward osmosis (FO) process, *Desalination* 349 (2014) 51–59.
- [132] Y. Miura, Y. Watanabe, S. Okabe, Membrane biofouling in pilot-scale membrane bioreactors (MBRs) treating municipal wastewater: impact of biofilm formation, *Environ. Sci. Technol.* 41 (2007) 632–638.
- [133] R. Valladares Linares, Hybrid membrane system for desalination and wastewater treatment: integrating forward osmosis and low pressure reverse osmosis, 2014.
- [134] S. Madaeni, N.T. Hasankiadeh, A. Kurdian, A. Rahimpour, Modeling and optimization of membrane fabrication using artificial neural network and genetic algorithm, *Separ. Purif. Technol.* 76 (2010) 33–43.
- [135] S.S. Bucs, R.V. Linares, J.S. Vrouwenvelder, C. Picioreanu, Biofouling in forward osmosis systems: an experimental and numerical study, *Water Res.* 106 (2016) 86–97.
- [136] Z. Wang, J. Tang, C. Zhu, Y. Dong, Q. Wang, Z. Wu, Chemical cleaning protocols for thin film composite (TFC) polyamide forward osmosis membranes used for municipal wastewater treatment, *J. Membr. Sci.* 475 (2015) 184–192.
- [137] T. Fujioka, K. Nguyen, A. Hoang, T. Ueyama, H. Yasui, M. Terashima, L. Nghiem, Biofouling mitigation by chloramination during forward osmosis filtration of wastewater, *Int. J. Environ. Res. Public Health* 15 (2018) 2124.
- [138] A. Bogler, S. Lin, E. Bar-Zeev, Biofouling of membrane distillation, forward osmosis and pressure retarded osmosis: principles, impacts and future directions, *J. Membr. Sci.* 542 (2017) 378–398.
- [139] X. Wang, T. Hu, Z. Wang, X. Li, Y. Ren, Permeability recovery of fouled forward osmosis membranes by chemical cleaning during a long-term operation of anaerobic osmotic membrane bioreactors treating low-strength wastewater, *Water Res.* 123 (2017) 505–512.
- [140] Y. Chun, D. Mulcahy, L. Zou, I.S. Kim, A short review of membrane fouling in forward osmosis processes, *Membranes* 7 (2017) 30.
- [141] K. Lutchmiah, A.R.D. Verliefde, K. Roest, L.C. Rietveld, E.R. Cornelissen, Forward osmosis for application in wastewater treatment: a review, *Water Res.* 58 (2014) 179–197.
- [142] P. Bacchin, D. Si-Hassen, V. Starov, M.J. Clifton, P. Aimar, A unifying model for concentration polarization, gel-layer formation and particle deposition in cross-flow membrane filtration of colloidal suspensions, *Chem. Eng. Sci.* 57 (2002) 77–91.
- [143] M.R. Esfahani, S.A. Aktij, Z. Dabaghian, M.D. Firouzjaei, A. Rahimpour, J. Eke, I. C. Escobar, M. Abolhassani, L.F. Greenlee, A.R. Esfahani, A. Sadmani, N. Koutahzadeh, Nanocomposite membranes for water separation and purification: fabrication, modification, and applications, *Separ. Purif. Technol.* 213 (2019) 465–499.
- [144] M. Mozafari, S.F. Seyedpour, S.K. Salestan, A. Rahimpour, A.A. Shamsabadi, M. D. Firouzjaei, M.R. Esfahani, A. Tiraferrri, H. Mohsenian, M. Sangermano, M. Soroush, Facile Cu-BTC surface modification of thin chitosan film coated polyethersulfone membranes with improved antifouling properties for sustainable removal of manganese, *J. Membr. Sci.* 588 (2019) 117200.
- [145] M.R. Esfahani, N. Koutahzadeh, A.R. Esfahani, M.D. Firouzjaei, B. Anderson, L. Peck, A novel gold nanocomposite membrane with enhanced permeation, rejection and self-cleaning ability, *J. Membr. Sci.* 573 (2019) 309–319.
- [146] V. Kochkodan, D.J. Johnson, N. Hilal, Polymeric membranes: surface modification for minimizing (bio) colloidal fouling, *Adv. Colloid Interface Sci.* 206 (2014) 116–140.
- [147] S. Mokhtari, A. Rahimpour, A.A. Shamsabadi, S. Habibzadeh, M. Soroush, Enhancing performance and surface antifouling properties of polysulfone ultrafiltration membranes with salicylate-alumoxane nanoparticles, *Appl. Surf. Sci.* 393 (2017) 93–102.
- [148] C.Y. Tang, Y.-N. Kwon, J.O. Leckie, Effect of membrane chemistry and coating layer on physicochemical properties of thin film composite polyamide RO and NF membranes: II. Membrane physicochemical properties and their dependence on polyamide and coating layers, *Desalination* 242 (2009) 168–182.
- [149] T. Humplik, J. Lee, S.C. O'hern, B.A. Fellman, M.A. Baig, S.F. Hassan, M.A. Atieh, F. Rahman, T. Laoui, R. Karnik, Nanostructured materials for water desalination, *Nanotechnology* 22 (2011) 292001.
- [150] J. Gilron, S. Belfer, P. Väisänen, M. Nyström, Effects of surface modification on antifouling and performance properties of reverse osmosis membranes, *Desalination* 140 (2001) 167–179.
- [151] F.-Q. Nie, Z.-K. Xu, P. Ye, J. Wu, P. Seta, Acrylonitrile-based copolymer membranes containing reactive groups: effects of surface-immobilized poly (ethylene glycol) s on anti-fouling properties and blood compatibility, *Polymer* 45 (2004) 399–407.
- [152] M. Asadollahi, D. Bastani, S.A. Musavi, Enhancement of surface properties and performance of reverse osmosis membranes after surface modification: a review, *Desalination* 420 (2017) 330–383.
- [153] B. Van der Bruggen, M. Mänttäri, M. Nyström, Drawbacks of applying nanofiltration and how to avoid them: a review, *Separ. Purif. Technol.* 63 (2008) 251–263.
- [154] S. Novak, U. Maver, Š. Peternel, P. Venturini, M. Bele, M. Gaberšček, Electrophoretic deposition as a tool for separation of protein inclusion bodies from host bacteria in suspension, *Colloid. Surf. Physicochem. Eng. Asp.* 340 (2009) 155–160.
- [155] J.D. Bryers, Biofilms and the technological implications of microbial cell adhesion, *Colloids Surfaces B Biointerfaces* 2 (1994) 9–23.
- [156] F. Perreault, H. Jaramillo, M. Xie, M. Ude, M. Elimelech, Biofouling mitigation in forward osmosis using graphene oxide functionalized thin-film composite membranes, *Environ. Sci. Technol.* 50 (2016) 5840–5848.
- [157] M. Morra, C. Cassinelli, Polymer Edition, Bacterial adhesion to polymer surfaces: a critical review of surface thermodynamic approaches, *J. Biomater. Sci. Polym. Ed.* 9 (1998) 55–74.
- [158] A. Tiraferrri, Y. Kang, E.P. Giannelis, M. Elimelech, Highly hydrophilic thin-film composite forward osmosis membranes functionalized with surface-tailored nanoparticles, *ACS Appl. Mater. Interfaces* 4 (2012) 5044–5053.
- [159] A. Tiraferrri, M. Elimelech, Direct quantification of negatively charged functional groups on membrane surfaces, *J. Membr. Sci.* 389 (2012) 499–508.
- [160] D. Rana, T. Matsuura, Surface modifications for antifouling membranes, *Chem. Rev.* 110 (2010) 2448–2471.
- [161] Z.-K. Xu, X.-J. Huang, L.-S. Wan, Surface engineering of polymer membranes, 2009.
- [162] J. Ayyavoo, T.P.N. Nguyen, B.-M. Jun, I.-C. Kim, Y.-N. Kwon, Protection of polymeric membranes with antifouling surfacing via surface modifications, *Colloid. Surf. Physicochem. Eng. Asp.* 506 (2016) 190–201.
- [163] J. Li, L. Yin, G. Qiu, X. Li, Q. Liu, J. Xie, A photo-bactericidal thin film composite membrane for forward osmosis, *J. Mater. Chem.* 3 (2015) 6781–6786.
- [164] C. Xing, L. Liu, H. Tang, X. Feng, Q. Yang, S. Wang, G.C. Bazan, Design guidelines for conjugated polymers with light-activated anticancer activity, *Adv. Funct. Mater.* 21 (2011) 4058–4067.
- [165] C. Ringot, V. Sol, R. Granet, P. Krausz, Porphyrin-grafted cellulose fabric: new photobactericidal material obtained by “click-chemistry” reaction, *Mater. Lett.* 63 (2009) 1889–1891.
- [166] X. Zhao, R. Zhang, Y. Liu, M. He, Y. Su, C. Gao, Z. Jiang, Antifouling membrane surface construction: chemistry plays a critical role, *J. Membr. Sci.* 551 (2018) 145–171.
- [167] C. Wang, Q. Yan, H.-B. Liu, X.-H. Zhou, S.-J. Xiao, Different EDC/NHS activation mechanisms between PAA and PMAA brushes and the following amidation reactions, *Langmuir* 27 (2011) 12058–12068.
- [168] A. Tiraferrri, C.D. Vecitis, M. Elimelech, Covalent binding of single-walled carbon nanotubes to polyamide membranes for antimicrobial surface properties, *ACS Appl. Mater. Interfaces* 3 (2011) 2869–2877.
- [169] P.K.S. Mural, S. Jain, S. Kumar, G. Madras, S. Bose, Unimpeded permeation of water through biocidal graphene oxide sheets anchored on to 3D porous polyolefinic membranes, *Nanoscale* 8 (2016) 8048–8057.
- [170] M.D. Guiver, G.P. Robertson, Chemical modification of polysulfones: a facile method of preparing azide derivatives from lithiated polysulfone intermediates, *Macromolecules* 28 (1995) 294–301.
- [171] B. Van der Bruggen, Chemical modification of polyethersulfone nanofiltration membranes: a review, *J. Appl. Polym. Sci.* 114 (2009) 630–642.
- [172] D. Lu, H. Zou, R. Guan, H. Dai, L. Lu, Sulfonation of polyethersulfone by chlorosulfonic acid, *Polym. Bull.* 54 (2005) 21–28.
- [173] Z. Yang, Y. Wu, J. Wang, B. Cao, C.Y. Tang, In situ reduction of silver by polydopamine: a novel antimicrobial modification of a thin-film composite polyamide membrane, *Environ. Sci. Technol.* 50 (2016) 9543–9550.
- [174] X. Li, A. Sotto, J. Li, B. Van der Bruggen, Progress and perspectives for synthesis of sustainable antifouling composite membranes containing in situ generated nanoparticles, *J. Membr. Sci.* 524 (2017) 502–528.

- [175] X. Zhu, R. Bai, K.-H. Wee, C. Liu, S.-L. Tang, Membrane surfaces immobilized with ionic or reduced silver and their anti-biofouling performances, *J. Membr. Sci.* 363 (2010) 278–286.
- [176] X. Cao, M. Tang, F. Liu, Y. Nie, C. Zhao, Immobilization of silver nanoparticles onto sulfonated polyethersulfone membranes as antibacterial materials, *Colloids Surfaces B Biointerfaces* 81 (2010) 555–562.
- [177] Y.Y.N. Bonggotgetsakul, R.W. Cattrall, S.D. Kolev, A method for the coating of a polymer inclusion membrane with a monolayer of silver nanoparticles, *J. Membr. Sci.* 428 (2013) 142–149.
- [178] Q. Liu, Z. Zhou, G. Qiu, J. Li, J. Xie, J.Y. Lee, Surface reaction route to increase the loading of antimicrobial Ag nanoparticles in forward osmosis membranes, *ACS Sustain. Chem. Eng.* 3 (2015) 2959–2966.
- [179] M. Ben-Sasson, X. Lu, S. Nejati, H. Jaramillo, M. Elimelech, In situ surface functionalization of reverse osmosis membranes with biocidal copper nanoparticles, *Desalination* 388 (2016) 1–8.
- [180] W. Li, Metal–organic framework membranes: production, modification, and applications, *Prog. Mater. Sci.* 100 (2019) 21–63.
- [181] W. Li, Y. Zhang, Q. Li, G. Zhang, Metal–organic framework composite membranes: synthesis and separation applications, *Chem. Eng. Sci.* 135 (2015) 232–257.
- [182] M. Wu, H. Ye, F. Zhao, B. Zeng, High-quality metal–organic framework ZIF-8 membrane supported on electrodeposited ZnO/2-methylimidazole nanocomposite: efficient adsorbent for the enrichment of acidic drugs, *Sci. Rep.* 7 (2017) 39778.
- [183] I. Stassen, N. Campagnol, J. Franssaer, P. Vereecken, D. De Vos, R. Ameloot, Solvent-free synthesis of supported ZIF-8 films and patterns through transformation of deposited zinc oxide precursors, *Cryst. Eng. Commun.* 15 (2013) 9308–9311.
- [184] M. Dadashi Firouzaei, A. Arabi Shamsabadi, S. Aghapour Aktij, S.F. Seyedpour, M. Sharifian Gh, A. Rahimpour, M. Rabhani Esfahani, M. Ulbricht, M. Soroush, Exploiting synergetic effects of graphene oxide and a silver-based metal–organic framework to enhance antifouling and anti-biofouling properties of thin film nanocomposite membranes, *ACS Appl. Mater. Interfaces* 10 (2018) 42967–42978.
- [185] A. Rahimpour, S.F. Seyedpour, S. Aghapour Aktij, M. Dadashi Firouzaei, A. Zirehpour, A. Arabi Shamsabadi, S. Khoshhal Salestan, M. Jabbari, M. Soroush, Simultaneous improvement of antimicrobial, antifouling, and transport properties of forward osmosis membranes with immobilized highly-compatible polyrhodanine nanoparticles, *Environ. Sci. Technol.* 52 (2018) 5246–5258.
- [186] J. Quirós, K. Boltes, S. Aguado, R.G. de Villoria, J.J. Vilatela, R. Rosal, Antimicrobial metal–organic frameworks incorporated into electrospon fibers, *Chem. Eng. J.* 262 (2015) 189–197.
- [187] M.S. Mauter, Y. Wang, K.C. Okemgbo, C.O. Osuji, E.P. Giannelis, M. Elimelech, Antifouling ultrafiltration membranes via post-fabrication grafting of biocidal nanomaterials, *ACS Appl. Mater. Interfaces* 3 (2011) 2861–2868.
- [188] I. Banerjee, R.C. Pangule, R.S. Kane, Antifouling coatings: recent developments in the design of surfaces that prevent fouling by proteins, bacteria, and marine organisms, *Adv. Mater.* 23 (2011) 690–718.
- [189] X. Lu, J. Ye, D. Zhang, R. Xie, R.F. Bogale, Y. Sun, L. Zhao, Q. Zhao, G. Ning, Silver carboxylate metal–organic frameworks with highly antibacterial activity and biocompatibility, *J. Inorg. Biochem.* 138 (2014) 114–121.
- [190] A. Tăbăcaru, C. Pettinari, F. Marchetti, C. di Nicola, K.V. Domasevitch, S. Galli, N. Masciocchi, S. Scuri, I. Grappasonni, M. Cocchioni, Antibacterial action of 4, 4'-bipyrazolyl-based silver (I) coordination polymers embedded in PE disks, *Inorg. Chem.* 51 (2012) 9775–9788.
- [191] R. Ameloot, F. Vermoortele, W. Vanhove, M.B.J. Roefsaers, B.F. Sels, D.E. De Vos, Interfacial synthesis of hollow metal–organic framework capsules demonstrating selective permeability, *Nat. Chem.* 3 (2011) 382.
- [192] F. Razi, I. Sawada, Y. Ohmukai, T. Maruyama, H. Matsuyama, The improvement of antibiofouling efficiency of polyethersulfone membrane by functionalization with zwitterionic monomers, *J. Membr. Sci.* 401 (2012) 292–299.
- [193] J. Wang, Z. Wang, J. Wang, S. Wang, Improving the water flux and bio-fouling resistance of reverse osmosis (RO) membrane through surface modification by zwitterionic polymer, *J. Membr. Sci.* 493 (2015) 188–199.
- [194] K.W. Kolewe, K.M. Dobosz, K.A. Rieger, C.-C. Chang, T. Emrick, J.D. Schiffman, Antifouling electrospun nanofiber mats functionalized with polymer zwitterions, *ACS Appl. Mater. Interfaces* 8 (2016) 27585–27593.
- [195] H.-Y. Yu, Y. Kang, Y. Liu, B. Mi, Grafting polyzwitterions onto polyamide by click chemistry and nucleophilic substitution on nitrogen: a novel approach to enhance membrane fouling resistance, *J. Membr. Sci.* 449 (2014) 50–57.
- [196] X. Zhang, J. Tian, S. Gao, W. Shi, Z. Zhang, F. Cui, S. Zhang, S. Guo, X. Yang, H. Xie, Surface functionalization of TFC FO membranes with zwitterionic polymers: improvement of antifouling and salt-responsive cleaning properties, *J. Membr. Sci.* 544 (2017) 368–377.
- [197] J. Wang, T. Xiao, R. Bao, T. Li, Y. Wang, D. Li, X. Li, T. He, Zwitterionic surface modification of forward osmosis membranes using N-aminoethyl piperazine propane sulfonate for grey water treatment, *Process Saf. Environ. Prot.* 116 (2018) 632–639.
- [198] C. Liu, J. Lee, J. Ma, M. Elimelech, Antifouling thin-film composite membranes by controlled architecture of zwitterionic polymer brush layer, *Environ. Sci. Technol.* 51 (2017) 2161–2169.
- [199] S. Jiang, Z. Cao, Ultralow-fouling, functionalizable, and hydrolyzable zwitterionic materials and their derivatives for biological applications, *Adv. Mater.* 22 (2010) 920–932.
- [200] A.R. Statz, R.J. Meagher, A.E. Barron, P.B. Messersmith, New peptidomimetic polymers for antifouling surfaces, *J. Am. Chem. Soc.* 127 (2005) 7972–7973.
- [201] X. Lu, S. Romero-Vargas Castrillón, D.L. Shaffer, J. Ma, M. Elimelech, In situ surface chemical modification of thin-film composite forward osmosis membranes for enhanced organic fouling resistance, *Environ. Sci. Technol.* 47 (2013) 12219–12228.
- [202] S.R.-V. Castrillón, X. Lu, D.L. Shaffer, M. Elimelech, Amine enrichment and poly(ethylene glycol)(PEG) surface modification of thin-film composite forward osmosis membranes for organic fouling control, *J. Membr. Sci.* 450 (2014) 331–339.
- [203] Z. Liu, X. An, C. Dong, S. Zheng, B. Mi, Y. Hu, Modification of thin film composite polyamide membranes with 3D hyperbranched polyglycerol for simultaneous improvement in their filtration performance and antifouling properties, *J. Mater. Chem.* 5 (2017) 23190–23197.
- [204] L. Shen, X. Zhang, J. Zuo, Y. Wang, Performance enhancement of TFC FO membranes with polyethyleneimine modification and post-treatment, *J. Membr. Sci.* 534 (2017) 46–58.
- [205] X. Bao, Q. Wu, W. Shi, W. Wang, Z. Zhu, Z. Zhang, R. Zhang, X. Zhang, B. Zhang, Y. Guo, Insights into simultaneous ammonia-selective and anti-fouling mechanism over forward osmosis membrane for resource recovery from domestic wastewater, *J. Membr. Sci.* 573 (2019) 135–144.
- [206] W.-Z. Qiu, H.-C. Yang, Z.-K. Xu, Dopamine-assisted co-deposition: an emerging and promising strategy for surface modification, *Adv. Colloid Interface Sci.* 256 (2018) 111–125.
- [207] R.-X. Zhang, L. Braeken, P. Luis, X.-L. Wang, B.V.d. Bruggen, Novel binding procedure of TiO₂ nanoparticles to thin film composite membranes via self-polymerized polydopamine, *J. Membr. Sci.* 437 (2013) 179–188.
- [208] F. Bernsmann, V. Ball, F. Addiego, A. Ponche, M. Michel, J.J.d.A. Gracio, V. Toniazio, D. Ruch, Dopamine–melanin film deposition depends on the used oxidant and buffer solution, *Langmuir* 27 (2011) 2819–2825.
- [209] S. Hong, Y.S. Na, S. Choi, I.T. Song, W.Y. Kim, H. Lee, Non-covalent self-assembly and covalent polymerization co-contribute to polydopamine formation, *Adv. Funct. Mater.* 22 (2012) 4711–4717.
- [210] Q. Ye, F. Zhou, W. Liu, Bioinspired catecholic chemistry for surface modification, *Chem. Soc. Rev.* 40 (2011) 4244–4258.
- [211] A. Schulze, M. Went, A. Prager, Membrane functionalization with hyperbranched polymers, *Materials* 9 (2016) 706.
- [212] X. Bao, Q. Wu, W. Shi, W. Wang, H. Yu, Z. Zhu, X. Zhang, Z. Zhang, R. Zhang, F. Cui, Polyamidoamine dendrimer grafted forward osmosis membrane with superior ammonia selectivity and robust antifouling capacity for domestic wastewater concentration, *Water Res.* 153 (2019) 1–10.
- [213] S. Azari, L. Zou, Using zwitterionic amino acid L-DOPA to modify the surface of thin film composite polyamide reverse osmosis membranes to increase their fouling resistance, *J. Membr. Sci.* 401 (2012) 68–75.
- [214] S. Kasemset, A. Lee, D.J. Miller, B.D. Freeman, M.M. Sharma, Effect of polydopamine deposition conditions on fouling resistance, physical properties, and permeation properties of reverse osmosis membranes in oil/water separation, *J. Membr. Sci.* 425 (2013) 208–216.
- [215] R.R. Choudhury, J.M. Gohil, S. Mohanty, S.K. Nayak, Antifouling, fouling release and antimicrobial materials for surface modification of reverse osmosis and nanofiltration membranes, *J. Mater. Chem.* 6 (2018) 313–333.
- [216] R. Zhang, Y. Liu, M. He, Y. Su, X. Zhao, M. Elimelech, Z. Jiang, Antifouling membranes for sustainable water purification: strategies and mechanisms, *Chem. Soc. Rev.* 45 (2016) 5888–5924.
- [217] F. Li, Q. Cheng, Q. Tian, B. Yang, Q. Chen, Biofouling behavior and performance of forward osmosis membranes with bioinspired surface modification in osmotic membrane bioreactor, *Bioresour. Technol.* 211 (2016) 751–758.
- [218] Z. Liu, Y. Hu, Sustainable antibiofouling properties of thin film composite forward osmosis membrane with rechargeable silver nanoparticles loading, *ACS Appl. Mater. Interfaces* 8 (2016) 21666–21673.
- [219] K.R. Zodorow, M.E. Tousley, M. Elimelech, Mitigating biofouling on thin-film composite polyamide membranes using a controlled-release platform, *J. Membr. Sci.* 453 (2014) 84–91.
- [220] J. Yin, Y. Yang, Z. Hu, B. Deng, Attachment of silver nanoparticles (AgNPs) onto thin-film composite (TFC) membranes through covalent bonding to reduce membrane biofouling, *J. Membr. Sci.* 441 (2013) 73–82.
- [221] S.H. Kim, S.-Y. Kwak, B.-H. Sohn, T.H. Park, Design of TiO₂ nanoparticle self-assembled aromatic polyamide thin-film-composite (TFC) membrane as an approach to solve biofouling problem, *J. Membr. Sci.* 211 (2003) 157–165.
- [222] J.-H. Li, B.-F. Yan, X.-S. Shao, S.-S. Wang, H.-Y. Tian, Q.-Q. Zhang, Influence of Ag/TiO₂ nanoparticle on the surface hydrophilicity and visible-light response activity of polyvinylidene fluoride membrane, *Appl. Surf. Sci.* 324 (2015) 82–89.
- [223] M. Ben-Sasson, K.R. Zodorow, Q. Gengeng, Y. Kang, E.P. Giannelis, M. Elimelech, Surface functionalization of thin-film composite membranes with copper nanoparticles for antimicrobial surface properties, *Environ. Sci. Technol.* 48 (2013) 384–393.
- [224] Y. Liao, Y. Wang, X. Feng, W. Wang, F. Xu, L. Zhang, Antibacterial surfaces through dopamine functionalization and silver nanoparticle immobilization, *Mater. Chem. Phys.* 121 (2010) 534–540.
- [225] M. Sile-Yuksel, B. Tas, D.Y. Koseoglu-Imer, I. Koyuncu, Effect of silver nanoparticle (AgNP) location in nanocomposite membrane matrix fabricated with different polymer type on antibacterial mechanism, *Desalination* 347 (2014) 120–130.
- [226] C. Liu, A.F.D. Faria, M. Elimelech, Mitigation of biofilm development on thin-film composite membranes functionalized with zwitterionic polymers and silver nanoparticles environmental science and technology, *Environ. Sci. Technol.* 51 (2016) 182–191.

- [227] M. Hu, K. Zhong, Y. Liang, S.H. Ehrman, B. Mi, Effects of particle morphology on the antibiofouling performance of silver embedded polysulfone membranes and rate of silver leaching, *Ind. Eng. Chem. Res.* 56 (2017) 2240–2246.
- [228] A. Panáček, L. Kvitek, R. Prucek, M. Kolář, R. Večeřová, N. Pizúrová, V.K. Sharma, T.J. Nevěčná, R. Zboril, Silver colloid nanoparticles: synthesis, characterization, and their antibacterial activity, *J. Phys. Chem. B* 110 (2006) 16248–16253.
- [229] D.Y. Koseoglu-Imer, B. Kose, M. Altinbas, I. Koyuncu, The production of polysulfone (PS) membrane with silver nanoparticles (AgNP): physical properties, filtration performances, and biofouling resistances of membranes, *J. Membr. Sci.* 428 (2013) 620–628.
- [230] K. Chamakura, R. Perez-Ballester, Z. Luo, S. Bashir, J. Liu, Comparison of bactericidal activities of silver nanoparticles with common chemical disinfectants, *Colloids Surf., B* 84 (2011) 88–96.
- [231] Z. Liu, L. Qi, X. An, C. Liu, Y. Hu, Surface engineering of thin film composite polyamide membranes with silver nanoparticles through layer-by-layer interfacial polymerization for antibacterial properties, *ACS Appl. Mater. Interfaces* 9 (2017) 40987–40997.
- [232] J. Zhu, J. Hou, Y. Zhang, M. Tian, T. He, J. Liu, V. Chen, Polymeric antimicrobial membranes enabled by nanomaterials for water treatment, *J. Membr. Sci.* 550 (2018) 173–197.
- [233] G. Wyszogrodzka, B. Marszałek, B. Gil, P. Dorożyński, Metal-organic frameworks: mechanisms of antibacterial action and potential applications, *Drug Discov. Today* 21 (2016) 1009–1018.
- [234] R. Hausman, T. Gullinkala, I.C. Escobar, Development of copper-charged polypropylene feedspacers for biofouling control, *J. Membr. Sci.* 358 (2010) 114–121.
- [235] A.K. Chatterjee, R. Chakraborty, T. Basu, Mechanism of antibacterial activity of copper nanoparticles, *Nanotechnology* 25 (2014) 135101.
- [236] Z. Liu, Y. Hu, C. Liu, Z. Zhou, Surface-independent one-pot chelation of copper ions onto filtration membranes to provide antibacterial properties, *Chem. Commun.* 52 (2016) 12245–12248.
- [237] A.C. Abreu, R.R. Tavares, A. Borges, F. Mergulhão, M. Simões, Current and emergent strategies for disinfection of hospital environments, *J. Antimicrob. Chemother.* 68 (2013) 2718–2732.
- [238] S. Khoshhal, A. Ghoreyshi, M. Jahanshahi, Study of the temperature and solvent content effects on the structure of Cu–BTC metal organic framework for hydrogen storage, *RSC Adv.* 5 (2015) 24758–24768.
- [239] M. Berchel, T. Le Gall, C. Denis, S. Le Hir, F. Quentel, C. Elléouet, T. Montier, J.-M. Rueff, J.-Y. Salaün, J.-P. Haelter, A silver-based metal-organic framework material as a ‘reservoir’ of bactericidal metal ions, *New J. Chem.* 35 (2011) 1000–1003.
- [240] G. Wyszogrodzka, B. Marszałek, B. Gil, P. Dorożyński, Metal-organic frameworks: mechanisms of antibacterial action and potential applications, *Drug Discov. Today* 21 (2016) 1009–1018.
- [241] S.F. Seyedpour, A. Rahimpour, G. Najafpour, Facile in-situ assembly of silver-based MOFs to surface functionalization of TFC membrane: a novel approach toward long-lasting biofouling mitigation, *J. Membr. Sci.* 573 (2019) 257–269.
- [242] J. Zhu, J. Wang, J. Hou, Y. Zhang, J. Liu, B. Van der Bruggen, Graphene-based antimicrobial polymeric membranes: a review, *J. Mater. Chem.* 5 (2017) 6776–6793.
- [243] B. Liu, K. Zhou, Recent progress on graphene-analogous 2D nanomaterials: properties, modeling and applications, *Prog. Mater. Sci.* 100 (2019) 99–169.
- [244] S.M. Dizaj, A. Mennati, S. Jafari, K. Khezri, K. Adibkia, Antimicrobial activity of carbon-based nanoparticles, *Adv. Pharmaceut. Bull.* 5 (2015) 19.
- [245] F. Perreault, M.E. Tousley, M. Elimelech, Thin-film composite polyamide membranes functionalized with biocidal graphene oxide nanosheets, *Environ. Sci. Technol. Lett.* 1 (2014) 71–76.
- [246] S. Liu, T.H. Zeng, M. Hofmann, E. Burcombe, J. Wei, R. Jiang, J. Kong, Y. Chen, Antibacterial activity of graphite, graphite oxide, graphene oxide, and reduced graphene oxide: membrane and oxidative stress, *ACS Nano* 5 (2011) 6971–6980.
- [247] W. Choi, J. Choi, J. Bang, J.-H. Lee, Layer-by-layer assembly of graphene oxide nanosheets on polyamide membranes for durable reverse-osmosis applications, *ACS Appl. Mater. Interfaces* 5 (2013) 12510–12519.
- [248] R.R. Nair, H.A. Wu, P.N. Jayaram, I.V. Grigorieva, A.K. Geim, Unimpeded permeation of water through helium-leak-tight graphene-based membranes, *Science* 335 (2012) 442–444.
- [249] H. Ji, H. Sun, X. Qu, Antibacterial applications of graphene-based nanomaterials: recent achievements and challenges, *Adv. Drug Deliv. Rev.* 105 (2016) 176–189.
- [250] W. Hu, C. Peng, W. Luo, M. Lv, X. Li, D. Li, Q. Huang, C. Fan, Graphene-based antibacterial paper, *ACS Nano* 4 (2010) 4317–4323.
- [251] H.M. Hegab, A. ElMekawy, T.G. Barclay, A. Michelmore, L. Zou, C.P. Saint, M. Ginic-Markovic, Fine-tuning the surface of forward osmosis membranes via grafting graphene oxide: performance patterns and biofouling propensity, *ACS Appl. Mater. Interfaces* 7 (2015) 18004–18016.
- [252] S.S. Madaeni, P.A. Rahim, J. Barzin, Preparation of polysulfone ultrafiltration membranes for milk concentration: effect of additives on morphology and performance, *Iran. Polym. J.* 14 (2005) 421–428.
- [253] E. Rusen, A. Mocanu, L.C. Nistor, A. Dinescu, I. Călinescu, G. Mustățe, S.t. I. Voicu, C. Andronescu, A. Diacon, Design of antimicrobial membrane based on polymer colloids/multiwall carbon nanotubes hybrid material with silver nanoparticles, *ACS Appl. Mater. Interfaces* 6 (2014) 17384–17393.
- [254] W. Yuan, J. Chen, G. Shi, Nanoporous graphene materials, *Mater. Today* 17 (2014) 77–85.
- [255] Y.C. Cheng, T.P. Kaloni, Z.Y. Zhu, U. Schwingschlögl, Oxidation of graphene in ozone under ultraviolet light, *Appl. Phys. Lett.* 101 (2012) 73110.
- [256] M. Qiu, C. He, Novel zwitterion-silver nanocomposite modified thin-film composite forward osmosis membrane with simultaneous improved water flux and biofouling resistance property, *Appl. Surf. Sci.* 455 (2018) 492–501.
- [257] P. Lu, S. Liang, T. Zhou, T. Xue, X. Mei, Q. Wang, Layered double hydroxide nanoparticle modified forward osmosis membranes via polydopamine immobilization with significantly enhanced chlorine and fouling resistance, *Desalination* 421 (2017) 99–109.
- [258] C. Petit, J. Burress, T.J. Bandosz, The synthesis and characterization of copper-based metal-organic framework/graphite oxide composites, *Carbon* 49 (2011) 563–572.
- [259] Z. Bian, J. Xu, S. Zhang, X. Zhu, H. Liu, J. Hu, Interfacial growth of metal organic framework/graphite oxide composites through pickering emulsion and their CO₂ capture performance in the presence of humidity, *Langmuir* 31 (2015) 7410–7417.
- [260] K.-Y.A. Lin, F.-K. Hsu, W.-D. Lee, Magnetic cobalt-graphene nanocomposite derived from self-assembly of MOFs with graphene oxide as an activator for peroxymonosulfate, *J. Mater. Chem.* 3 (2015) 9480–9490.
- [261] Q. Liu, G.-R. Xu, Graphene oxide (GO) as functional material in tailoring polyamide thin film composite (TFC PA) reverse osmosis (RO) membranes, *Desalination* 394 (2016) 162–175.
- [262] A. Soroush, W. Ma, Y. Silvino, M.S. Rahaman, Surface modification of thin film composite forward osmosis membrane by silver-decorated graphene-oxide nanosheets, *Environ. Sci.: Nano* 2 (2015) 395–405.
- [263] A. Soroush, W. Ma, M. Cyr, M.S. Rahaman, B. Asadishad, N. Tufenkji, In situ silver decoration on graphene oxide-treated thin film composite forward osmosis membranes: biocidal properties and regeneration potential, *Environ. Sci. Technol. Lett.* 3 (2016) 13–18.
- [264] A.F. Faria, C. Liu, M. Xie, F. Perreault, L.D. Nghiem, J. Ma, M. Elimelech, Thin-film composite forward osmosis membranes functionalized with graphene oxide-silver nanocomposites for biofouling control, *J. Membr. Sci.* 525 (2017) 146–156.
- [265] L. Qi, Y. Hu, Z. Liu, X. An, E. Bar-Zeev, Improved anti-biofouling performance of thin -film composite forward-osmosis membranes containing passive and active moieties, *Environ. Sci. Technol.* 52 (2018) 9684–9693.
- [266] C. Liu, J. Lee, C. Small, J. Ma, M. Elimelech, Comparison of organic fouling resistance of thin-film composite membranes modified by hydrophilic silica nanoparticles and zwitterionic polymer brushes, *J. Membr. Sci.* 544 (2017) 135–142.
- [267] A. Nguyen, L. Zou, C. Priest, Evaluating the antifouling effects of silver nanoparticles regenerated by TiO₂ on forward osmosis membrane, *J. Membr. Sci.* 454 (2014) 264–271.
- [268] D. Li, H. Wang, Recent developments in reverse osmosis desalination membranes, *J. Mater. Chem.* 20 (2010) 4551–4566.
- [269] A. Nguyen, S. Azari, L. Zou, Coating zwitterionic amino acid L-DOPA to increase fouling resistance of forward osmosis membrane, *Desalination* 312 (2013) 82–87.
- [270] H. Choi, A.C. Sofranko, D.D. Dionysiou, Nanocrystalline TiO₂ photocatalytic membranes with a hierarchical mesoporous multilayer structure: synthesis, characterization, and multifunction, *Adv. Funct. Mater.* 16 (2006) 1067–1074.
- [271] X. Zhao, Y. Su, J. Cao, Y. Li, R. Zhang, Y. Liu, Z. Jiang, Fabrication of antifouling polymer-inorganic hybrid membranes through the synergy of biomimetic mineralization and nonsolvent induced phase separation, *J. Mater. Chem.* 3 (2015) 7287–7295.
- [272] J.S. Louie, I. Pinna, M. Reinhard, Effects of surface coating process conditions on the water permeation and salt rejection properties of composite polyamide reverse osmosis membranes, *J. Membr. Sci.* 367 (2011) 249–255.
- [273] L.-P. Zhu, J.-Z. Yu, Y.-Y. Xu, Z.-Y. Xi, B.-K. Zhu, Surface modification of PVDF porous membranes via poly (DOPA) coating and heparin immobilization, *Colloids Surfaces B Biointerfaces* 69 (2009) 152–155.
- [274] N.D. Brault, C. Gao, H. Xue, M. Piliari, J. Homola, S. Jiang, Q. Yu, Ultra-low fouling and functionalizable zwitterionic coatings grafted onto SiO₂ via a biomimetic adhesive group for sensing and detection in complex media, *Biosens. Bioelectron.* 25 (2010) 2276–2282.
- [275] H.-C. Yang, J. Luo, Y. Lv, P. Shen, Z.-K. Xu, Surface engineering of polymer membranes via mussel-inspired chemistry, *J. Membr. Sci.* 483 (2015) 42–59.
- [276] Y. Gao, J. Wu, Z. Zhang, R. Jin, X. Zhang, et al., Synthesis of polypropylene/Mg 3 Al-X (X = CO 3 2–, NO 3 –, Cl–, SO 4 2–) LDH nanocomposites using a solvent mixing method: thermal and melt rheological, *J. Mater. Chem. A* 1 (2013) 9928–9934.
- [277] S. Liang, Y. Kang, A. Tiraferri, E.P. Giannelis, X. Huang, M. Elimelech, Highly hydrophilic polyvinylidene fluoride (PVDF) ultrafiltration membranes via postfabrication grafting of surface-tailored silica nanoparticles, *ACS Appl. Mater. Interfaces* 5 (2013) 6694–6703.
- [278] X.-M. Wang, X.-Y. Li, K. Shih, In situ embedment and growth of anhydrous and hydrated aluminum oxide particles on polyvinylidene fluoride (PVDF) membranes, *J. Membr. Sci.* 368 (2011) 134–143.
- [279] H.R. Ahn, T.M. Tak, Y.-N. Kwon, Preparation and applications of poly vinyl alcohol (PVA) modified cellulose acetate (CA) membranes for forward osmosis (FO) processes, *J. Desalin. Water Treat.* 53 (2015) 1–7.
- [280] S. Zhang, G. Qiu, Y.P. Ting, T.-S. Chung, Silver-PEGylated dendrimer nanocomposite coating for anti-fouling thin film composite membranes for water treatment, *Colloid. Surf. Physicochem. Eng. Asp.* 436 (2013) 207–214.
- [281] B.D. McCloskey, H.B. Park, H. Ju, B.W. Rowe, D.J. Miller, B.J. Chun, K. Kin, B. D. Freeman, Influence of polydopamine deposition conditions on pure water flux and foulant adhesion resistance of reverse osmosis, ultrafiltration, and microfiltration membranes, *J. Polym. Sci.* 51 (2010) 3472–3485.

- [282] C.X. Liu, D.R. Zhang, Y. He, X.S. Zhao, R. Bai, Modification of membrane surface for anti-biofouling performance: effect of anti-adhesion and anti-bacteria approaches, *J. Membr. Sci.* 346 (2010) 121–130.
- [283] G. Zhang, H. Yan, S. Ji, Z. Liu, Self-assembly of polyelectrolyte multilayer pervaporation membranes by a dynamic layer-by-layer technique on a hydrolyzed polyacrylonitrile ultrafiltration membrane, *J. Membr. Sci.* 292 (2007) 1–8.
- [284] W. Jin, A. Toutianoush, B. Tieke, Use of polyelectrolyte layer-by-layer assemblies as nanofiltration and reverse osmosis membranes, *Langmuir* 19 (2003) 2550–2553.
- [285] Q. Saren, C.Q. Qiu, C.Y. Tang, Synthesis and characterization of novel forward osmosis membranes based on layer-by-layer assembly, *Environ. Sci. Technol.* 45 (2011) 5201–5208.
- [286] M. Müller, T. Rieser, K. Lunkwitz, J. Meier-Haack, Polyelectrolyte complex layers: a promising concept for anti-fouling coatings verified by in-situ ATR-FTIR spectroscopy, *Macromol. Rapid Commun.* 20 (1999) 607–611.
- [287] C. Ba, D.A. Ladner, J. Economy, Using polyelectrolyte coatings to improve fouling resistance of a positively charged nanofiltration membrane, *J. Membr. Sci.* 347 (2010) 250–259.
- [288] S. Xiong, S. Xu, A. Phommachanh, M. Yi, Y. Wang, A versatile surface modification of TFC membrane by layer-by-layer assembly of phytic acid-metal complexes for comprehensively enhanced FO performance, *Environ. Sci. Technol.* 53 (2019) 3331–3341.
- [289] X. Liu, S. Qi, Y. Li, L. Yang, B. Cao, C.Y. Tang, Synthesis and characterization of novel antibacterial silver nanocomposite nanofiltration and forward osmosis membranes based on layer-by-layer assembly, *Water Res.* 47 (2013) 3081–3092.
- [290] S. Mazumder, J.O. Falkinham III, A.M. Dietrich, I.K. Puri, Role of hydrophobicity in bacterial adherence to carbon nanostructures and biofilm formation, *Biofouling* 26 (2010) 333–339.
- [291] X. Jin, M. Li, J. Wang, C. Marambio-Jones, F. Peng, X. Huang, R. Damoiseaux, E. M.V. Hoek, High-throughput screening of silver nanoparticle stability and bacterial inactivation in aquatic media: influence of specific ions, *Environ. Sci. Technol.* 44 (2010) 7321–7328.
- [292] M. Li, S. Pokhrel, X. Jin, L. Mädler, R. Damoiseaux, E.M.V. Hoek, Stability, bioavailability, and bacterial toxicity of ZnO and iron-doped ZnO nanoparticles in aquatic media, *Environ. Sci. Technol.* 45 (2010) 755–761.
- [293] H. Salehi, M. Rastgar, A. Shakeri, Anti-fouling and high water permeable forward osmosis membrane fabricated via layer by layer assembly of chitosan/graphene oxide, *Appl. Surf. Sci.* 413 (2017) 99–108.
- [294] Y. Mansourpanah, S. Madaeni, A. Rahimpour, Z. Kheirollahi, M. Adeli, Changing the performance and morphology of polyethersulfone/polyimide blend nanofiltration membranes using trimethylamine, *Desalination* 256 (2010) 101–107.
- [295] A. Mehrparvar, A. Rahimpour, M. Jahanshahi, Modified ultrafiltration membranes for humic acid removal, *J. Taiwan Inst. Chem. Eng.* 45 (2014) 275–282.
- [296] H.M. Hegab, L. Zou, Graphene oxide-assisted membranes: fabrication and potential applications in desalination and water purification, *J. Membr. Sci.* 484 (2015) 95–106.
- [297] V.C. Sanchez, A. Jachak, R.H. Hurt, A.B. Kane, Biological interactions of graphene-family nanomaterials: an interdisciplinary review, *Chem. Res. Toxicol.* 25 (2011) 15–34.
- [298] O. Akhavan, E. Ghaderi, K. Rahimi, Adverse effects of graphene incorporated in TiO₂ photocatalyst on minuscule animals under solar light irradiation, *J. Mater. Chem.* 22 (2012) 23260–23266.
- [299] H.M. Hegab, A. ElMekawy, T.G. Barclay, A. Micheltore, L. Zou, C.P. Saint, M. Ginic-Markovic, Effective in-situ chemical surface modification of forward osmosis membranes with polydopamine-induced graphene oxide for biofouling mitigation, *Desalination* 385 (2016) 126–137.
- [300] C. Liu, L. Shi, R. Wang, Crosslinked layer-by-layer polyelectrolyte nanofiltration hollow fiber membrane for low-pressure water softening with the presence of SO₄²⁻ in feed water, *J. Membr. Sci.* 486 (2015) 169–176.
- [301] S. Liu, D. Volkmer, D.G. Kurth, Functional polyoxometalate thin films via electrostatic layer-by-layer self-assembly, *J. Clust. Sci.* 14 (2003) 405–419.
- [302] H.M. Hegab, A. ElMekawy, A.M. Thomas, G. Barclay, L. Zou, M.G. M. Christopher P. Saint, Single-step assembly of multifunctional poly tannic acid-graphene oxide coating to reduce biofouling of forward osmosis membranes, *ACS Appl. Mater. Interfaces* 8 (2016) 17519–17528.
- [303] C. Boo, M. Elimelech, S. Hong, Fouling control in a forward osmosis process integrating seawater desalination and wastewater reclamation, *J. Membr. Sci.* 444 (2013) 148–156.
- [304] G. Wang, J. Yang, J. Park, X. Gou, B. Wang, H. Liu, J.C. Yao, Facile synthesis and characterization of graphene nanosheets, *J. Phys. Chem. C* 112 (2008) 8192–8195.
- [305] W.-P. Xu, L.-C. Zhang, J.-P. Li, Y. Lu, H.-H. Li, Y.-N. Ma, W.-D. Wang, S.-H. Yu, Facile synthesis of silver@ graphene oxide nanocomposites and their enhanced antibacterial properties, *J. Mater. Chem.* 21 (2011) 4593–4597.
- [306] D. Zhao, S. Yu, A review of recent advance in fouling mitigation of NF/RO membranes in water treatment: pretreatment, membrane modification, and chemical cleaning, *Desalin. Water Treat.* 55 (2015) 870–891.
- [307] N. Misdan, W.J. Lau, A.F. Ismail, Seawater Reverse Osmosis (SWRO) desalination by thin-film composite membrane—current development, challenges and future prospects, *Desalination* 287 (2012) 228–237.
- [308] S. Hermans, H. Mariën, C. Van Goethem, I.F. Vankelecom, Recent developments in thin film (nano)composite membranes for solvent resistant nanofiltration, *Curr. Opin. Chem. Eng.* 8 (2015) 45–54.
- [309] S. Hermans, H. Mariën, C. Van Goethem, I.F. Vankelecom, Recent developments in thin film (nano)composite membranes for solvent resistant nanofiltration, *Curr. Opin. Chem. Eng.* 8 (2015) 45–54.
- [310] C. Kong, T. Shintani, T. Tsuru, “Pre-seeding”-assisted synthesis of a high performance polyamide-zeolite nanocomposite membrane for water purification, *New J. Chem.* 34 (2010) 2101–2104.
- [311] M.A. Kuehne, R.Q. Song, N.N. Li, R.J. Petersen, Flux enhancement in TFC RO membranes, *Environ. Prog.* 20 (2001) 23–26.
- [312] N. Savage, M.S. Diallo, Nanomaterials and water purification: opportunities and challenges, *J. Nanoparticle Res.* 7 (2005) 331–342.
- [313] G.M. Geise, D.R. Paul, B.D. Freeman, Fundamental water and salt transport properties of polymeric materials, *Prog. Polym. Sci.* 39 (2014) 1–42.
- [314] J.G. Wijmans, R.W. Baker, The solution-diffusion model: a review, *J. Membr. Sci.* 107 (1995) 1–21.
- [315] G.M. Geise, H.S. Lee, D.J. Miller, B.D. Freeman, J.E. McGrath, D.R. Paul, Water purification by membranes: the role of polymer science, *J. Polym. Sci. B Polym. Phys.* 48 (2010) 1685–1718.
- [316] M.R. Esfahani, J.L. Tyler, H.A. Stretz, M.J. Wells, Effects of a dual nanofiller, nano-TiO₂ and MWCNT, for polysulfone-based nanocomposite membranes for water purification, *Desalination* 372 (2015) 47–56.
- [317] G. Mohammadi, M. Jahanshahi, A. Rahimpour, Fabrication and evaluation of Nafion nanocomposite membrane based on ZrO₂-TiO₂ binary nanoparticles as fuel cell MEA, *Int. J. Hydrogen Energy* 38 (2013) 9387–9394.
- [318] S. Zhao, L. Zou, C.Y. Tang, D. Mulcahy, Recent developments in forward osmosis: opportunities and challenges, *J. Membr. Sci.* 396 (2012) 1–21.
- [319] S.S. Sablani, M.F.A. Goosen, R. Al-Belushi, M. Wilf, Concentration polarization in ultrafiltration and reverse osmosis: a critical review, *Desalination* 141 (2001) 269–289.
- [320] A.G. Fane, P. Beatson, H. Li, Membrane fouling and its control in environmental applications, *Water Sci. Technol.* 41 (2000) 303–308.
- [321] J. Heim, E. Felder, M.N. Tahir, A. Kaltbeitzel, U.R. Heinrich, C. Brochhausen, V. Mailänder, W. Tremel, J. Brieger, Genotoxic effects of zinc oxide nanoparticles, *Nanoscale* 7 (2015) 8931–8938.
- [322] F. Mafuné, J.-y. Kohno, Y. Takeda, T. Kondow, H. Sawabe, Structure and stability of silver nanoparticles in aqueous solution produced by laser ablation, *J. Phys. Chem. B* 104 (2000) 8333–8337.
- [323] L. Shen, S. Xiong, Y. Wang, Graphene oxide incorporated thin-film composite membranes for forward osmosis applications, *Chem. Eng. Sci.* 143 (2016) 194–205.
- [324] S. Xia, L. Yao, Y. Zhao, N. Li, Y. Zheng, Preparation of graphene oxide modified polyamide thin film composite membranes with improved hydrophilicity for natural organic matter removal, *Chem. Eng. J.* 280 (2015) 720–727.
- [325] C. Tang, X. Li, Z. Li, J. Hao, L. Shen, S. Xiong, Y. Wang, S. Xia, L. Yao, Y. Zhao, N. Li, Y. Zheng, M. Hu, B. Mi, V. Vatanpour, S.S. Madaeni, A.R. Khataee, E. Salehi, S. Zinadini, H.A. Monfared, A. Mollahosseini, A. Rahimpour, M. Jahamshahi, M. Peyravi, M. Khavarpour, A. Sotto, A. Boromand, R. Zhang, P. Luis, J. M. Arsuaga, J. Kim, B. Van der Bruggen, A. Razmjou, J. Mansouri, V. Chen, J. Z. Wang, Y. Zhao, C.B. Zhang, M.R. Bao, G.R. Zhu, M. Wang, C.J. Gao, Progress on preparation of mono-dispersed spherical nano-silica, *Mater. Rev.* 25 (2011) 135–139.
- [326] C. Tang, X. Li, Z. Li, J. Hao, Interfacial hydrogen bonds and their influence mechanism on increasing the thermal stability of nano-SiO₂-modified meta-aramid fibres, *Polymers* 9 (2017) 504.
- [327] J.-W. Zha, Y.-H. Zhu, W.-K. Li, J. Bai, Z.-M. Dang, Low dielectric permittivity and high thermal conductivity silicone rubber composites with micro-nano-sized particles, *Appl. Phys. Lett.* 101 (2012) 62905.
- [328] Q. Zhao, J. Hou, J. Shen, J. Liu, Y. Zhang, Long-lasting antibacterial behavior of a novel mixed matrix water purification membrane, *J. Mater. Chem.* 3 (2015) 18696–18705.
- [329] B. Mi, Graphene oxide membranes for ionic and molecular sieving, *Science* 343 (2014) 740–742.
- [330] M.L. Lind, A.K. Ghosh, A. Jawor, X. Huang, W. Hou, Y. Yang, E.M.V. Hoek, Influence of zeolite crystal size on zeolite-polyamide thin film nanocomposite membranes, *Langmuir* 25 (2009) 10139–10145.
- [331] A. Mollahosseini, A. Rahimpour, M. Jahamshahi, M. Peyravi, M. Khavarpour, The effect of silver nanoparticle size on performance and antibacteriality of polysulfone ultrafiltration membrane, *Desalination* 306 (2012) 41–50.
- [332] A. Sotto, A. Boromand, R. Zhang, P. Luis, J.M. Arsuaga, J. Kim, B. Van der Bruggen, Effect of nanoparticle aggregation at low concentrations of TiO₂ on the hydrophilicity, morphology, and fouling resistance of PES-TiO₂ membranes, *J. Colloid Interface Sci.* 363 (2011) 540–550.
- [333] A. Razmjou, J. Mansouri, V. Chen, The effects of mechanical and chemical modification of TiO₂ nanoparticles on the surface chemistry, structure and fouling performance of PES ultrafiltration membranes, *J. Membr. Sci.* 378 (2011) 73–84.
- [334] G.-R. Xu, J.-N. Wang, C.-J. Li, Polyamide nanofilm composite membranes (NCMs) supported by chitosan coated electropositive nanofibrous membranes: preparation and separation performance research, *Desalination* 328 (2013) 31–41.
- [335] M. Son, H.-g. Choi, L. Liu, E. Celik, H. Park, H. Choi, Efficacy of carbon nanotube positioning in the polyethersulfone support layer on the performance of thin-film composite membrane for desalination, *Chem. Eng. J.* 266 (2015) 376–384.
- [336] Y.-H. La, J. Diep, R. Al-Rasheed, M. Nassar, E.I. Mouhoumed, A. Szymczyk, G. Dubois, The effect of cross-contamination in the sequential interfacial polymerization on the RO performance of polyamide bilayer membranes, *J. Membr. Sci.* 466 (2014) 348–356.

- [337] X. Xu, Y. Chen, P. Wan, K. Gasem, K. Wang, T. He, H. Adidharma, M. Fan, Extraction of lithium with functionalized lithium ion-sieves, *Prog. Mater. Sci.* 84 (2016) 276–313.
- [338] E. Zolghadr, V.F. Krapivin, C.A. Varotsos, Soldatov VY| New ecoinformatics tools in environmental science, *Environ. Earth Sci.* 75 (2016) 450.
- [339] A. Shakeri, K.-W. Lee, T. Pöschel, Limitation of stochastic rotation dynamics to represent hydrodynamic interaction between colloidal particles, *Phys. Fluids* 30 (2018), 013603.
- [340] J.R. Lead, K.J. Wilkinson, Environmental colloids and particles: current knowledge and future developments, *IUPAC Ser. Anal. Phys. Chem. Environ. Syst.* 10 (2007) 1.
- [341] B.v. Derjaguin, L. Landau, Theory of the stability of strongly charged lyophobic sols and of the adhesion of strongly charged particles in solutions of electrolytes, *Prog. Surf. Sci.* 43 (1993) 30–59.
- [342] P. Bacchin, P. Almar, R.W. Field, Critical and sustainable fluxes: theory, experiments and applications, *J. Membr. Sci.* 281 (2006) 42–69.
- [343] G. Amouzandeh, F. Mentink-Vigier, S. Helsper, F.A. Bagdasarian, J.T. Rosenberg, S. Grant, Magnetic resonance electrical property mapping at 21.1 T: a study of conductivity and permittivity in phantoms, ex vivo tissue and in vivo ischemia, *Phys. Med. Biol.* (2019).
- [344] E.J.W. Verwey, Theory of the stability of lyophobic colloids, *J. Phys. Chem.* 51 (1947) 631–636.
- [345] C. Vajner, H. Yan, L. Guo, M. Mathews, M. Kuhlman, S. Benefield, S. Ulrich, E. Zolghadr, P. Kung, L. Li, Thickness identification of epitaxial Bi₂Te₃ via optical contrast, *2D Mater.* 3 (2016), 021010.
- [346] J.S. Taurozzi, H. Arul, V.Z. Bosak, A.F. Burban, T.C. Voice, M.L. Bruening, V. Tarabara, Effect of filler incorporation route on the properties of polysulfone–silver nanocomposite membranes of different porosities, *J. Membr. Sci.* 325 (2008) 58–68.
- [347] J. Yin, B. Deng, Polymer-matrix nanocomposite membranes for water treatment, *J. Membr. Sci.* 479 (2015) 256–275.
- [348] A. Zirehpour, A. Rahimpour, A. Arabi Shamsabadi, M. Sharifian Gh, M. Soroush, Mitigation of thin-film composite membrane biofouling via immobilizing nano-sized biocidal reservoirs in the membrane active layer, *Environ. Sci. Technol.* 51 (2017) 5511–5522.
- [349] H.-L. Yang, J. Chun-Te Lin, C. Huang, Application of nanosilver surface modification to RO membrane and spacer for mitigating biofouling in seawater desalination, *Water Res.* 43 (2009) 3777–3786.
- [350] L. He, L.F. Dumée, C. Feng, L. Velleman, R. Reis, F. She, W. Gao, L. Kong, Promoted water transport across graphene oxide–poly (amide) thin film composite membranes and their antibacterial activity, *Desalination* 365 (2015) 126–135.
- [351] X. Wang, X. Wang, P. Xiao, J. Li, E. Tian, Y. Zhao, Y. Ren, High water permeable free-standing cellulose triacetate/graphene oxide membrane with enhanced antibiofouling and mechanical properties for forward osmosis, *Colloid. Surf. Physicochem. Eng. Asp.* 508 (2016) 327–335.
- [352] S. Aghapour Aktij, A. Rahimpour, A. Figoli, Low content nano-polyrhodanine modified polysulfone membranes with superior properties and their performance for wastewater treatment, *Environ. Sci.: Nano* 4 (2017) 2043–2054.
- [353] S. Aghapour Aktij, A. Zirepour, A. Mollahosseini, M.J. Taherzadeh, A. Tiraferri, A. Rahimpour, Feasibility of membrane processes for the recovery and purification of bio-based volatile fatty acids: a comprehensive review, *J. Ind. Eng. Chem.* (2019).